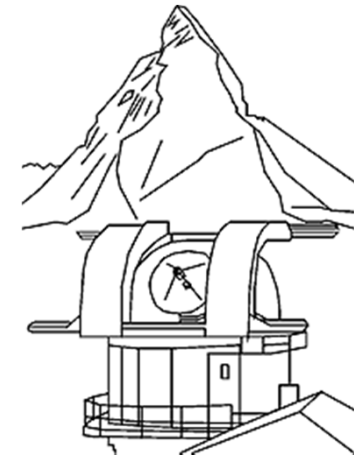


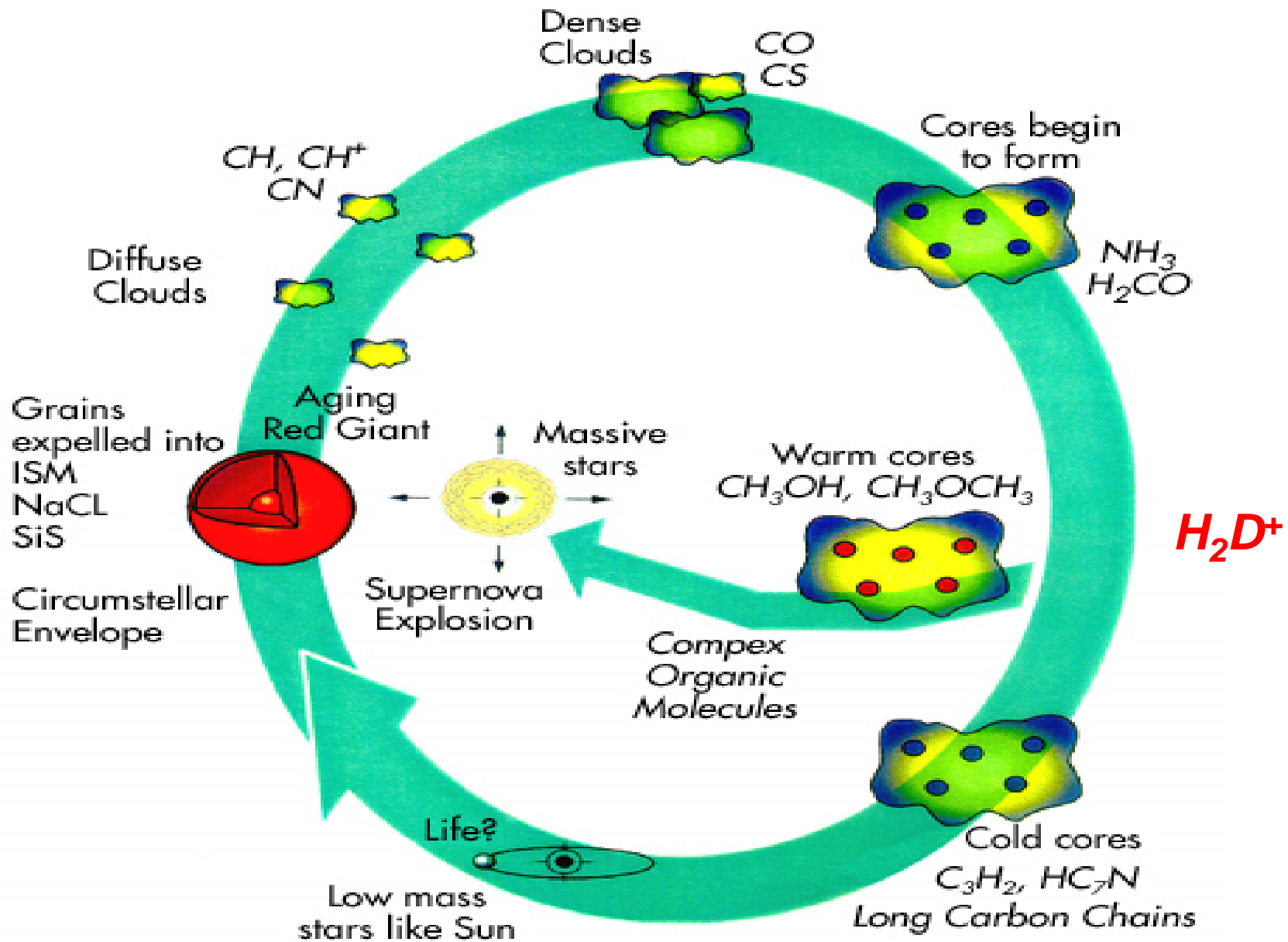
Cold Chemistry in Space and Laboratory

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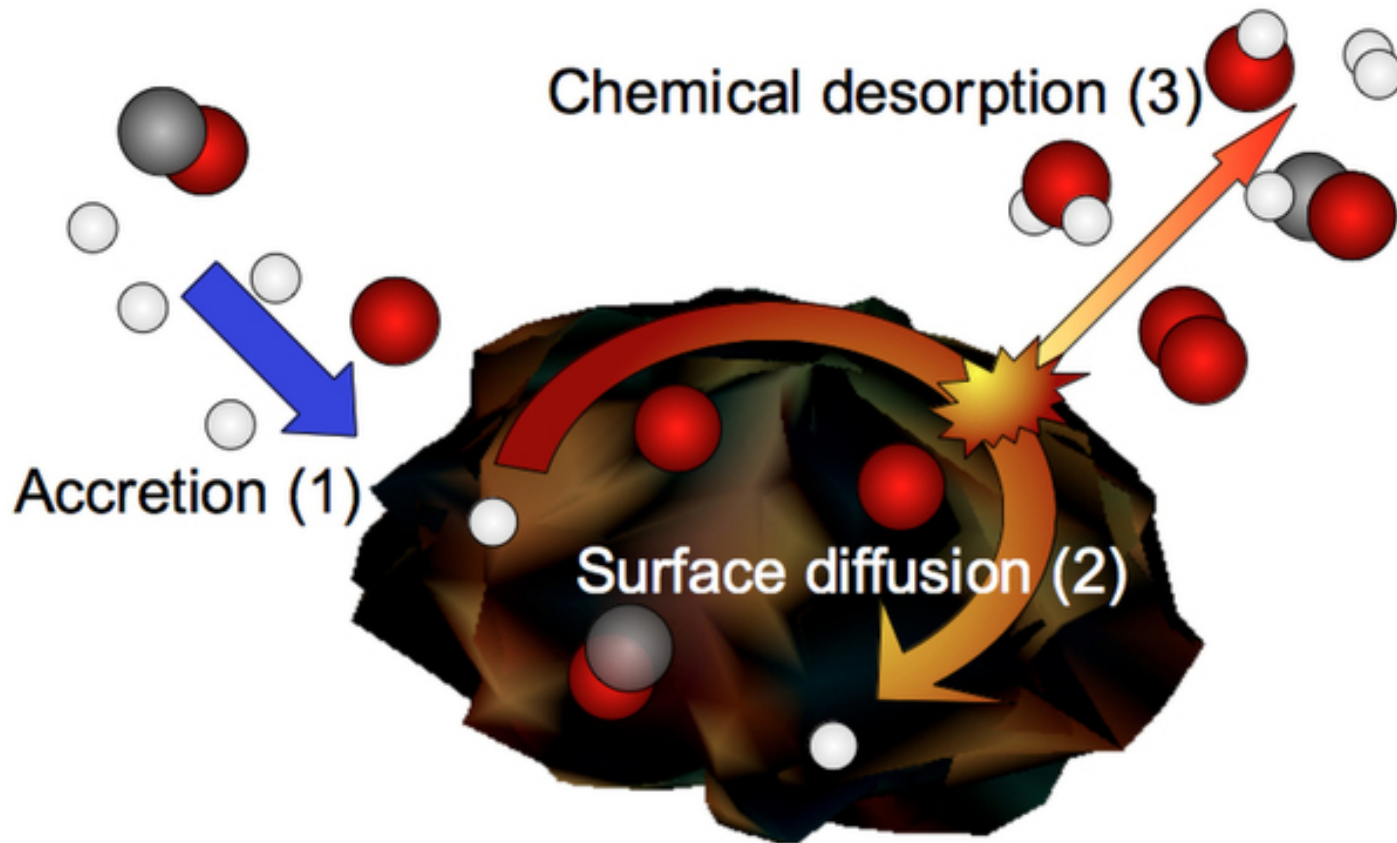


- H_2 Formation, OPR and Chemical Clocks
- $\text{H}_3^+ / \text{H}_2\text{D}^+$ Isotopic Fractionation, $\text{H}_3^+ / \text{H}_2\text{D}^+$, OPR
- $\text{H}_2\text{D}^+ + \text{H}_2$ THz Spectroscopy in Lab and Space

Life cycle of Stars



Hydrogen Formation on Grain Surfaces

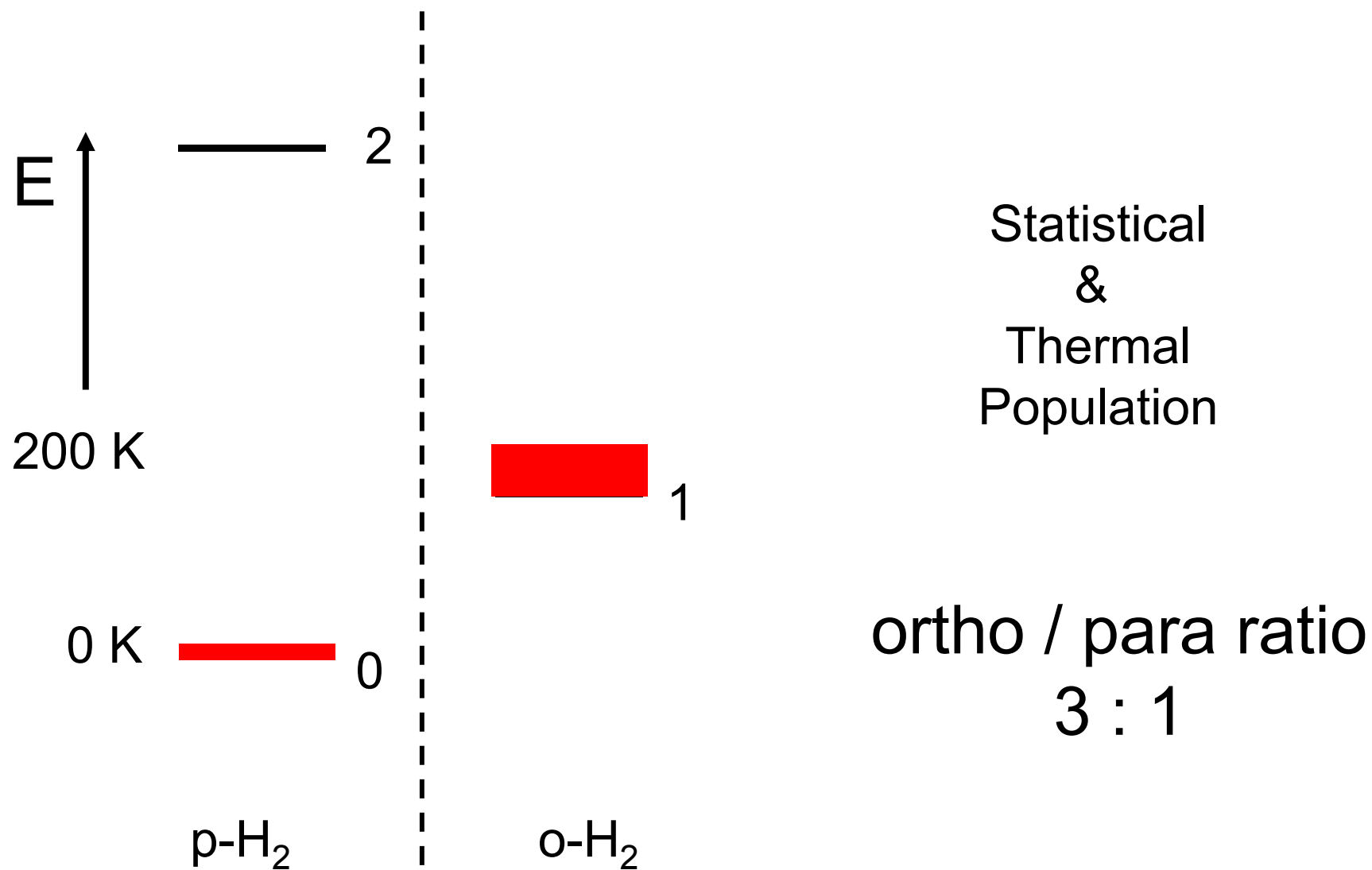


Symmetry considerations and Pauli Principle

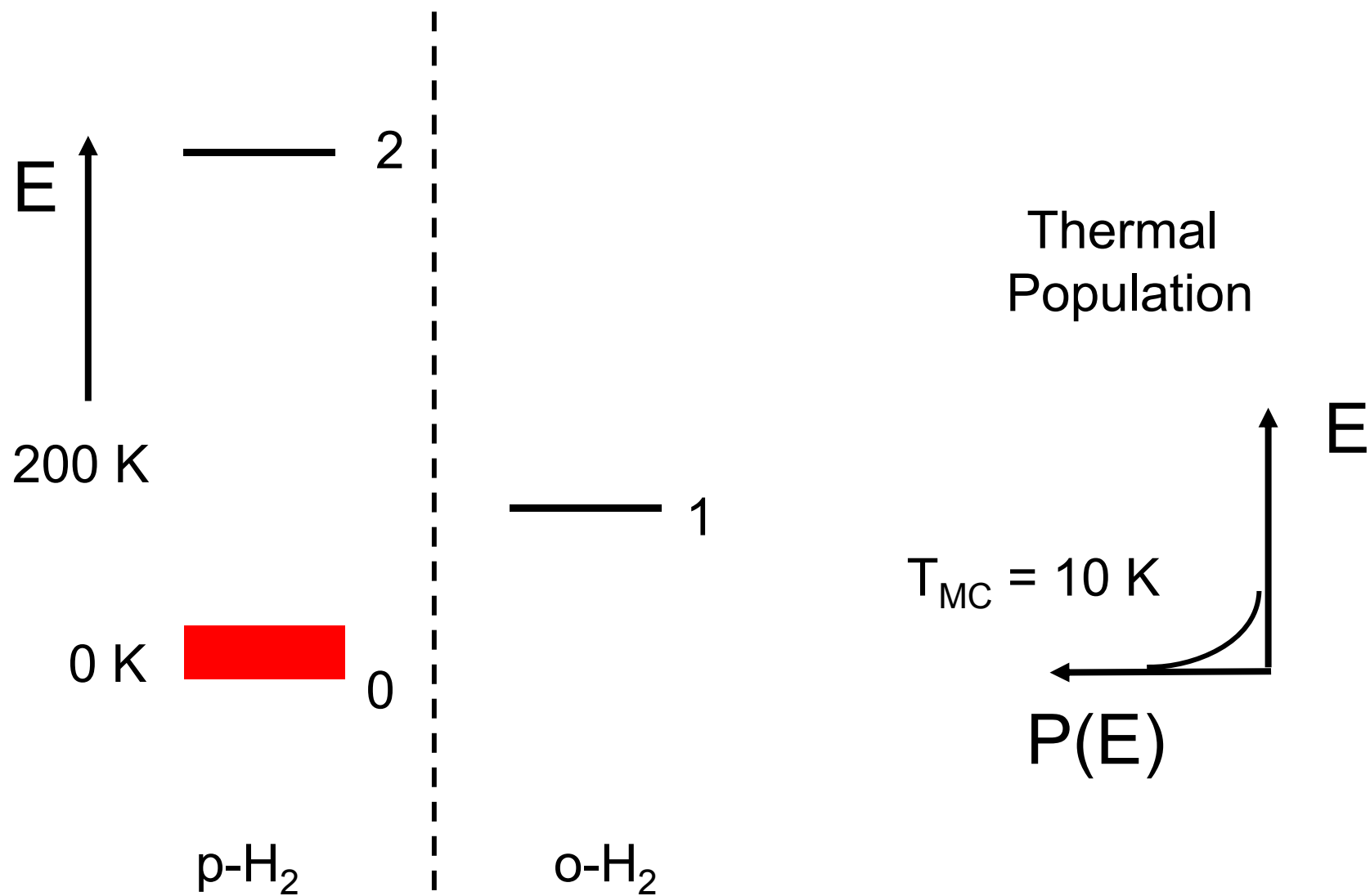
$$\Psi_{\text{tot}} = \Psi_{\text{el}} \cdot \Psi_{\text{vib}} \cdot \Psi_{\text{rot}} \cdot \Psi_{\text{nuc}}$$

para	even $J = 0$	$\uparrow\downarrow - \downarrow\uparrow$	a
ortho	odd $J = 1$	$\uparrow\downarrow + \downarrow\uparrow$ $\uparrow\uparrow$ $\downarrow\downarrow$	s

Lowest Rotational States of H₂

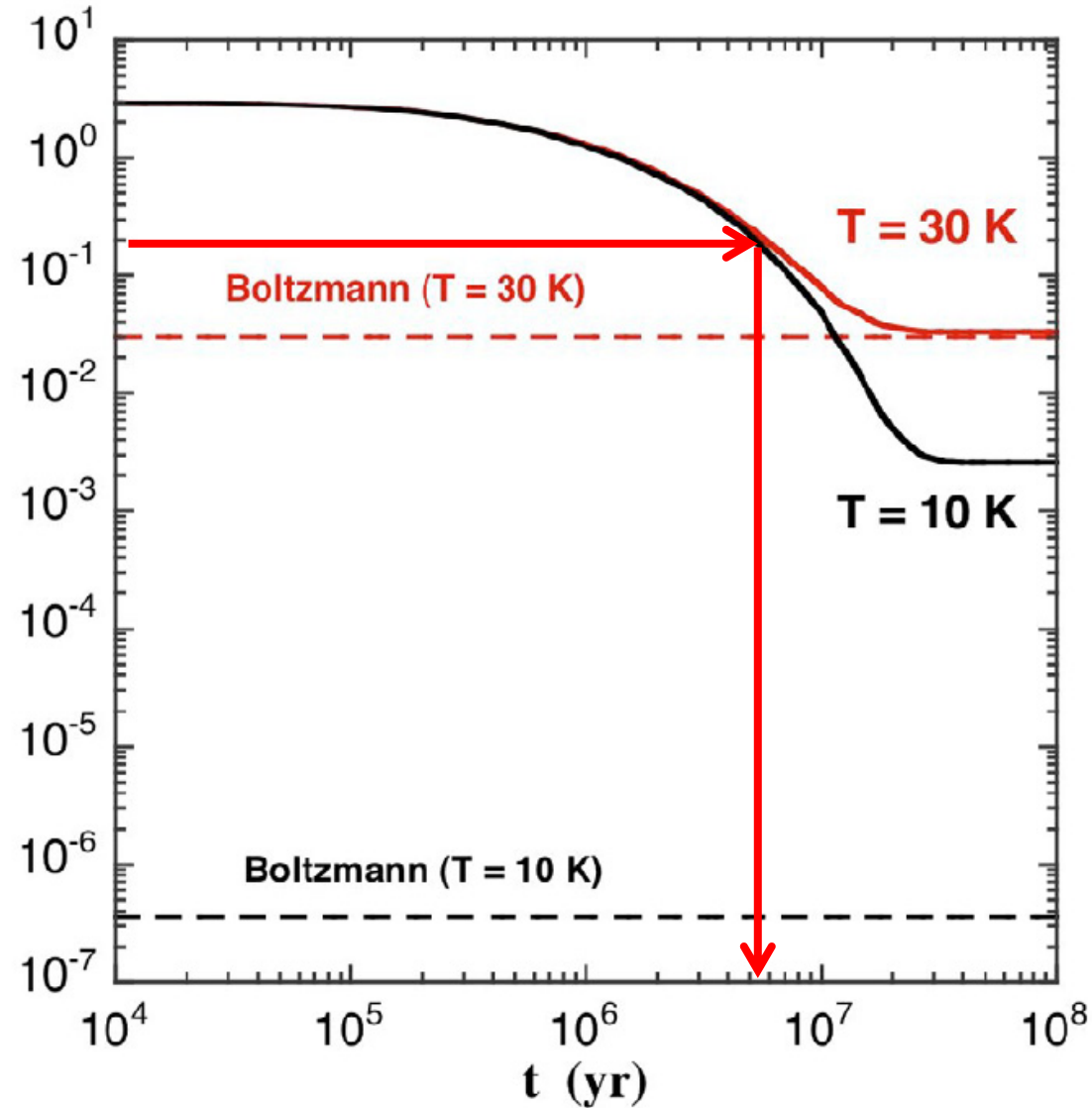
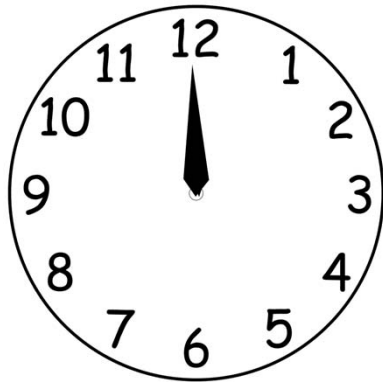


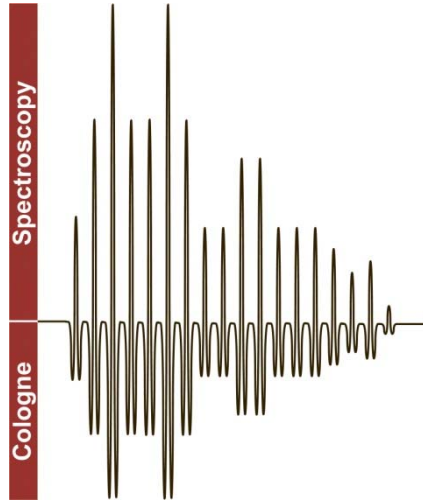
Lowest Rotational States of H₂



Chemical Clock

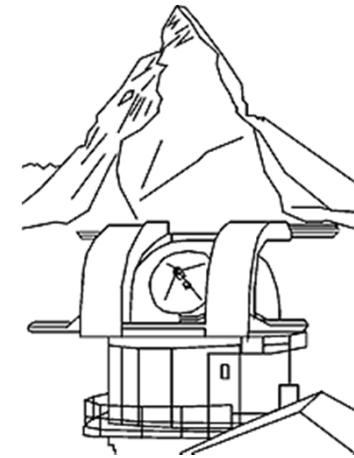
o/p
 H_2





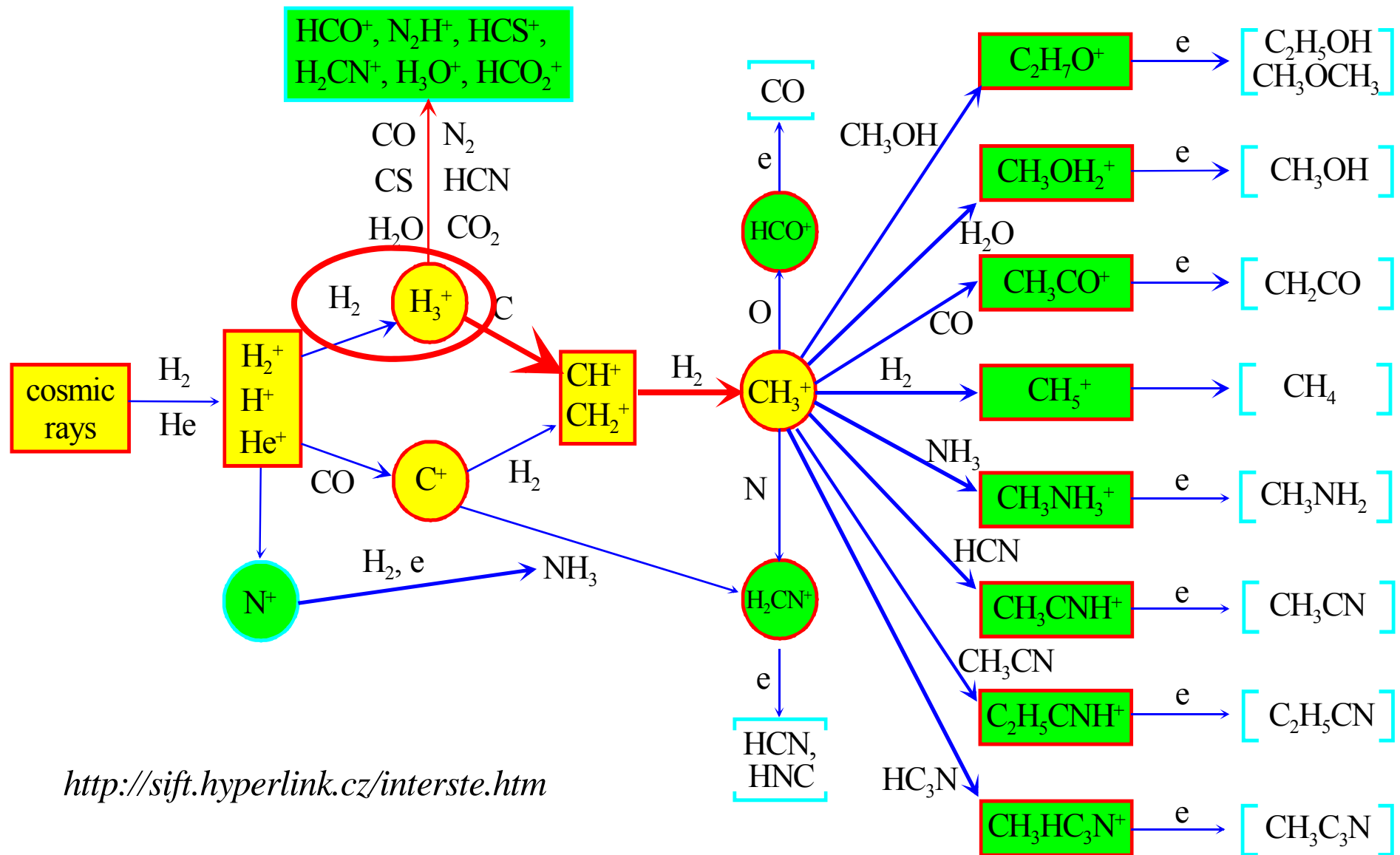
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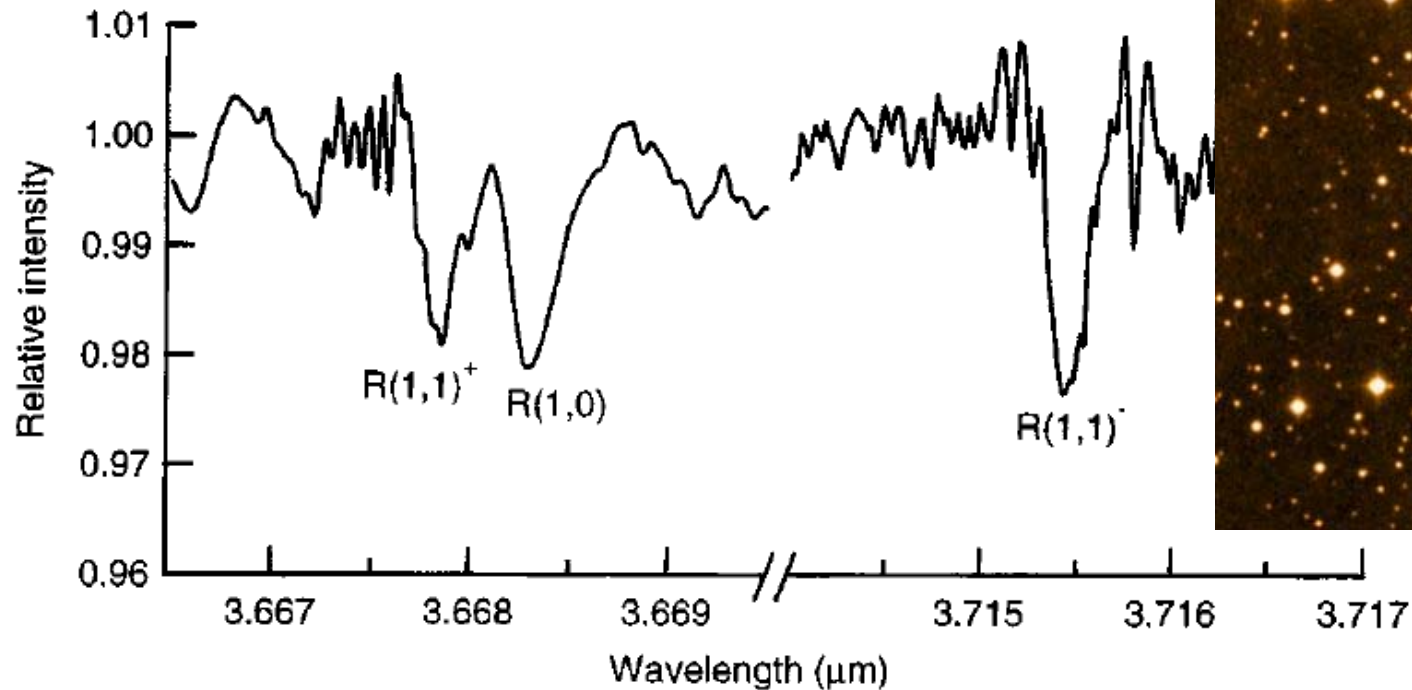
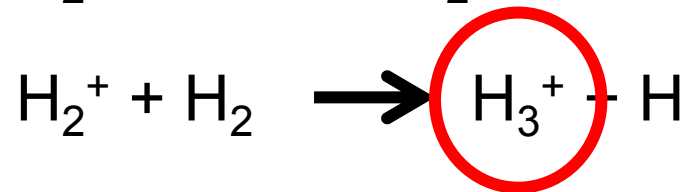
- H₂ Formation, OPR and Chemical Clocks
- H₃⁺ / H₂D⁺ Isotopic Fractionation, H₃⁺/H₂D⁺, OPR
- H₂D⁺ + H₂ THz Spectroscopy in Lab and Space

Initial Reactions in Dense Interstellar Clouds



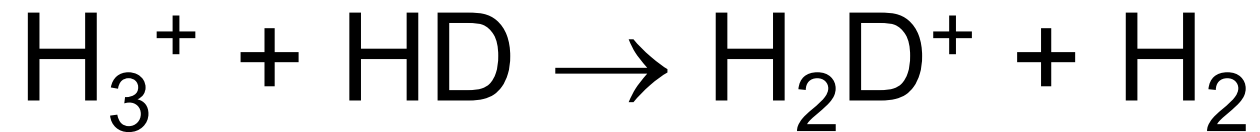
Detection of H_3^+ in the Diffuse Interstellar Medium Toward Cygnus OB2 No. 12

B. J. McCall,* T. R. Geballe, K. H. Hinkle, T. Oka



SCIENCE • VOL. 279 • 20 MARCH 1998

Isotopic Fractionation



E. Hugo, O. Asvany and S. Schlemmer, *J. Chem. Phys.* **130**, Art.-No. 164302 (2009)

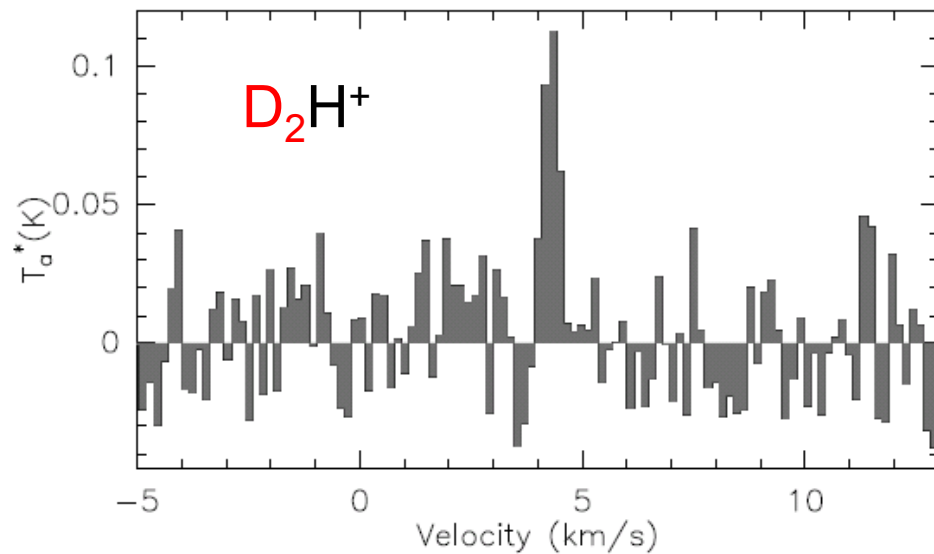
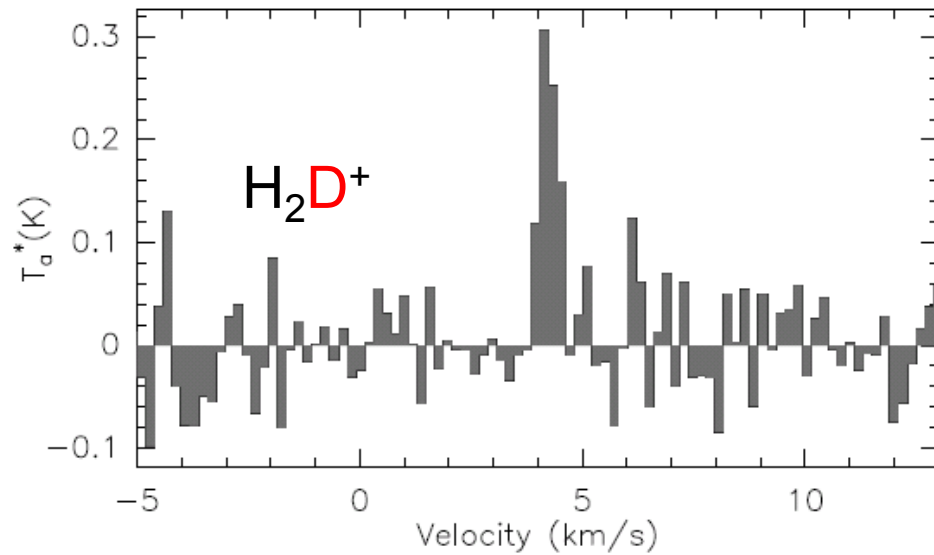
Deuterated Molecules in Interstellar Medium

Cosmic $[D]/[H] \sim 1.5 \cdot 10^{-5}$

Deuteriumreservoir

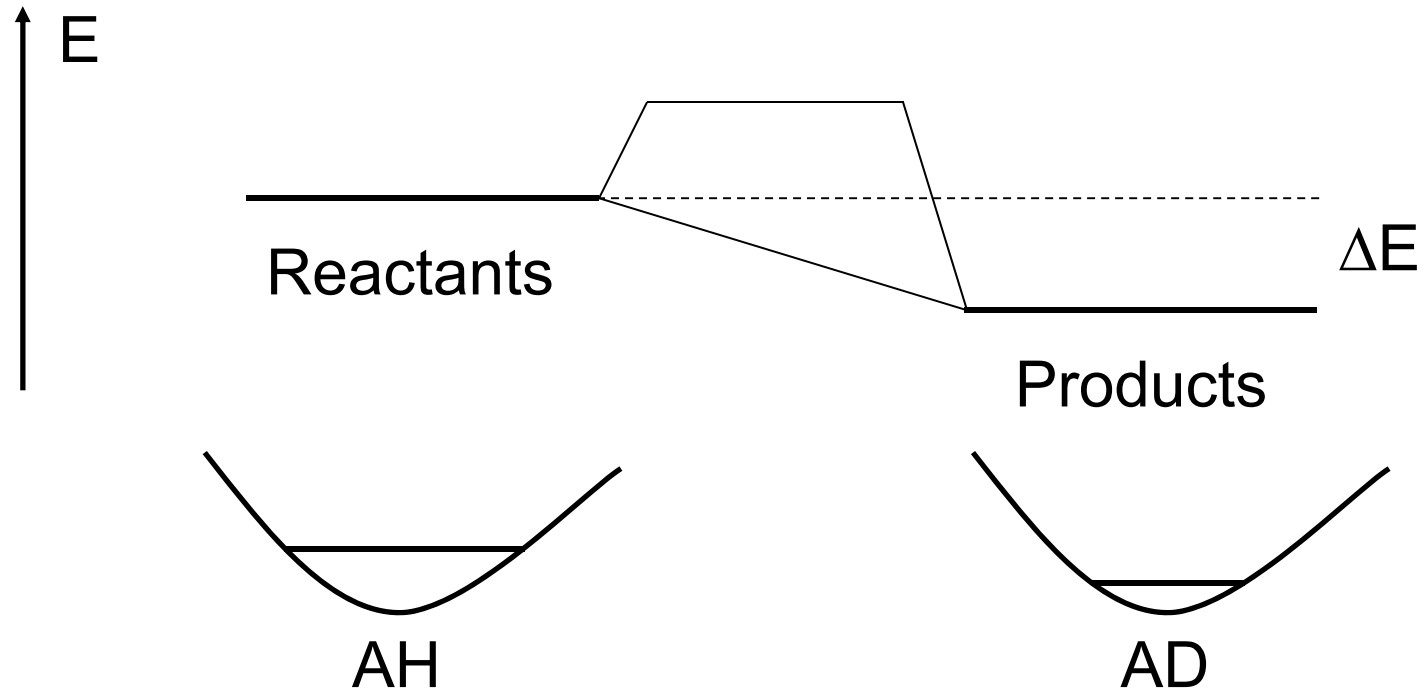
$$[HD]/[H_2] \sim 3.0 \cdot 10^{-5}$$

Isotope Enrichment
 $[AD]/[AH] \sim 0.1$



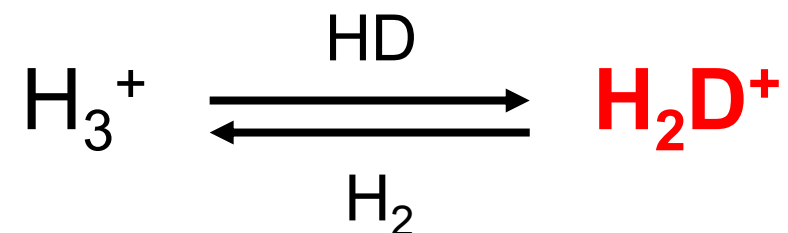
B. Parise, A. Belloche, F. Du, R. Güsten and K. Menten, A&A **526**, A31 (2011)
C. Vastel and T.G: Phillips, APJ, **606**, L127 (2004)

Primary Deuteration Reactions



Isotopic Fractionation

Ideal Case – Laboratory Situation

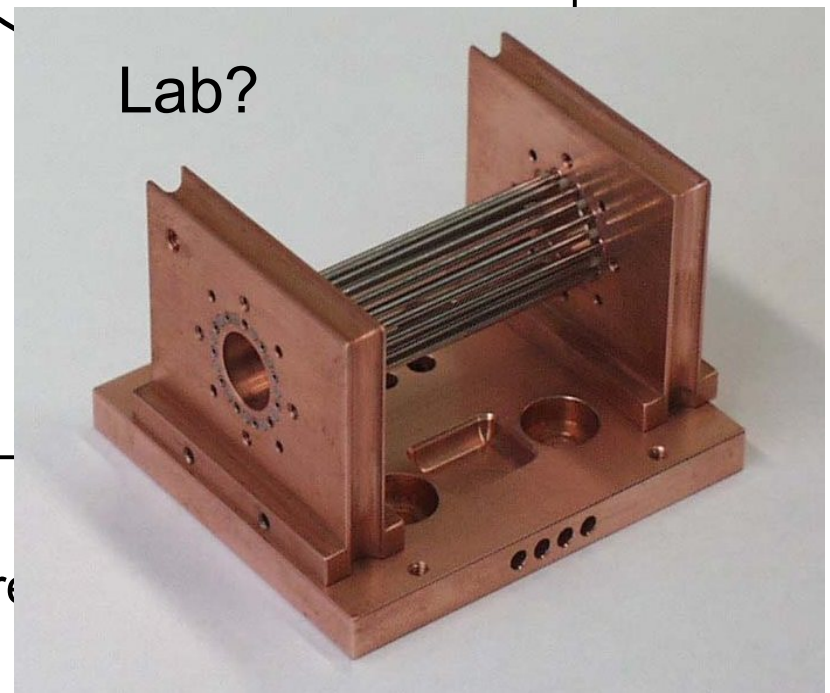
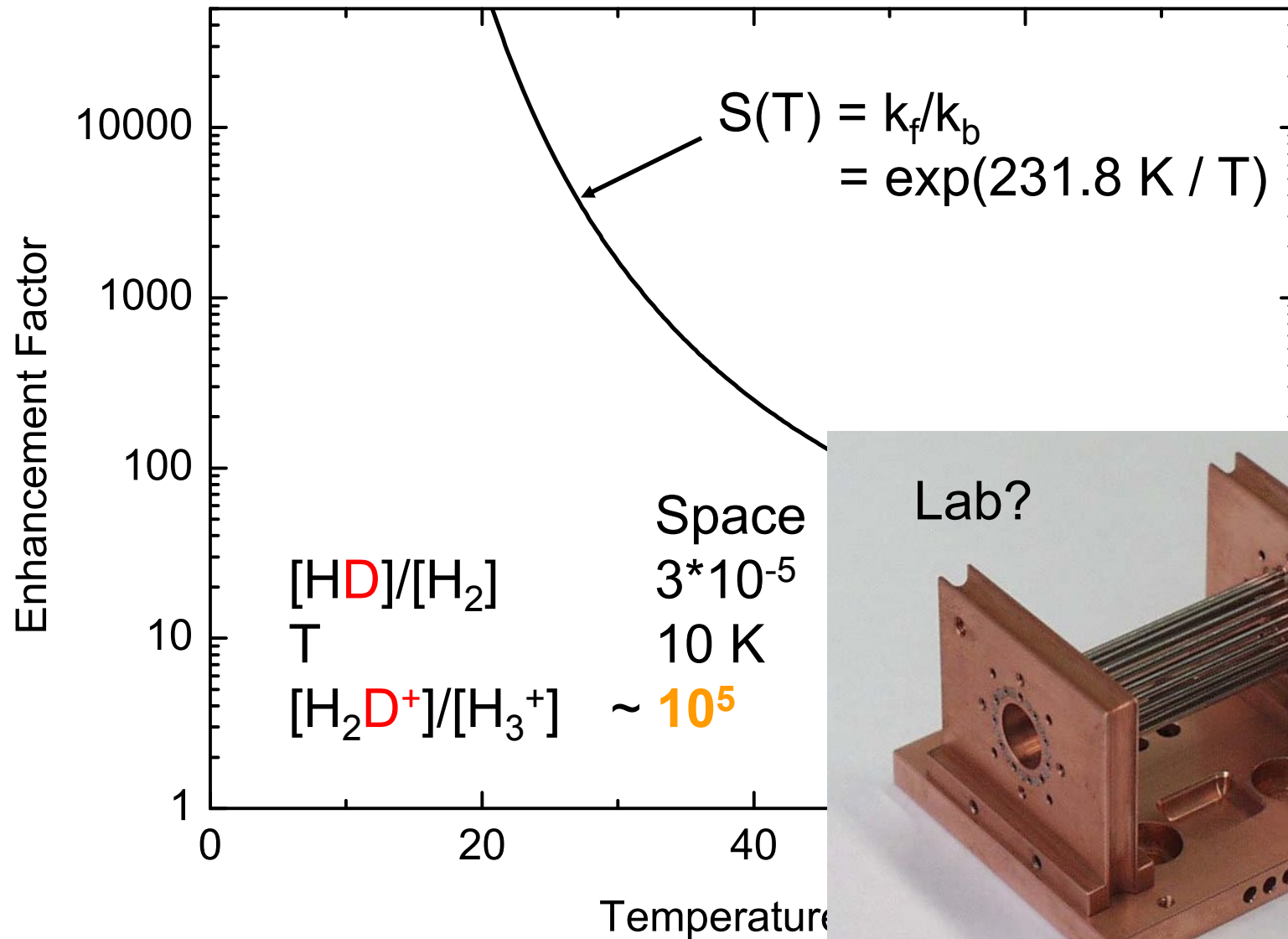


Equilibrium

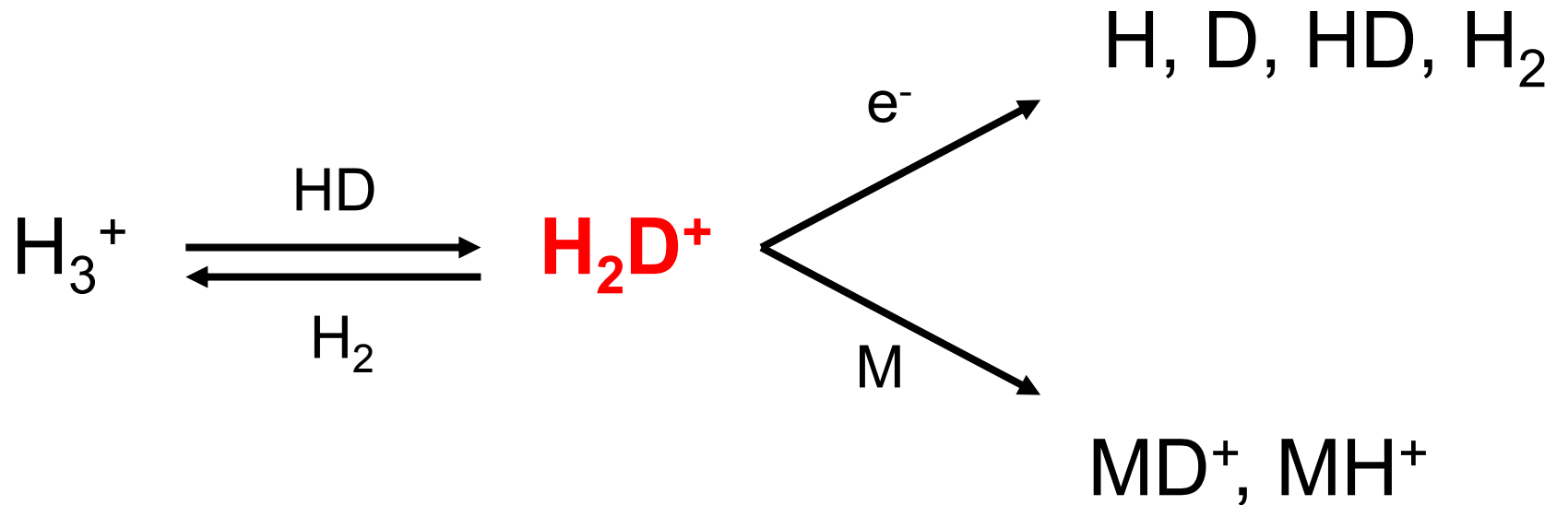
$$[\text{H}_2\text{D}^+]/[\text{H}_3^+] = S(T) [\text{HD}]/[\text{H}_2]$$

$$S(T) = k_f/k_b$$

Isotopic Fractionation



Isotopic Fractionation Cloud Model

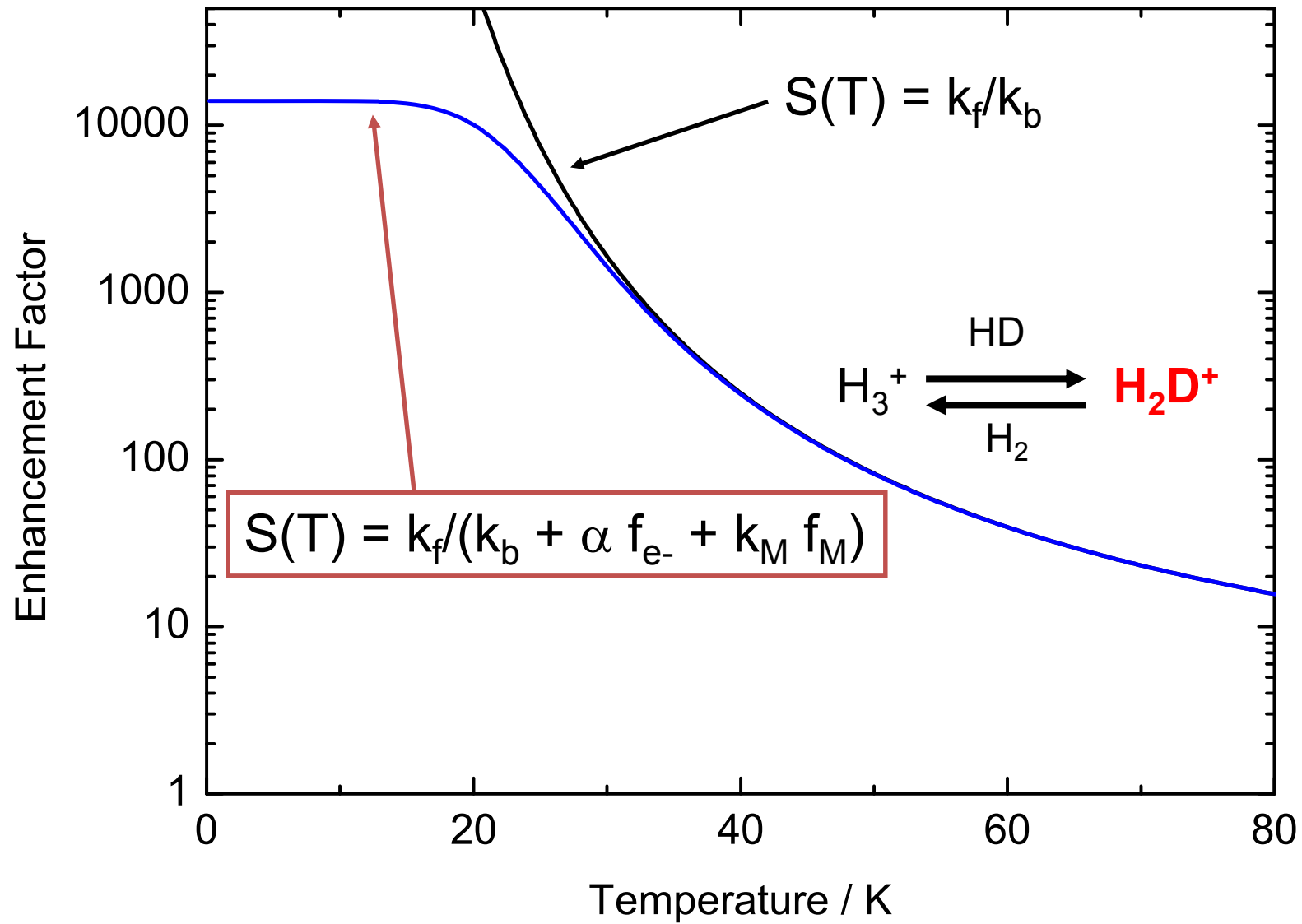


Equilibrium

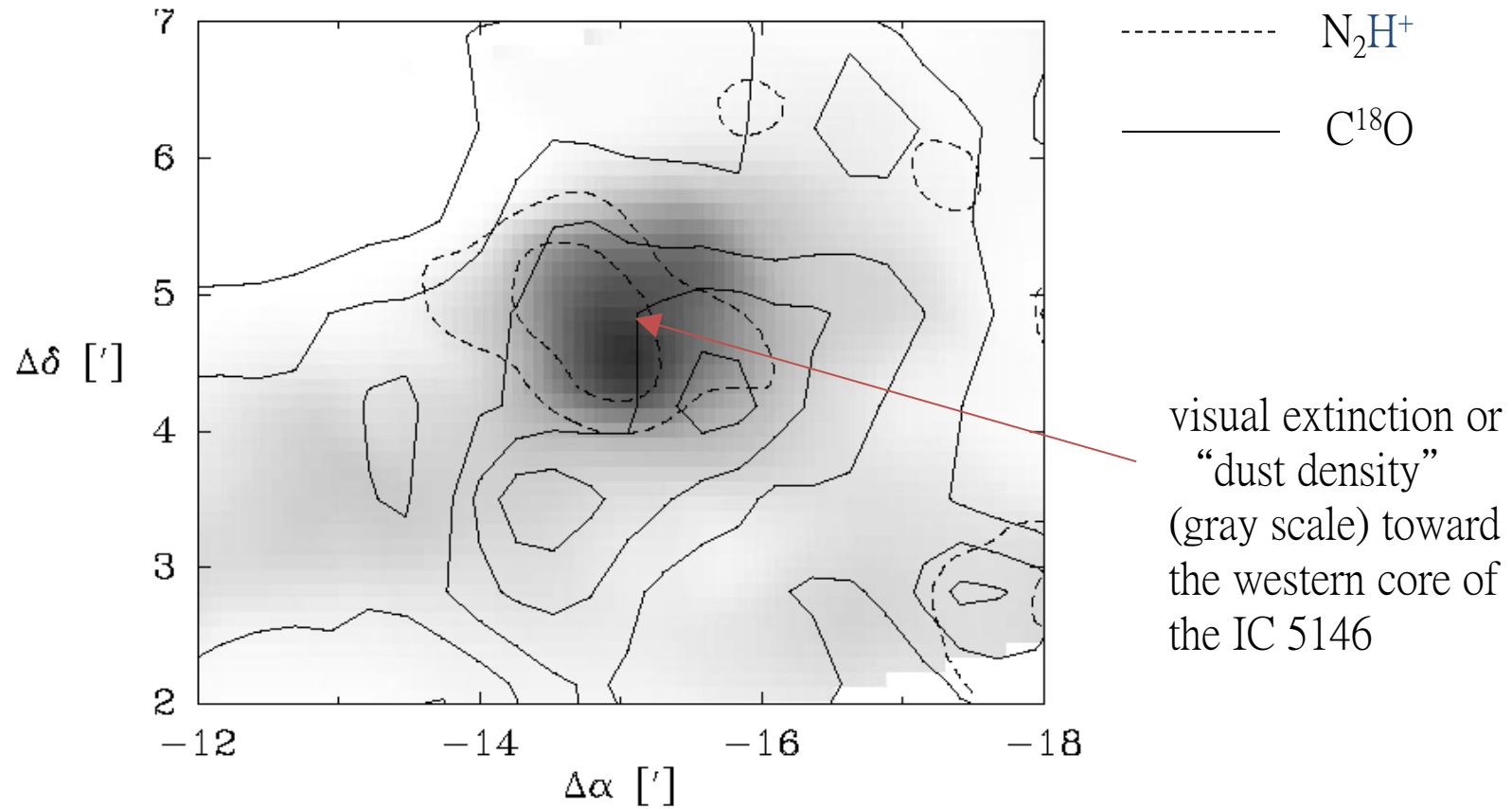
$$[\text{H}_2\text{D}^+]/[\text{H}_3^+] = S(T) [\text{HD}]/[\text{H}_2]$$

$$S(T) = k_f / (k_b + \alpha f_{e^-} + k_M f_M)$$

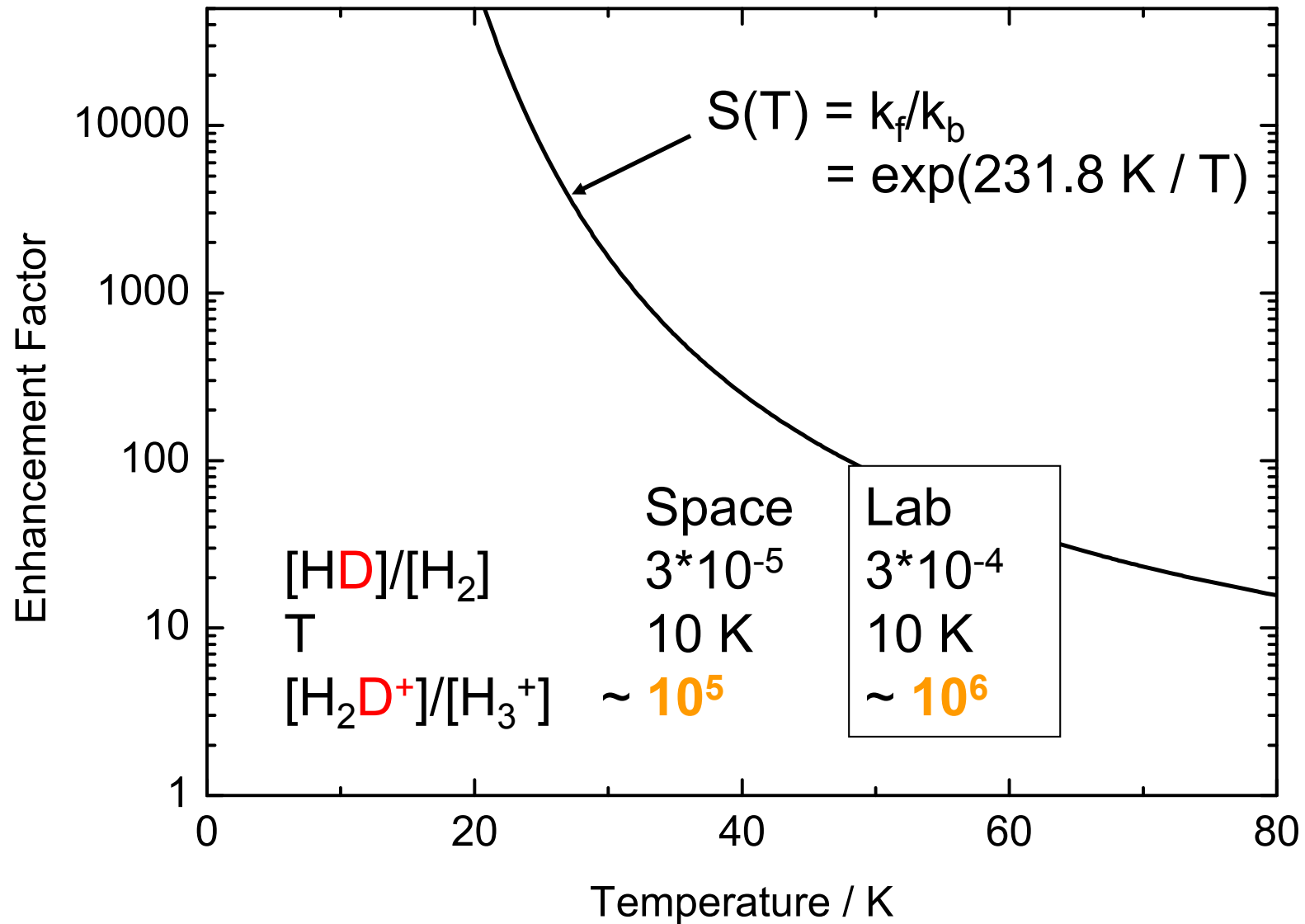
Isotopic Fractionation



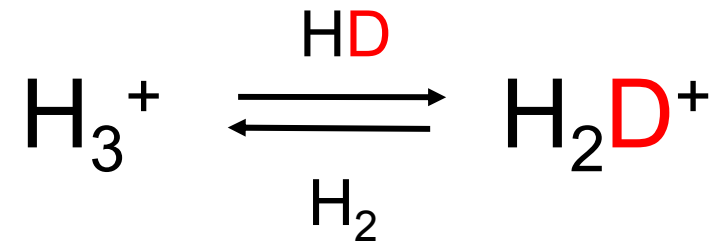
Astrophysical Observations



Isotopic Fractionation



Deuteration of H_3^+



Theory (Thermodynamics)

$[\text{HD}]/[\text{H}_2]$	$3 \cdot 10^{-4}$
T	10 K
S(T)	$3.6 \cdot 10^9$
$[\text{H}_2\text{D}^+]/[\text{H}_3^+]$	10^6

Experimental Method: Electrodynamical Trapping



Sandra Brünken

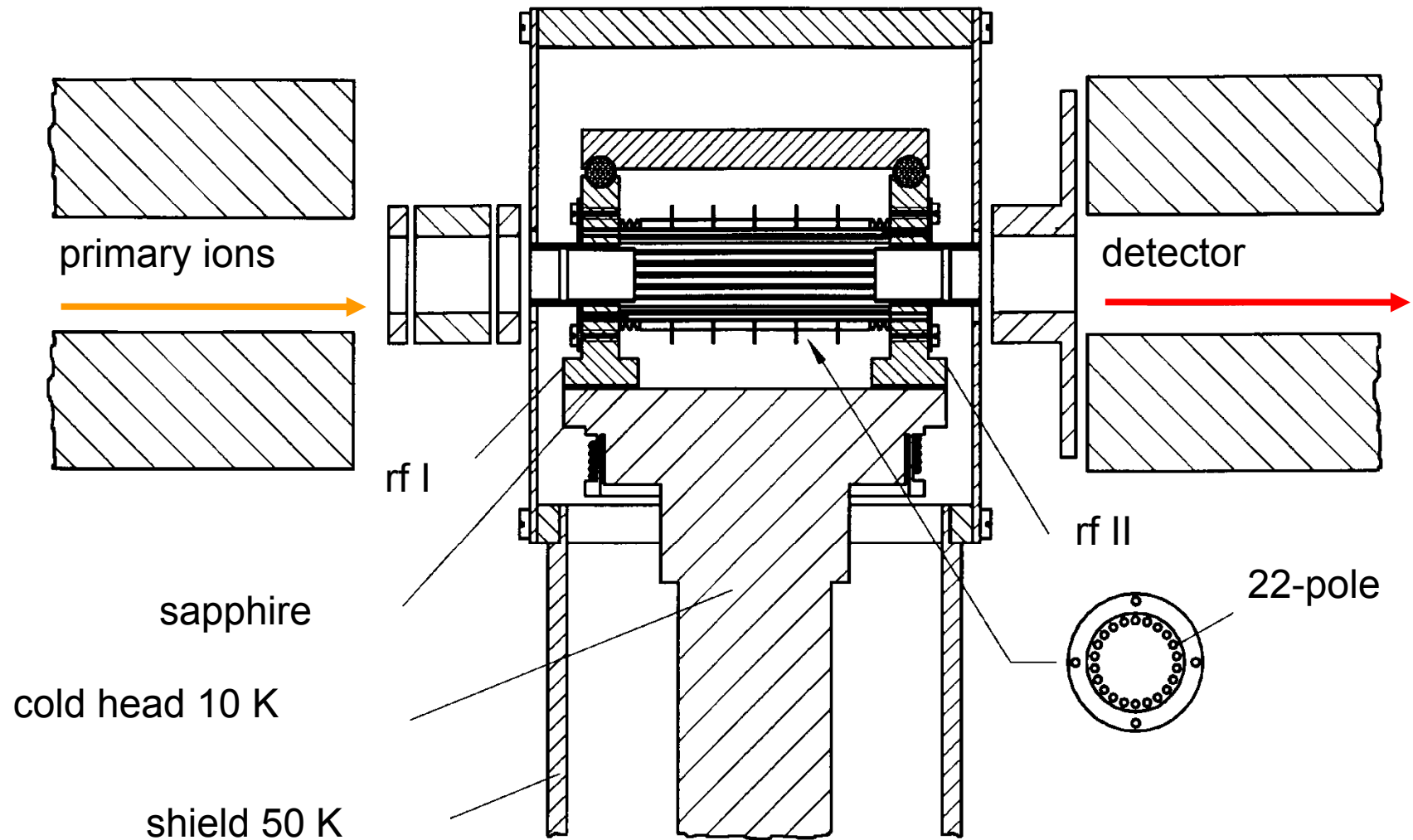


Dieter Gerlich

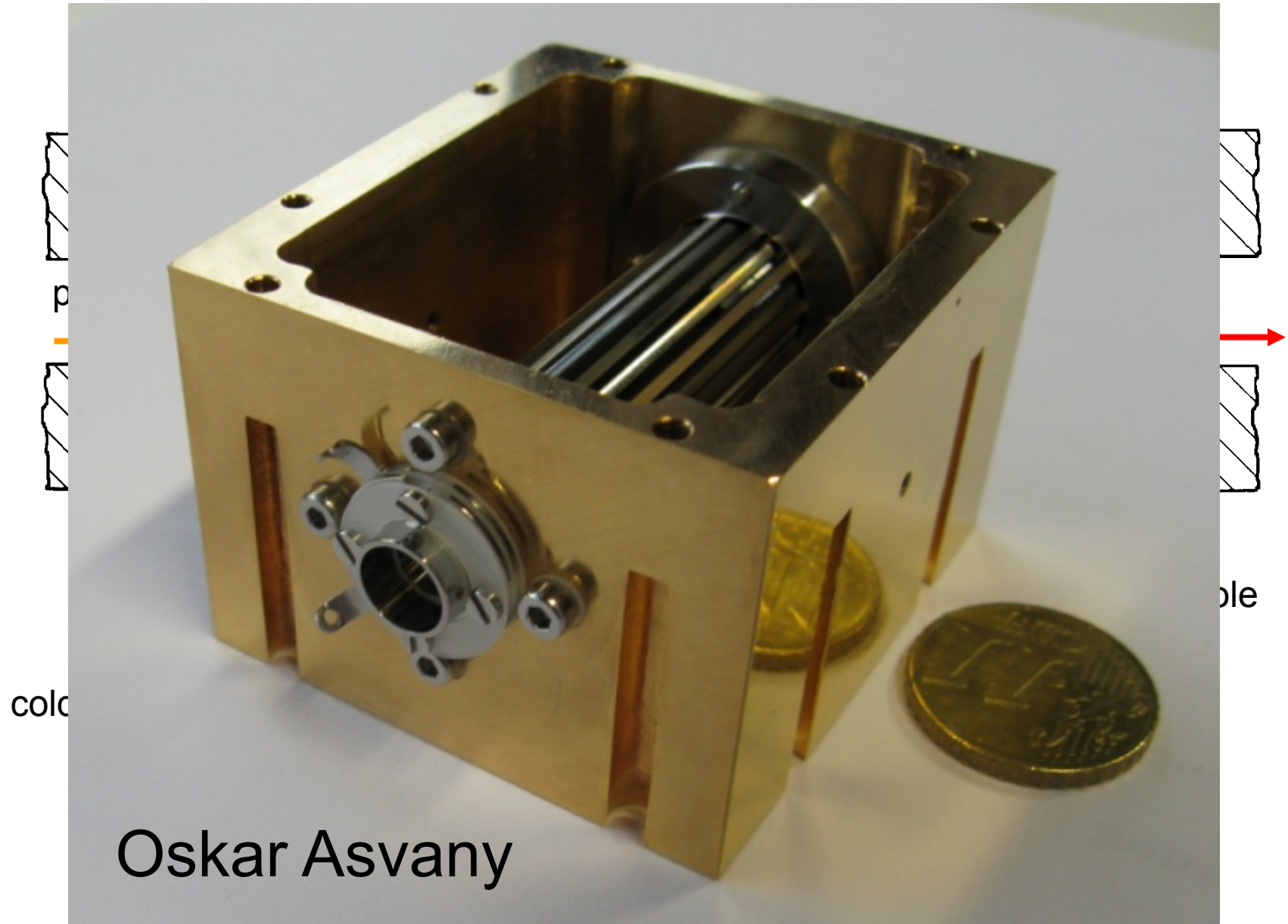


Oskar Asvany

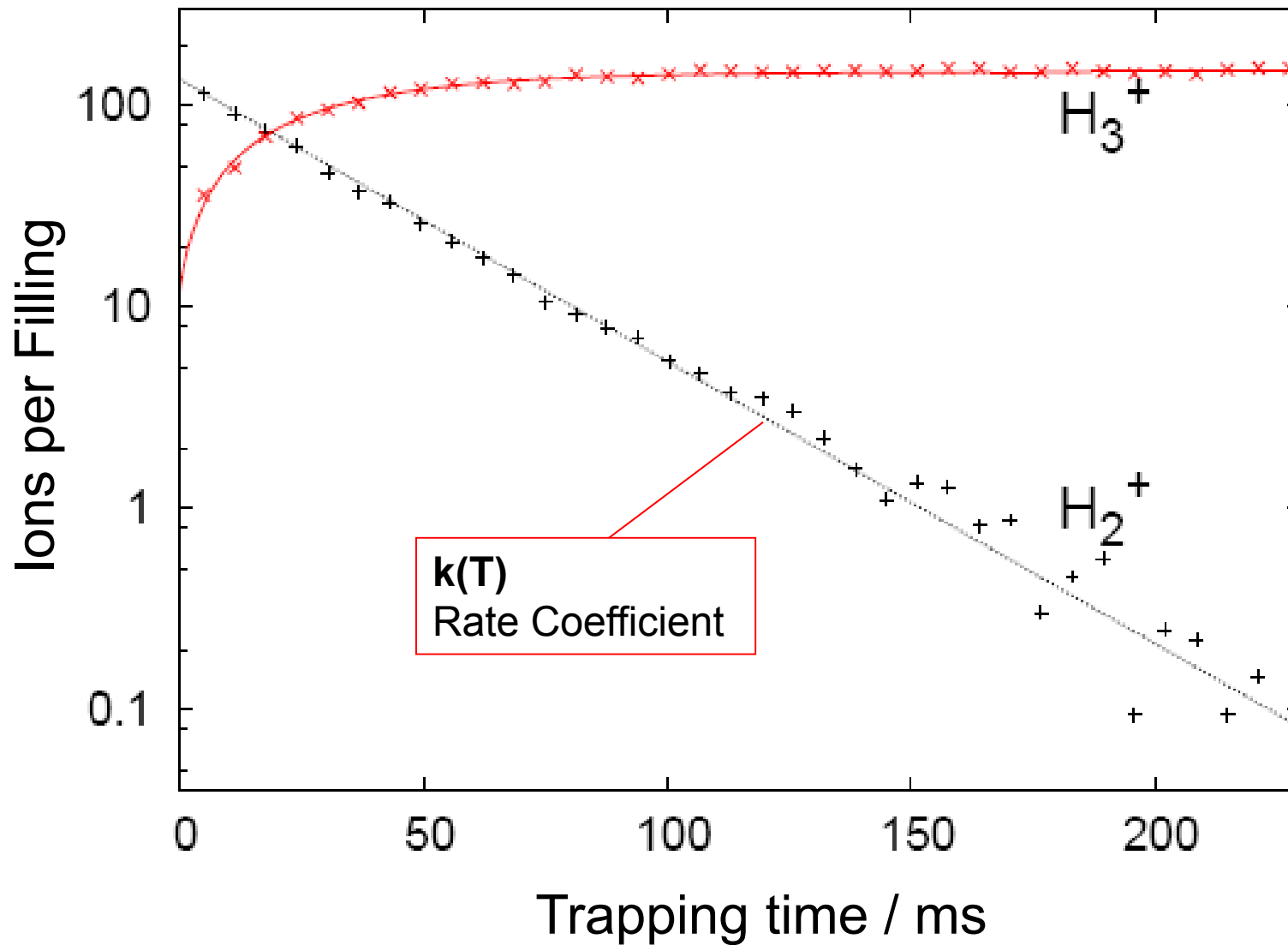
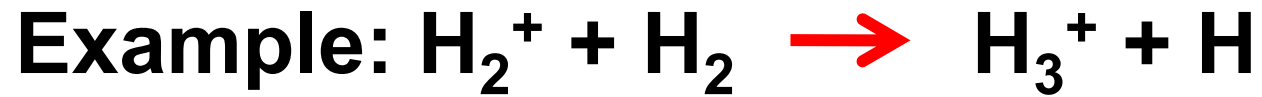
22-Pole Low Temperature Ion Trap



22-Pole Low Temperature Ion Trap

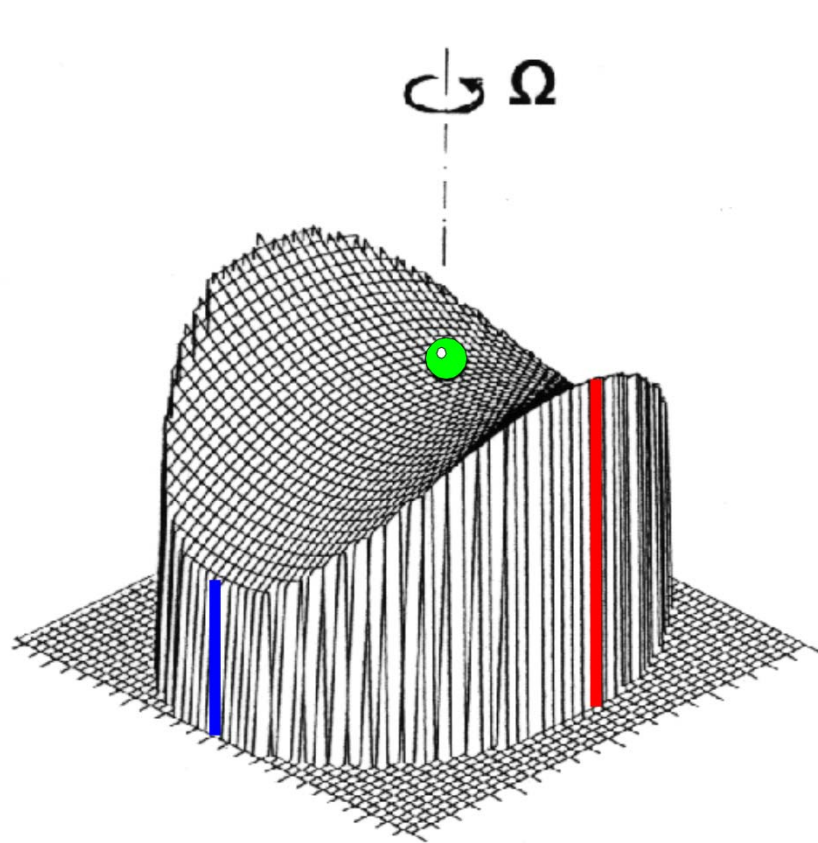


Oskar Asvany

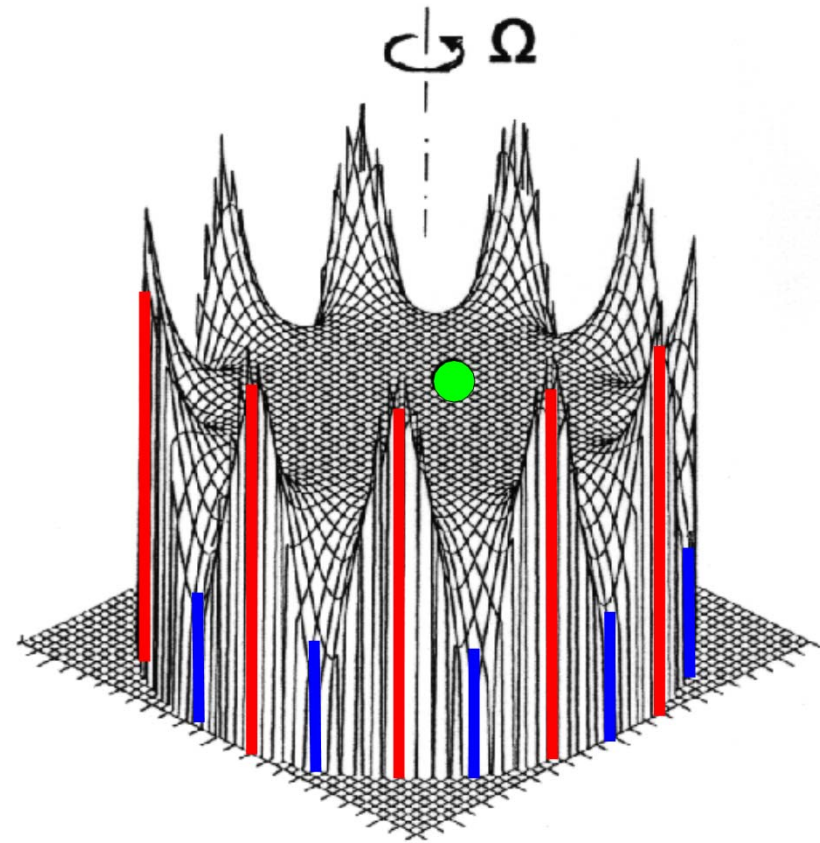


RF Ion Traps

Mechanical Model

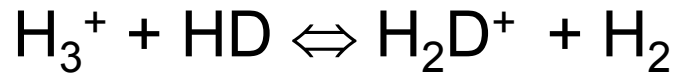


Quadrupole

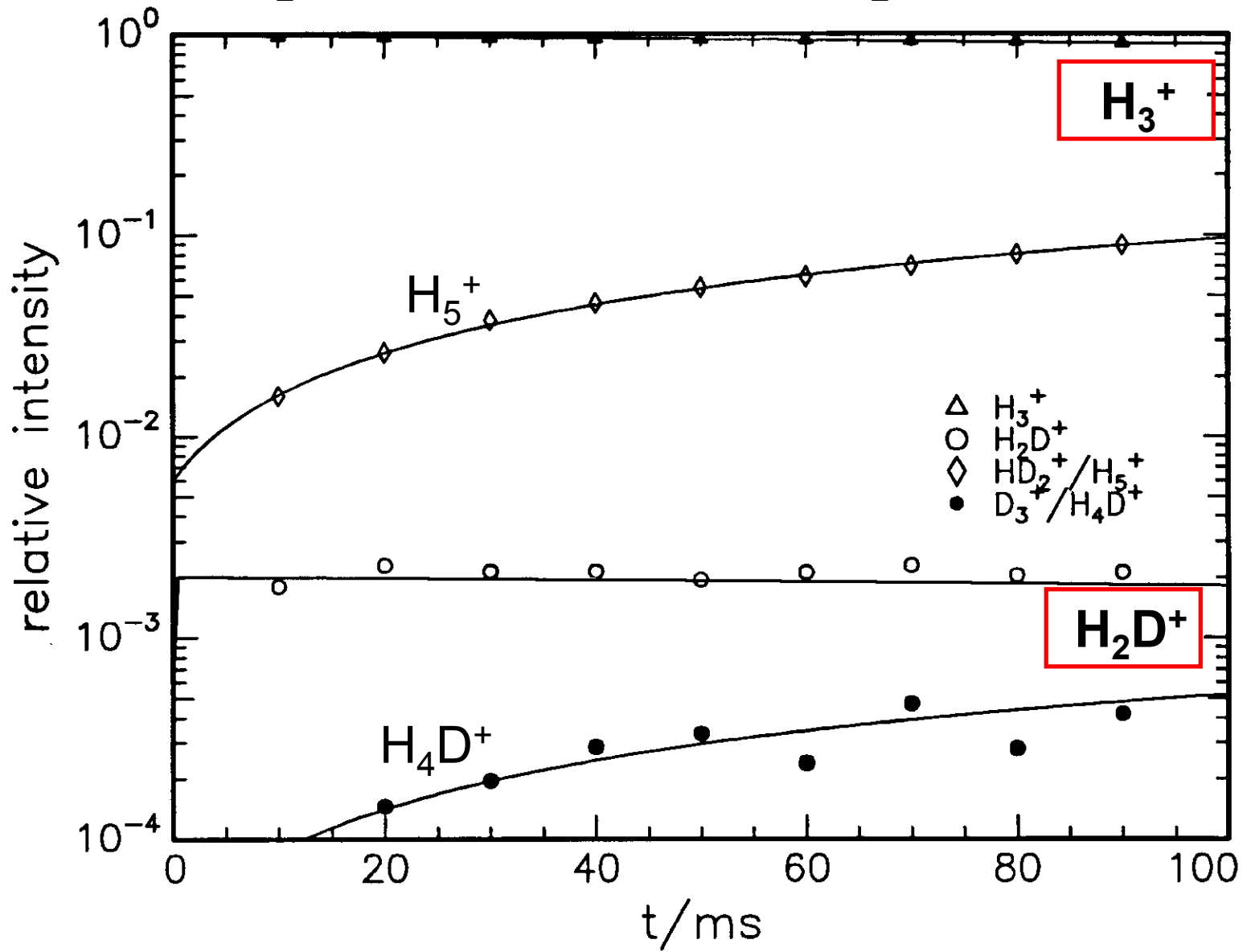


22-Pole

FAQ: Why 22 poles?

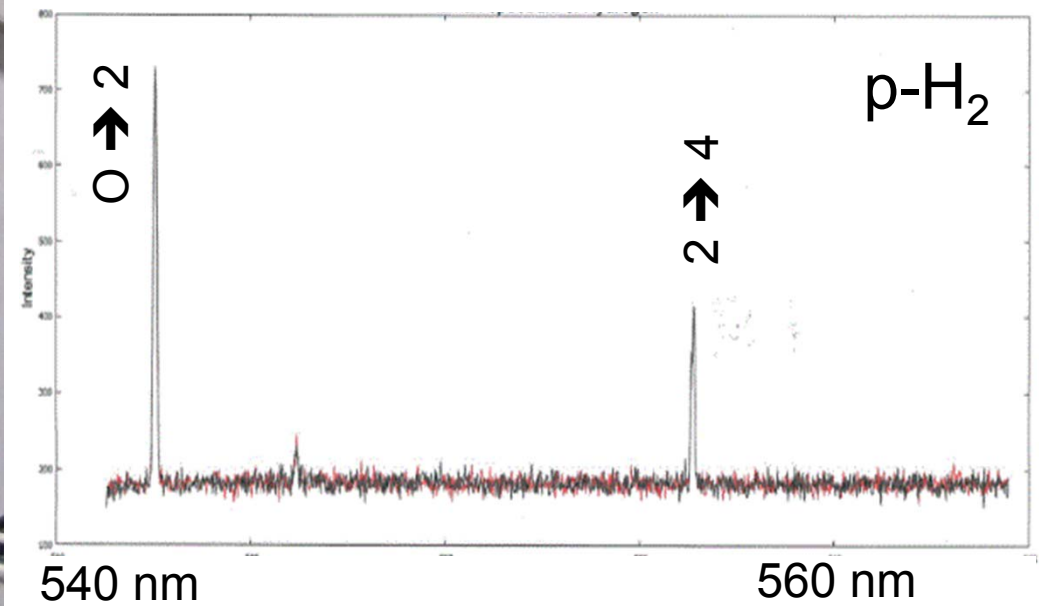
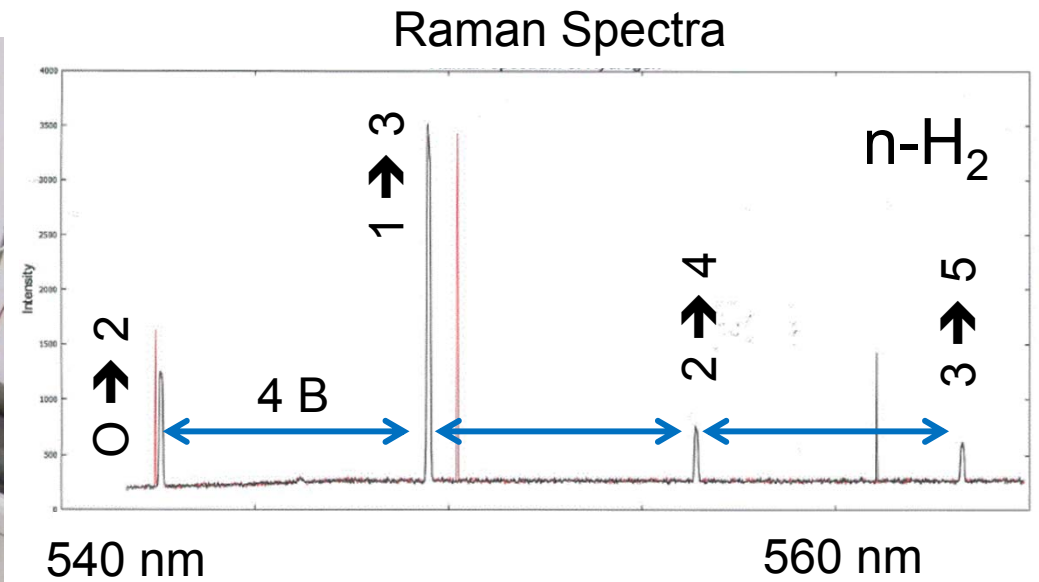
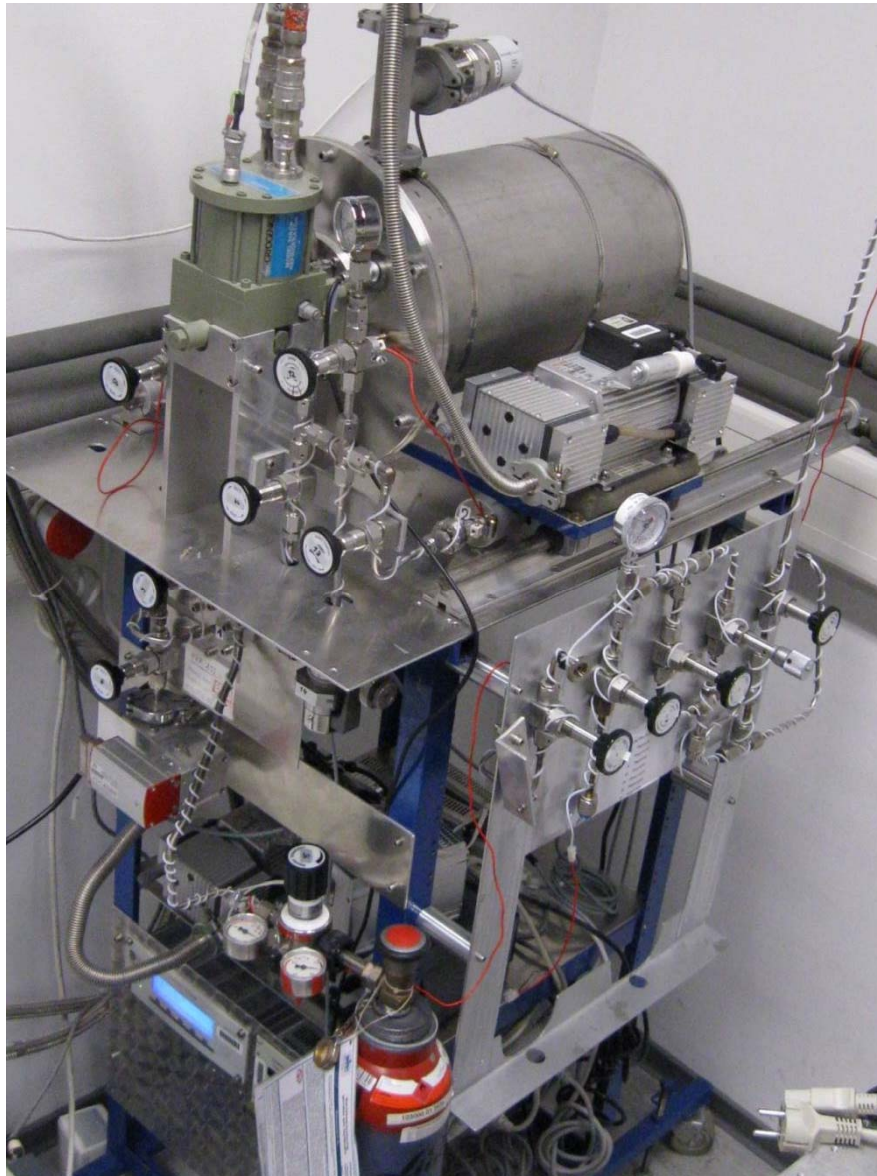


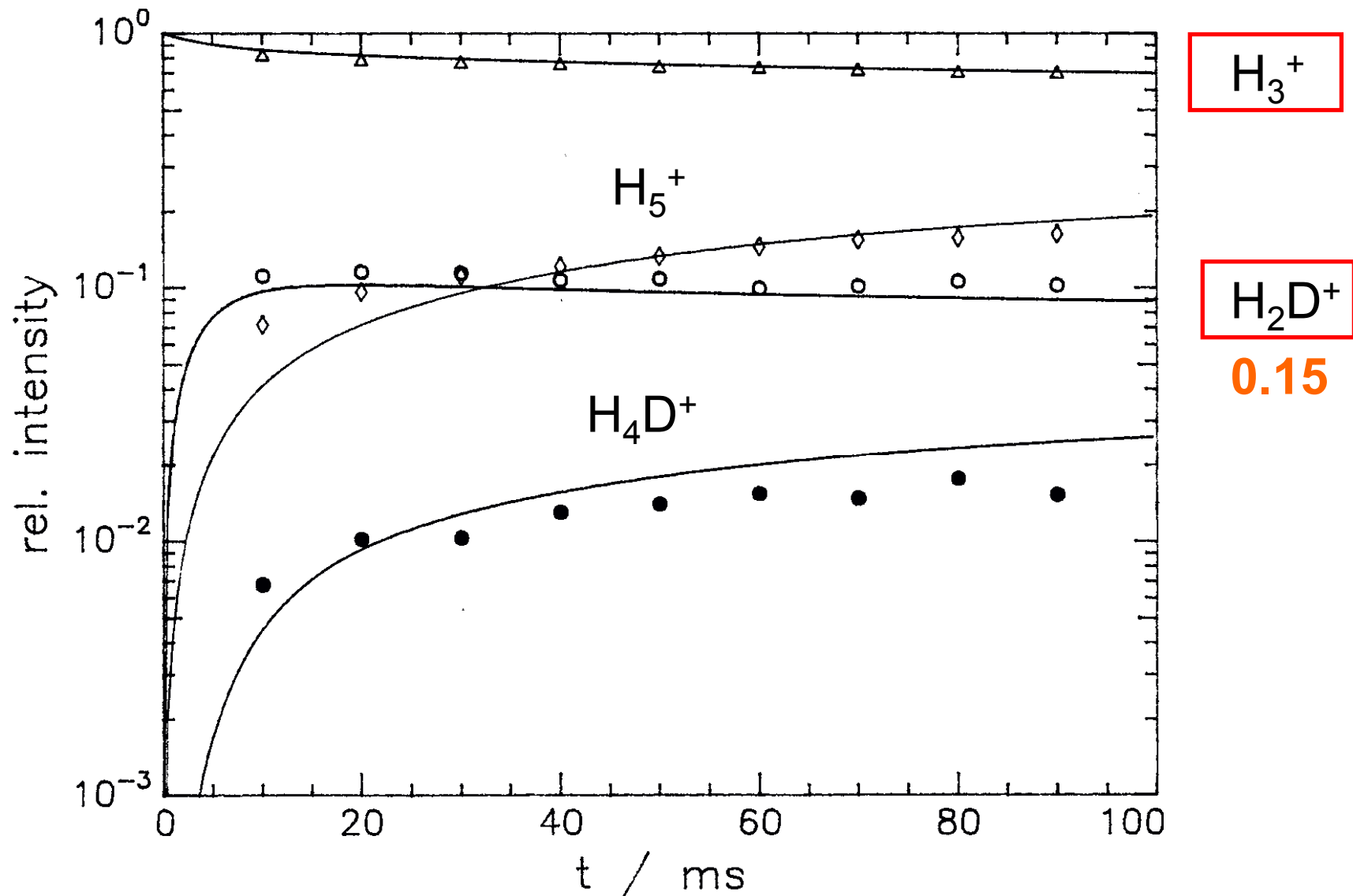
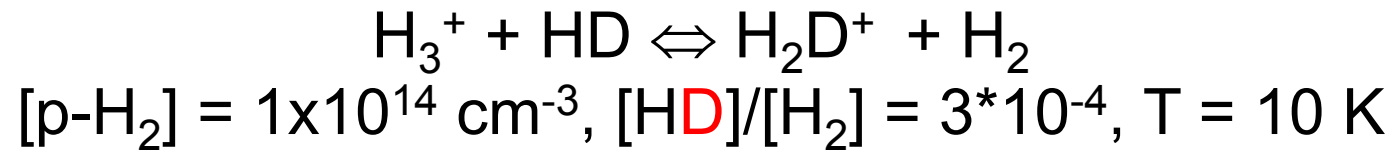
$[n\text{-H}_2] = 1.4 \times 10^{14} \text{ cm}^{-3}$, $[\text{HD}]/[\text{H}_2] = 3 \times 10^{-4}$, $T = 10 \text{ K}$



0.002

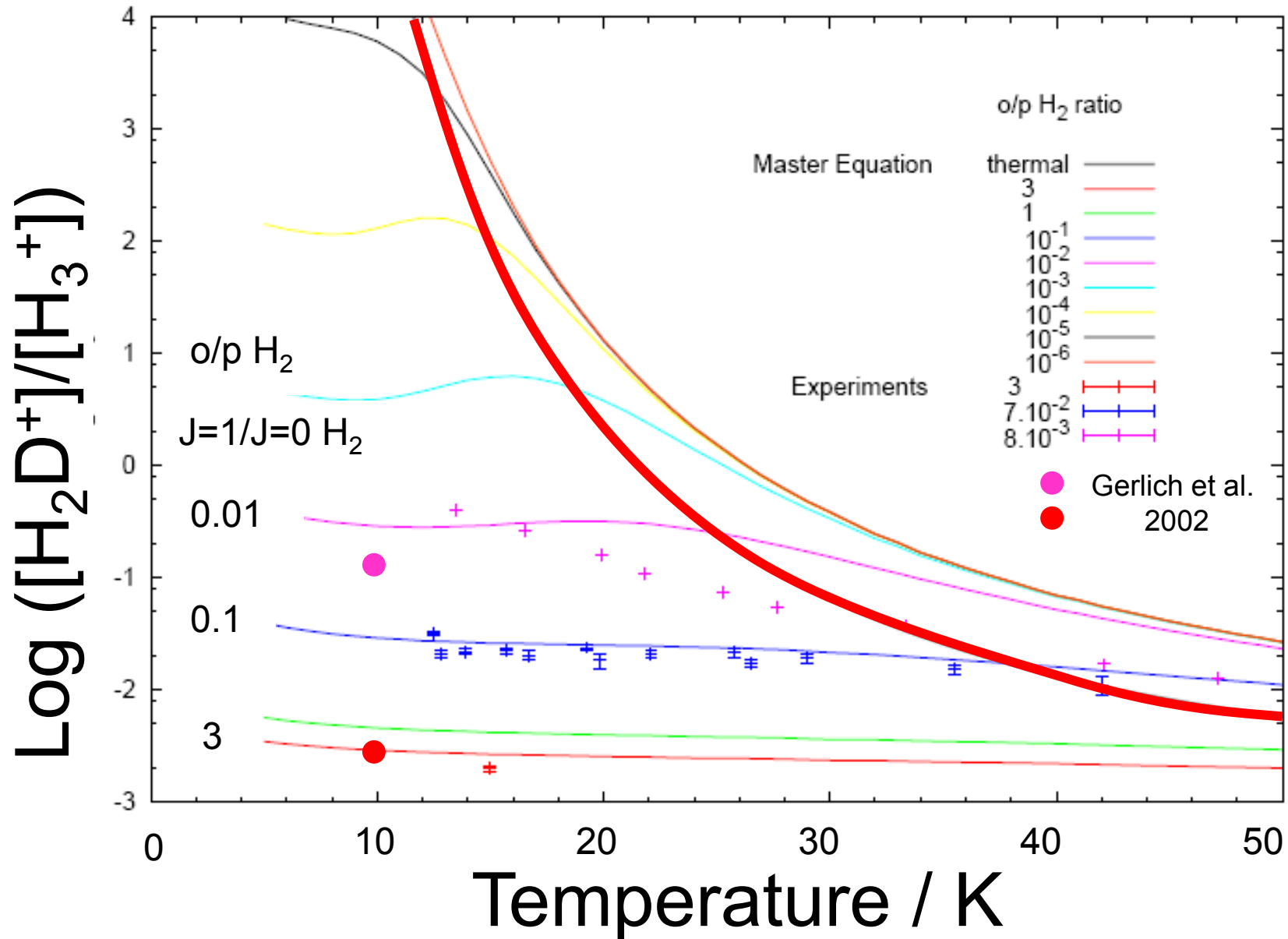
Current Experiments with para-H₂ (J=0,2,...)



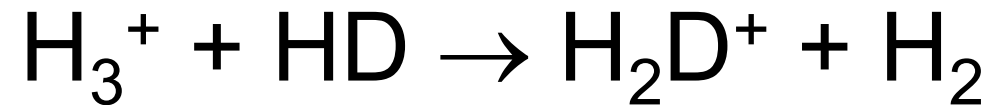


Experimental Results & Modelling

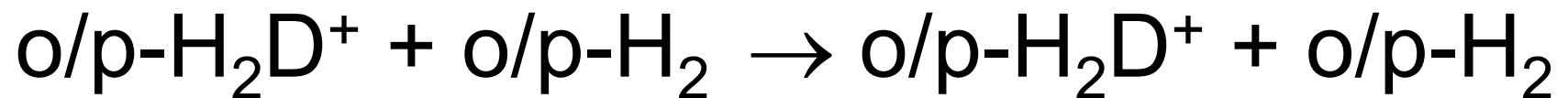
Hugo et al., J.Chem.Phys. 2009, **130**, 164302

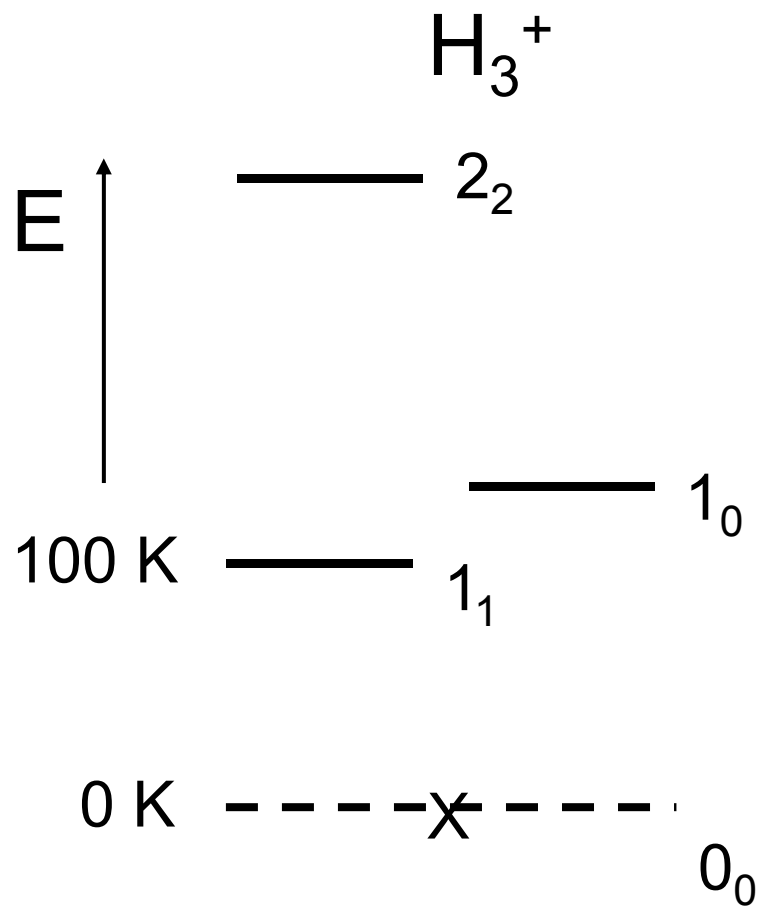


Isotopic Fractionation



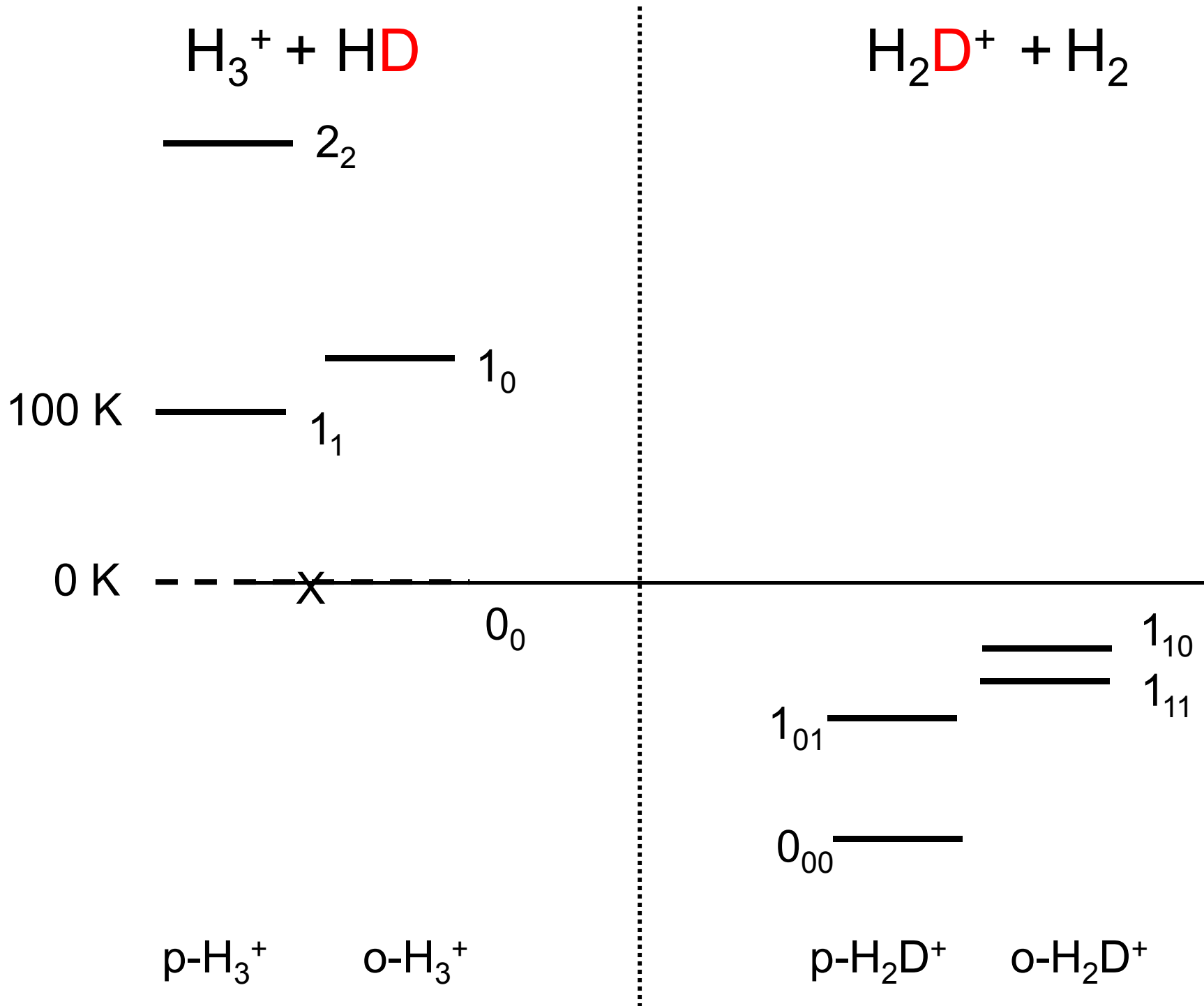
and the $\text{H}_2 / \text{H}_2\text{D}^+$ OPR

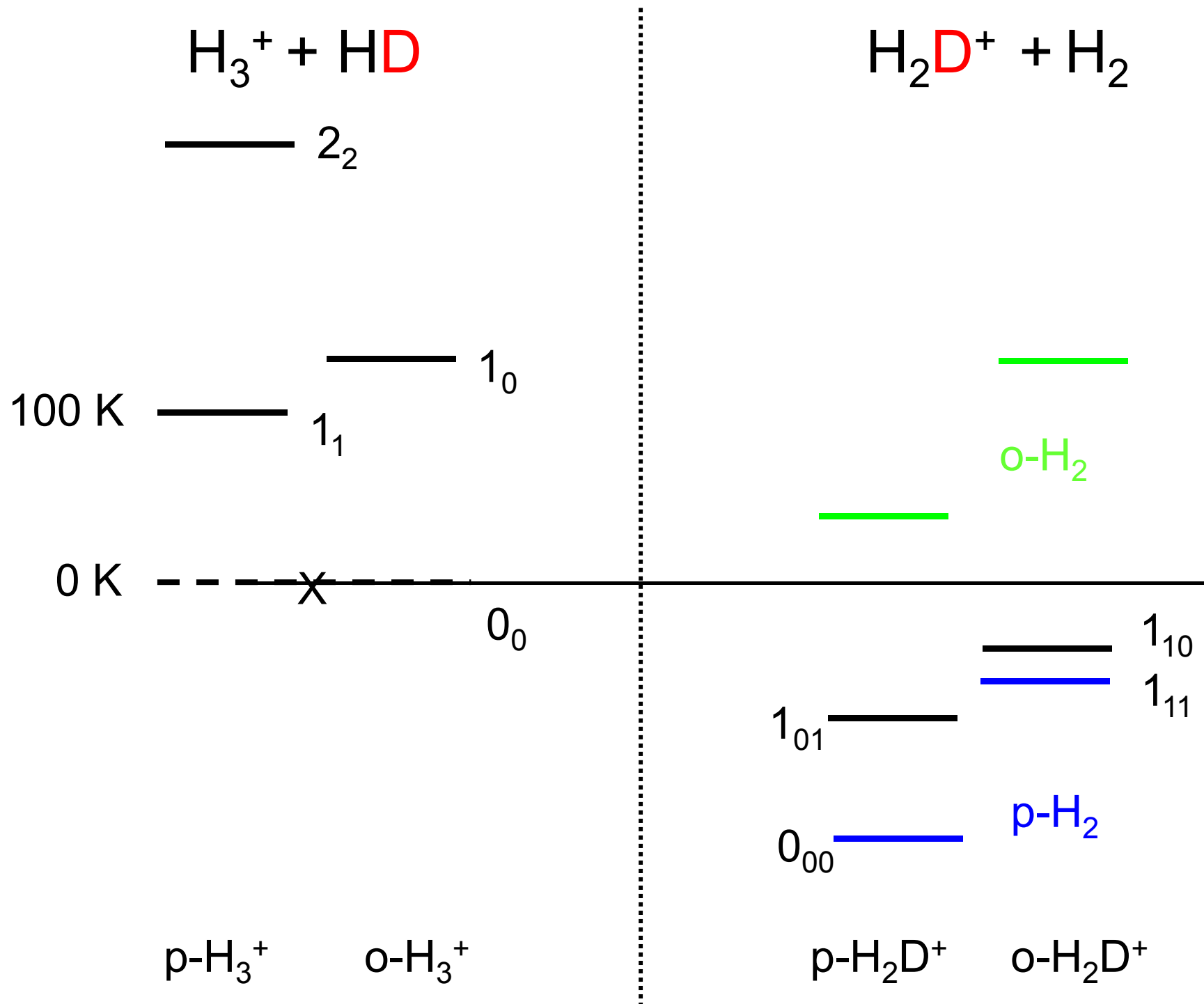




Lowest energy levels of H_3^+

$p\text{-H}_3^+$ $o\text{-H}_3^+$

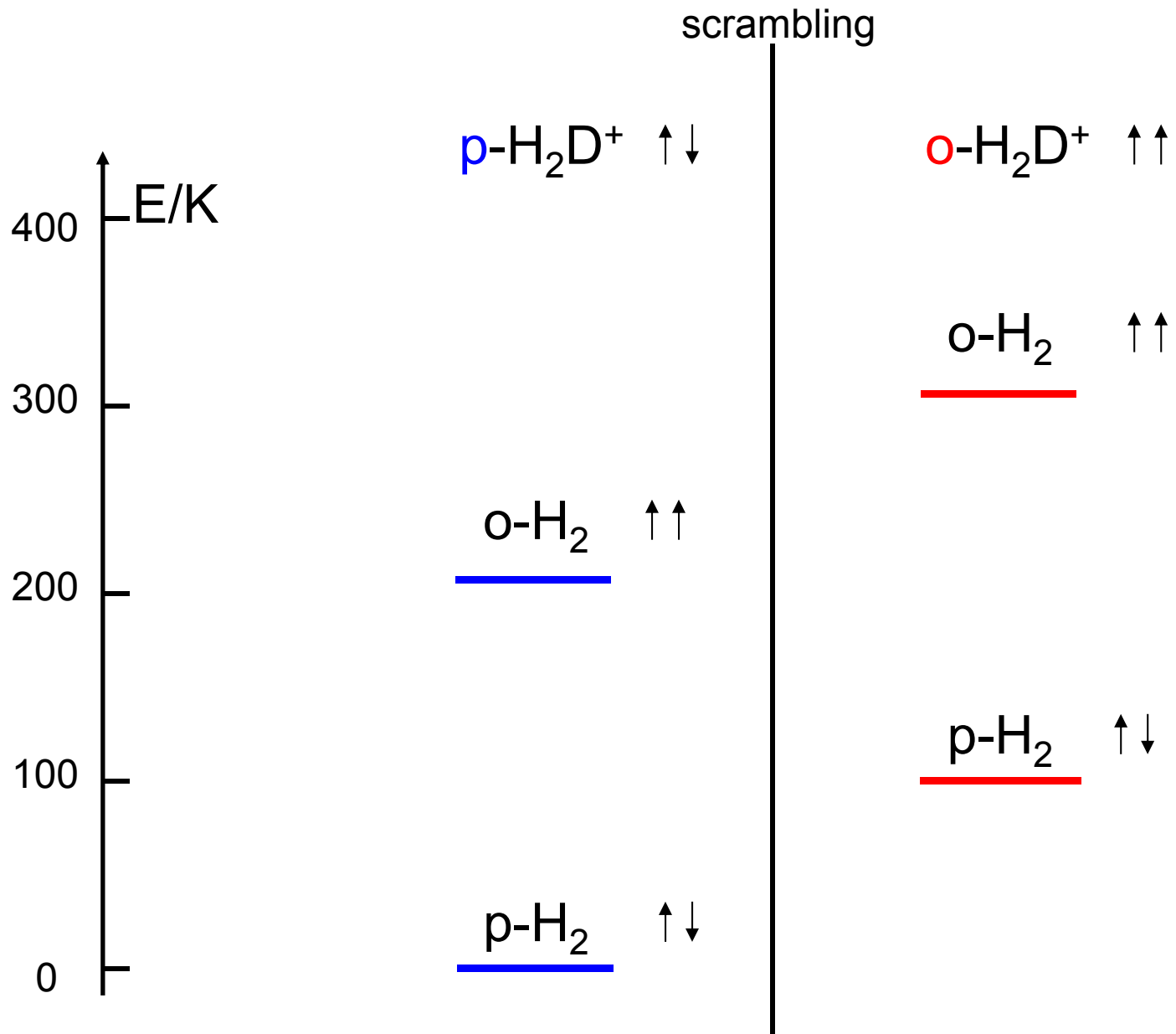




Role of Nuclear Spin?

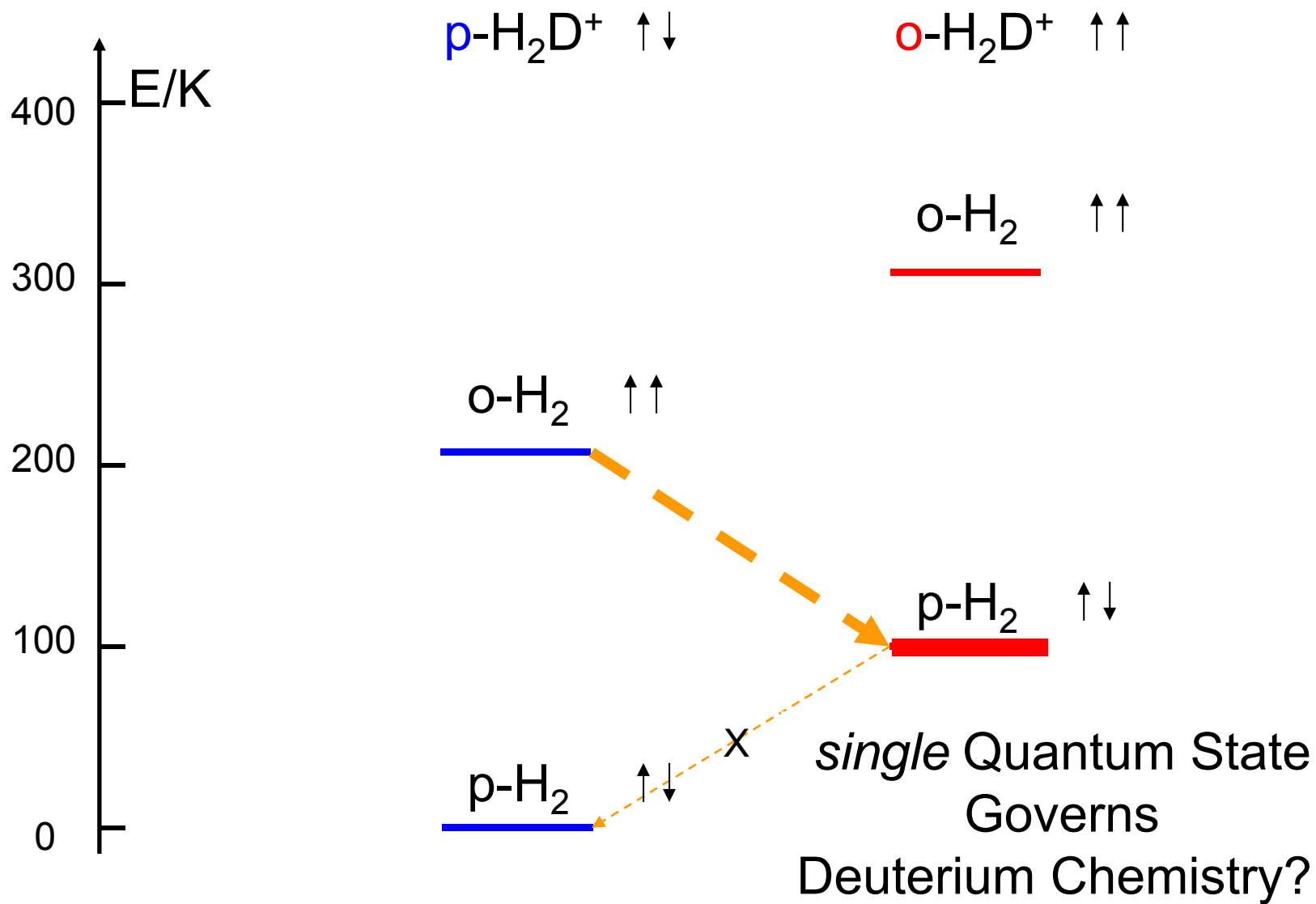
Conservation laws: E, J, P, I, ...

Para – Ortho Conversion



Para – Ortho Conversion

scrambling



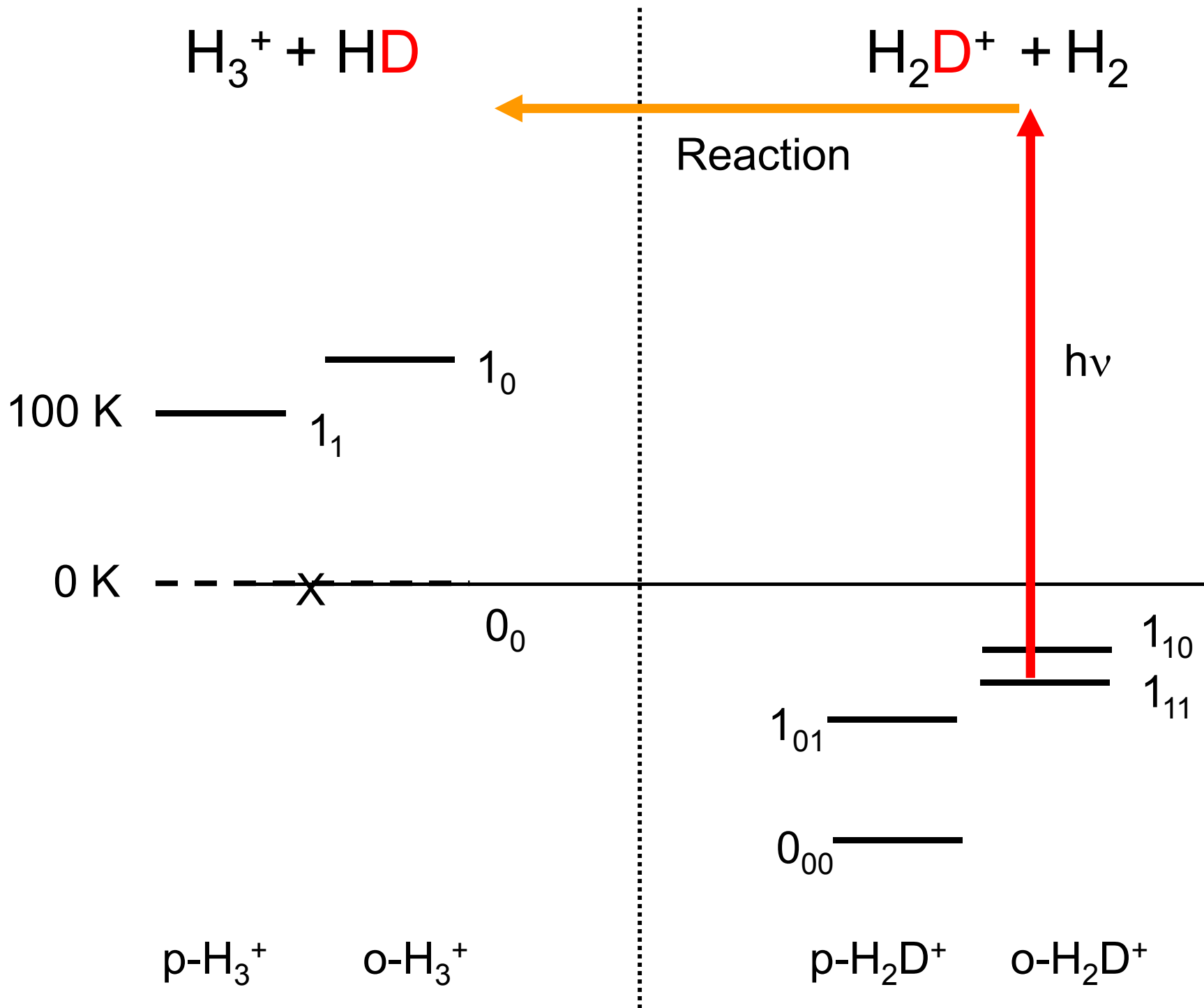
Laboratory Approach

H_2D^+ State Distributions

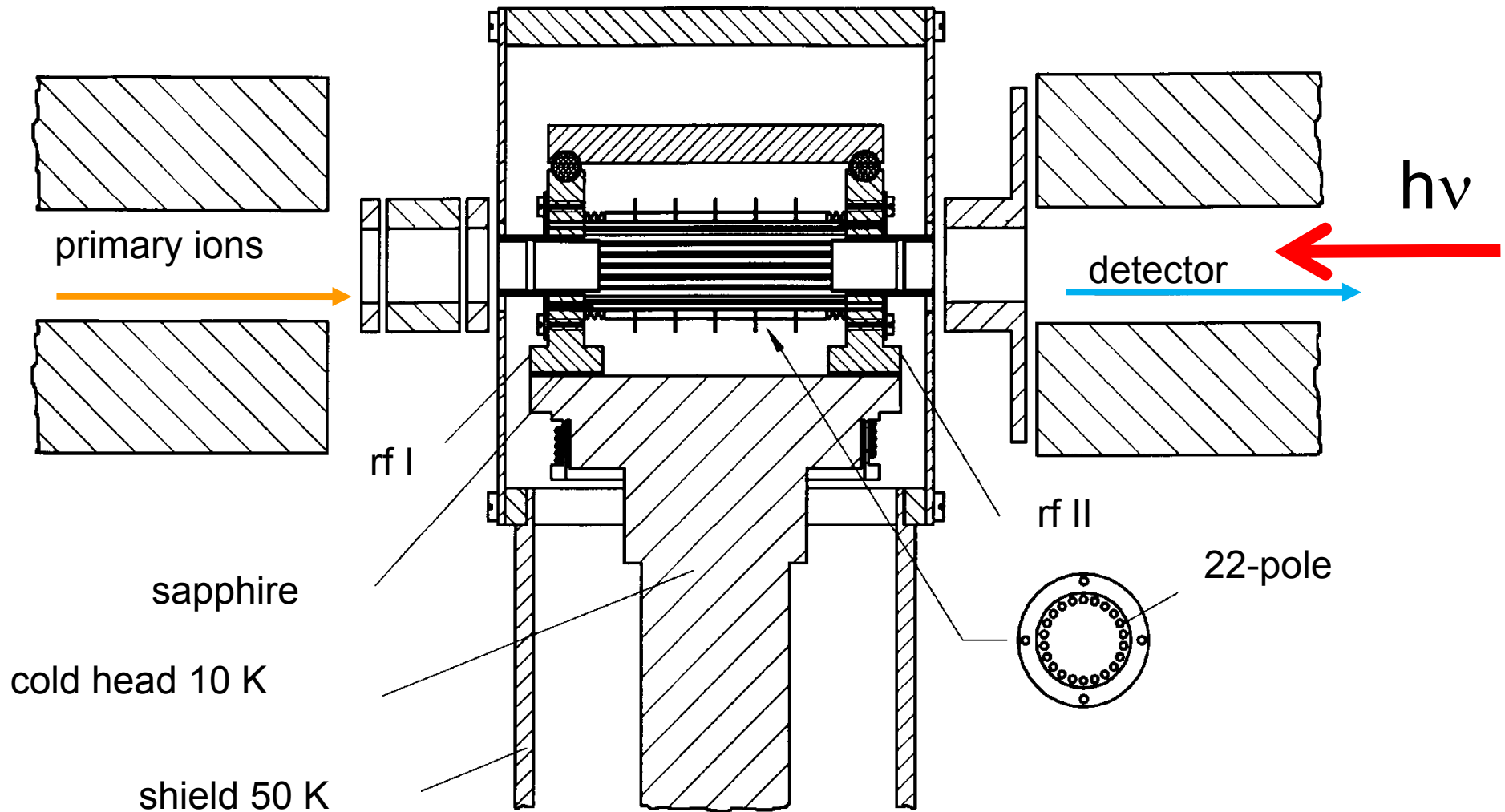
Translation

Rotation

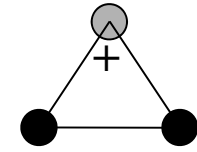
Nuclear Spin



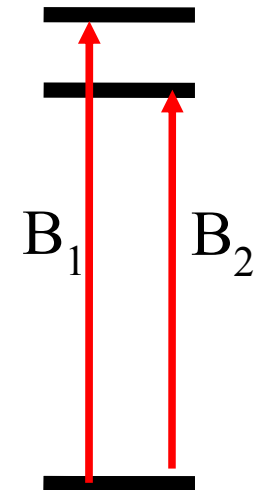
Light Induced Reactions Spectroscopy in Traps



relative B coefficients



transition	line position / cm^{-1}	laser power / mW	meas B_{rel}	calc B_{rel}
H_2D^+				
$0_{00} \rightarrow 1_{11}$	6466.532	1.8	1	1
$0_{00} \rightarrow 1_{01}$	6330.973	4.0	0.32 ± 0.02	0.31
$1_{11} \rightarrow 1_{10}$	6303.784	5.0	0.29	0.29
$1_{11} \rightarrow 0_{00}$	6340.688	5.3	0.27 ± 0.03	0.27
$1_{11} \rightarrow 2_{02}$	6459.036	4.1	0.35 ± 0.04	0.34
D_2H^+				
$0_{00} \rightarrow 1_{11}$	6536.319	1.6	1	1
$0_{00} \rightarrow 1_{11}$	6482.033	3.8	0.33 ± 0.02	0.32



$$\text{signal} \sim B \cdot \text{pop} \cdot P \cdot k^*$$

conclusions:

- 1) *ab initio* predicted (relative) B coefficients reliable
- 2) reaction probability independent of rovib. overtone excitation

Rotational Level Populations of H₂D⁺

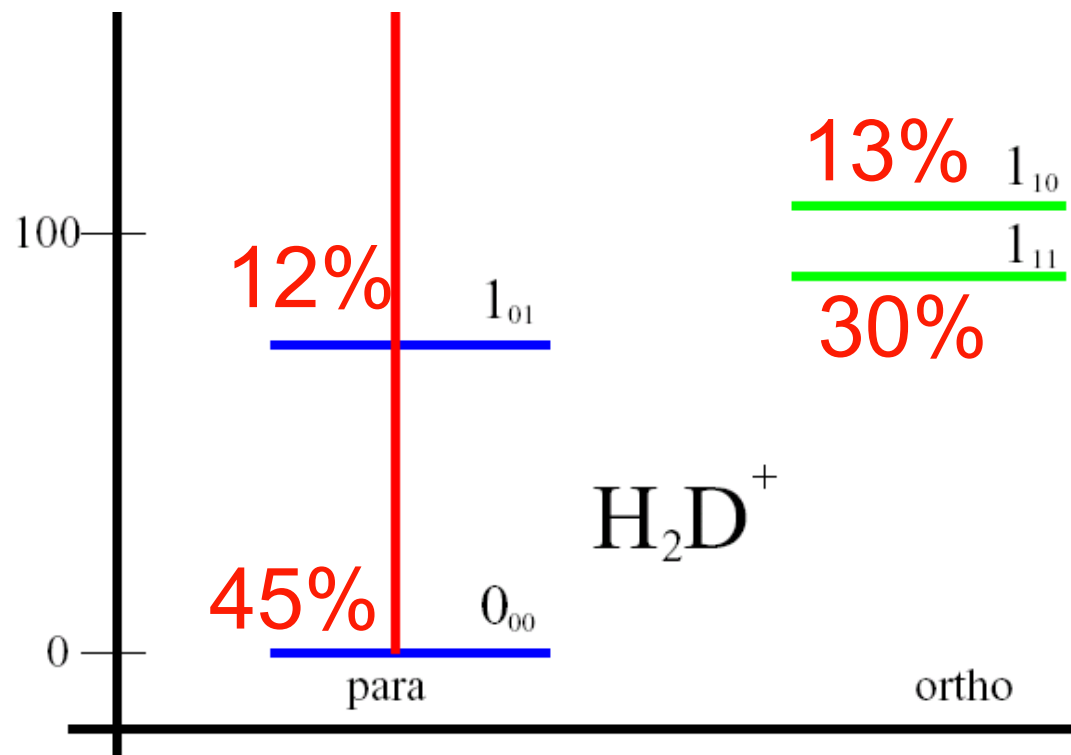
$$\text{pop} \sim \frac{[H_3^+]}{B \cdot P}$$

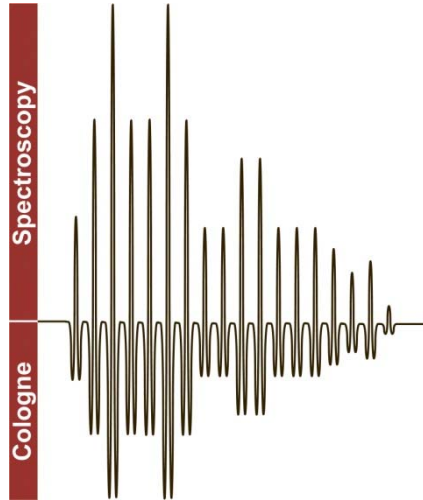
$$T_{\text{Doppler}} = (27 \pm 2) \text{K}$$

$$T_{\text{rot,para}} = (27 \pm 2) \text{K}$$

$$T_{\text{ortho/para}} = (35) \text{K}$$

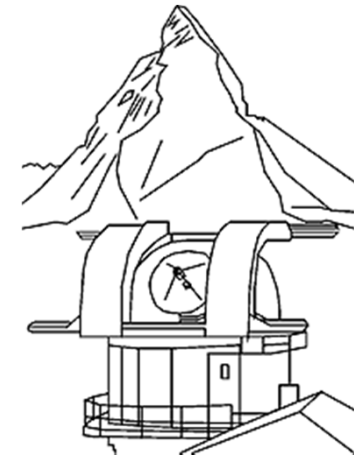
energy level / K





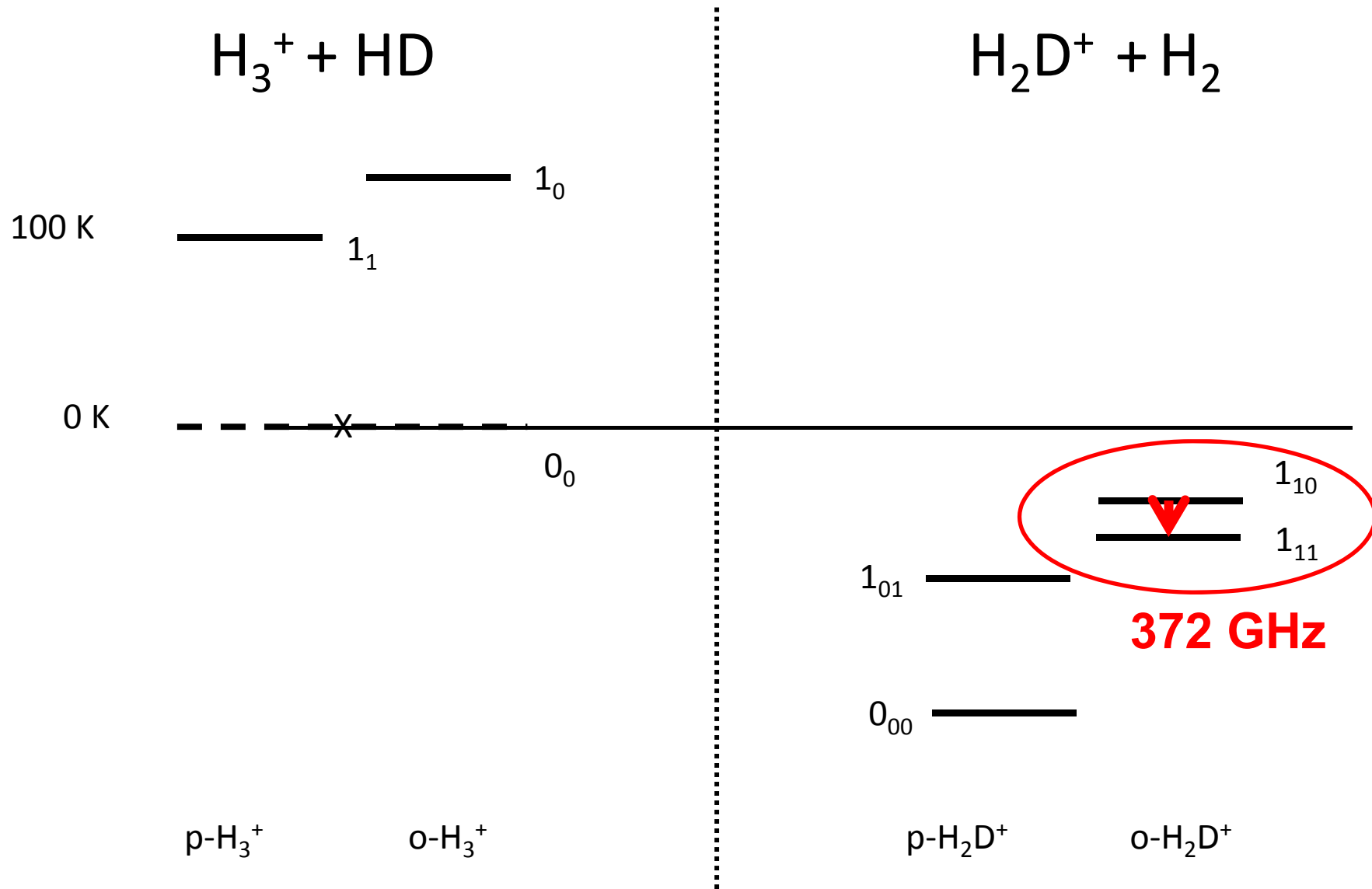
Cold Chemistry in Space and Laboratory

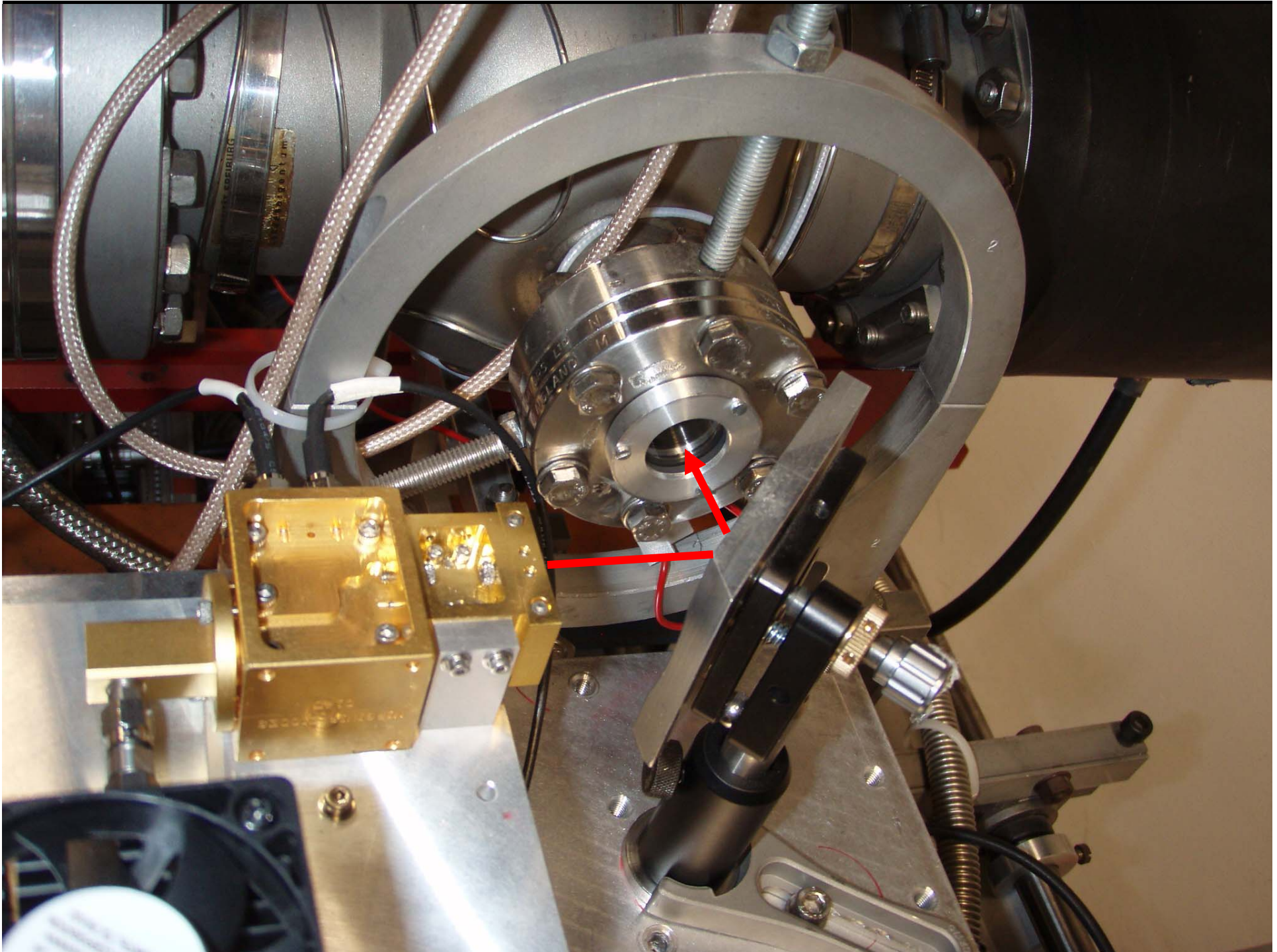
Stephan Schlemmer
Universität zu Köln



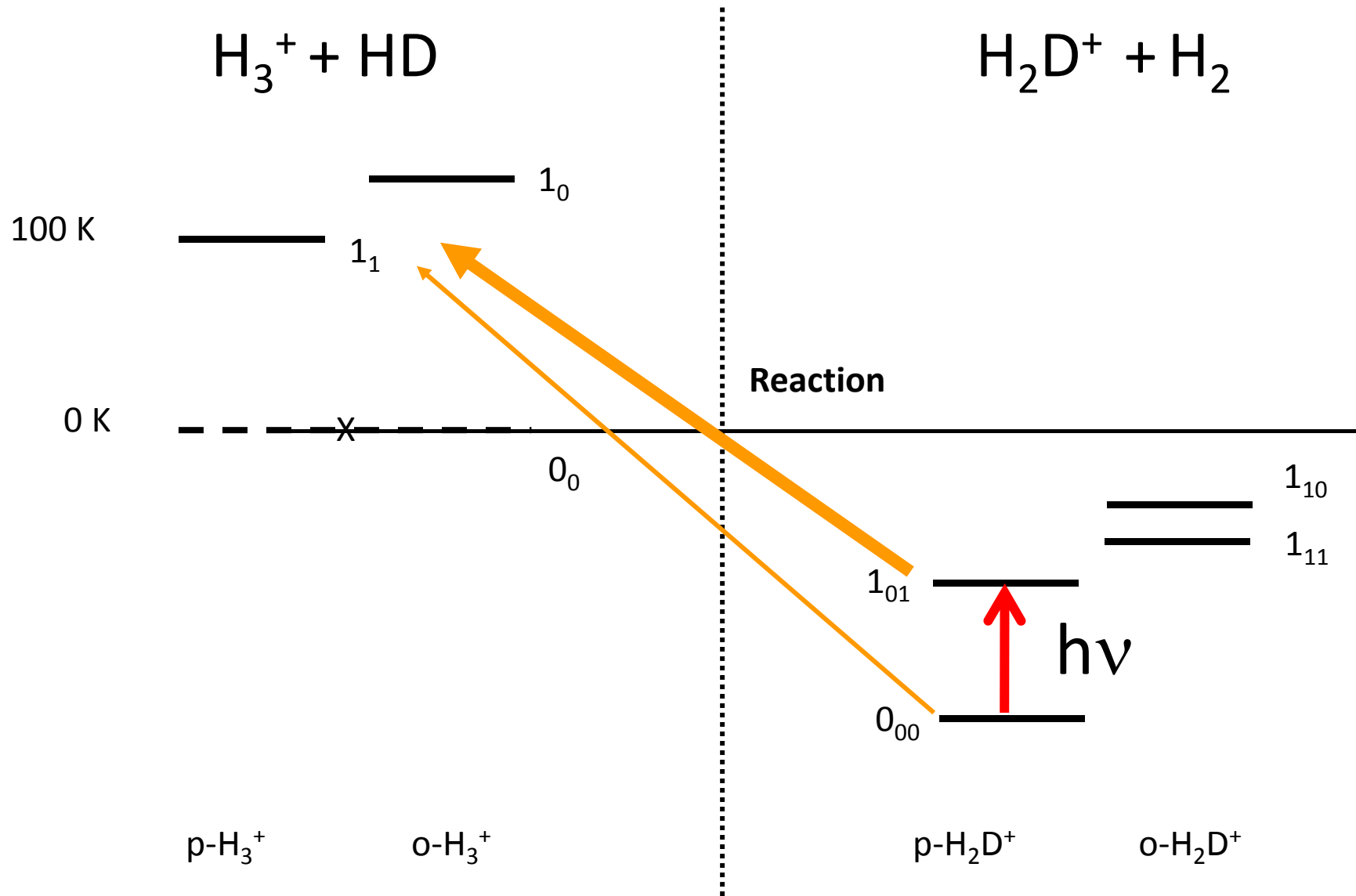
- H₂ Formation, OPR and Chemical Clocks
- H₃⁺ / H₂D⁺ Isotopic Fractionation, H₃⁺/H₂D⁺, OPR
- H₂D⁺ + H₂ THz Spectroscopy in Lab and Space

H₂D⁺ Detection in Space

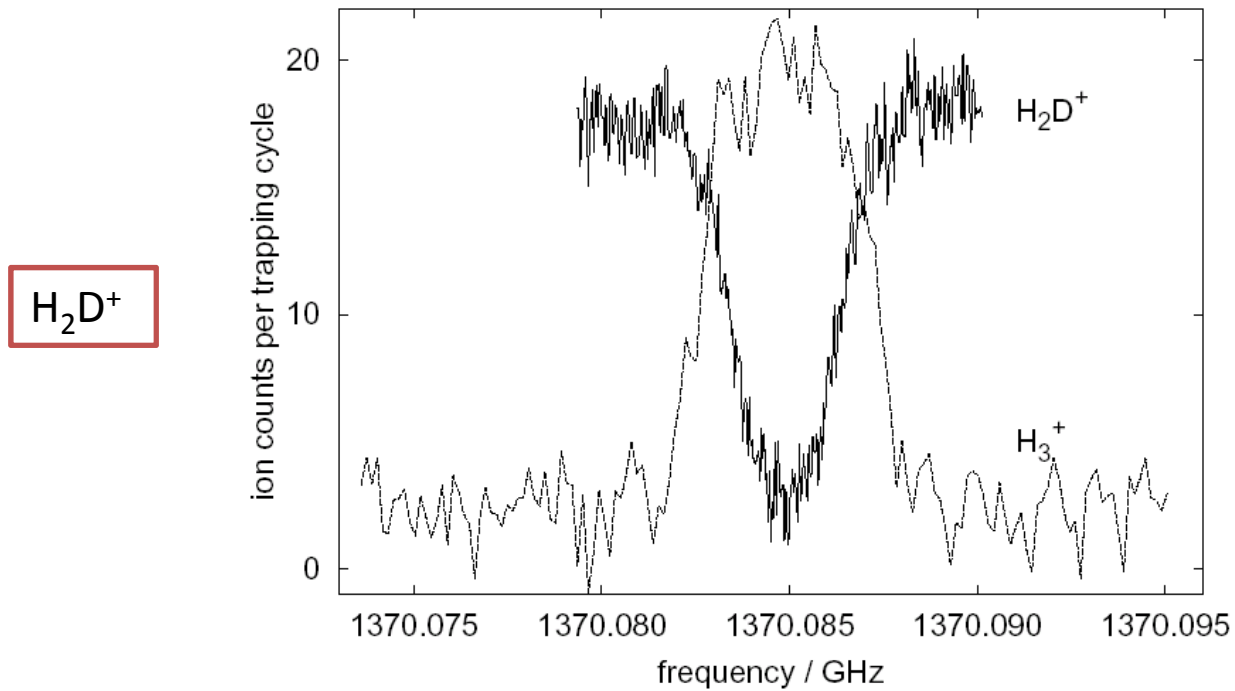




Light Induced Reactions probing H_2D^+



Results



	$\text{H}_2\text{D}^+ 1_{01} \leftarrow 0_{00}$
this work	1370084.880(20)
<i>ab initio</i> ^a	1369991.8
unpublished value ^b	1370146.0(3)

SOFIA



GREAT
Receiver

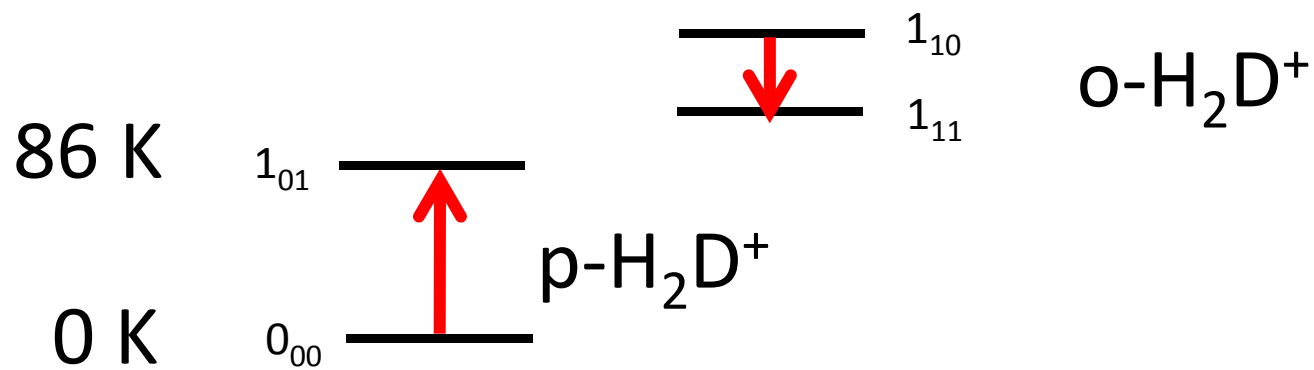
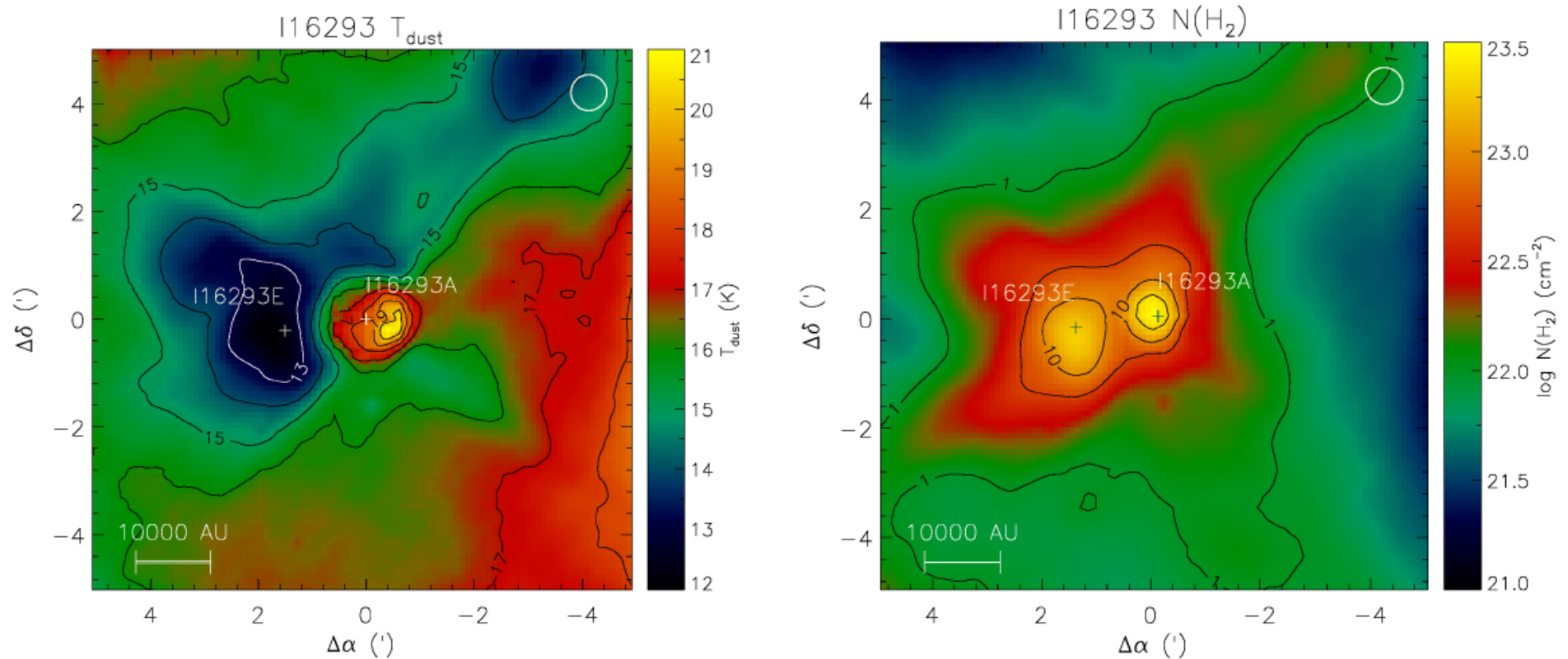
U_{00}

380(20)

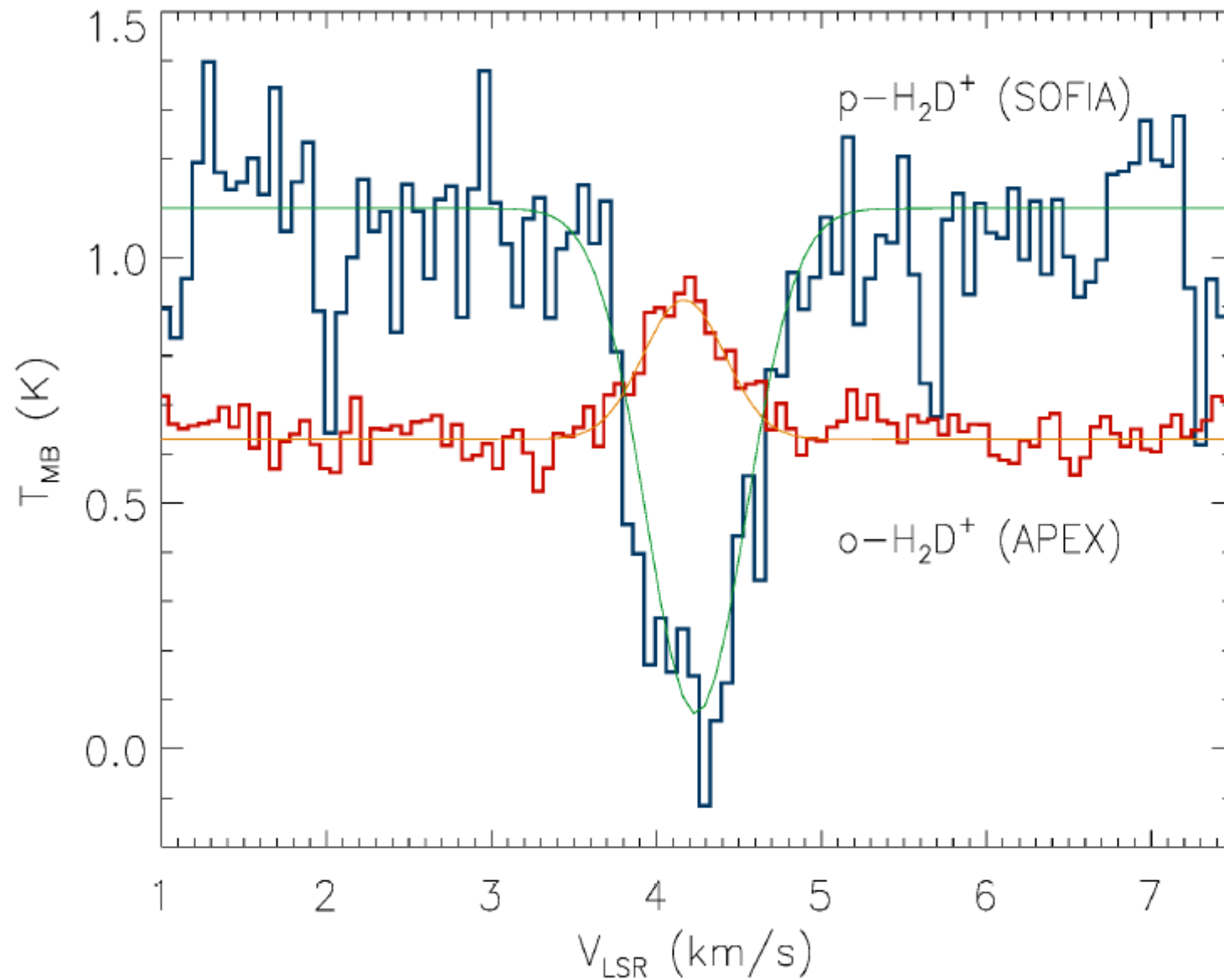
0(3)

$\Delta = 62 \text{ MHz (!)}$

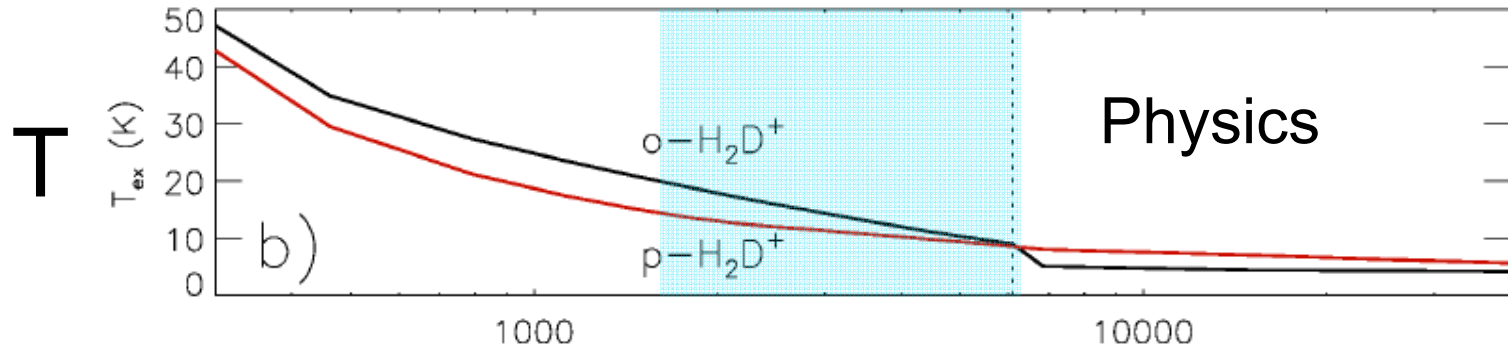
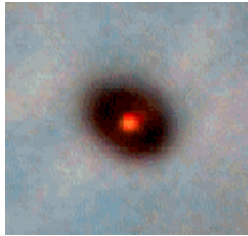
Protostellar Cloud Core I16293A



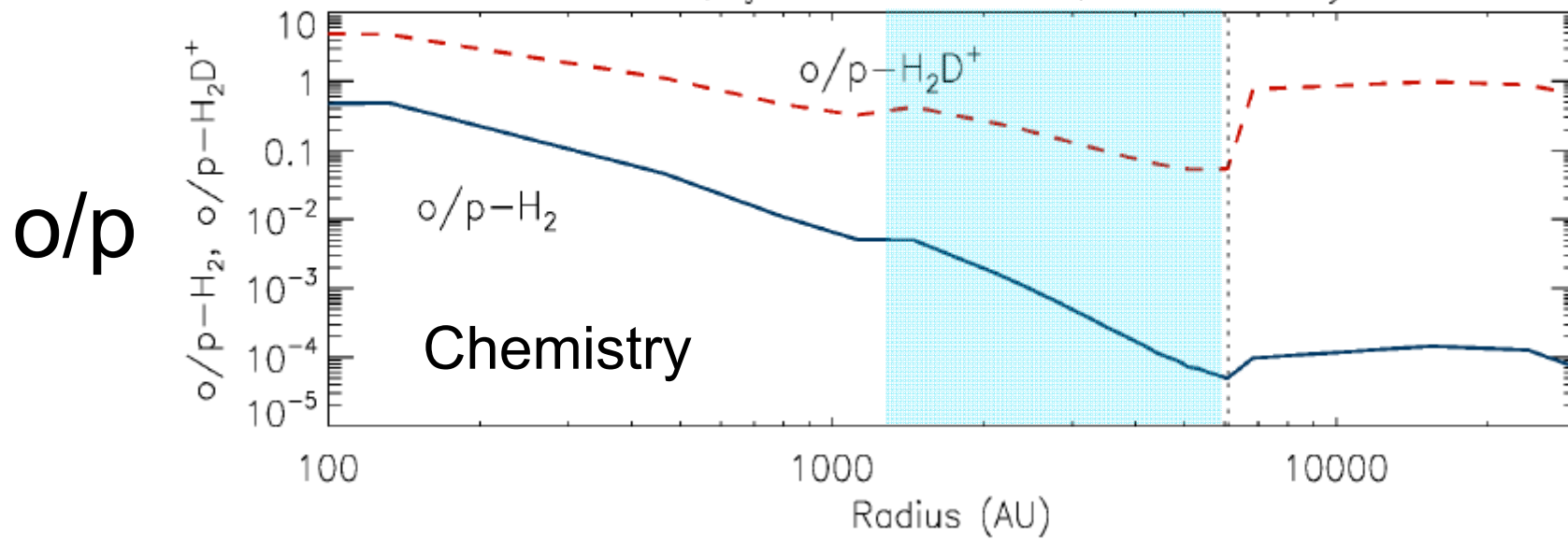
Para- H_2D^+ found in Space

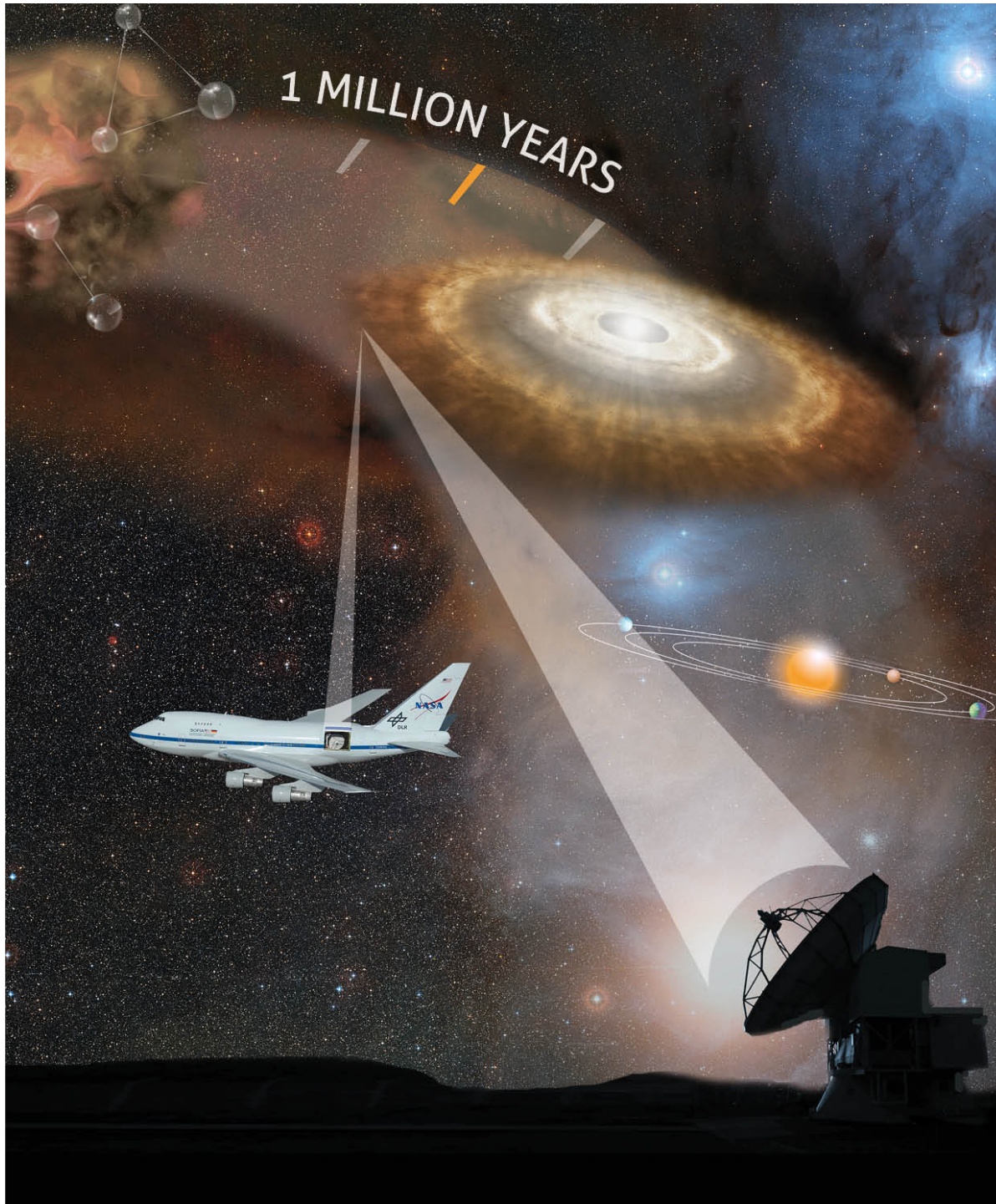


Astrochemical Modelling



Model 5, $\zeta = 1.3 \cdot 10^{-17} \text{ s}^{-1}$, time = 10^6 yr

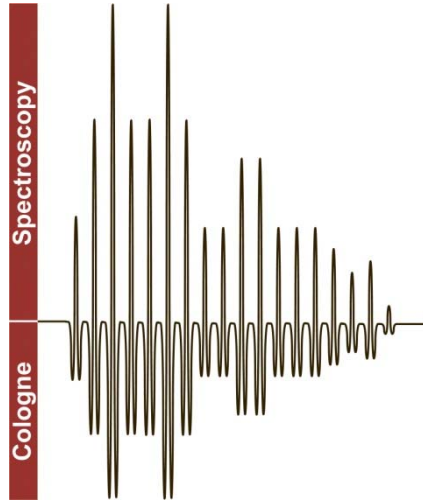




H_2D^+ observations
give an age of at least
one million years for a
cloud core forming
Sun-like stars

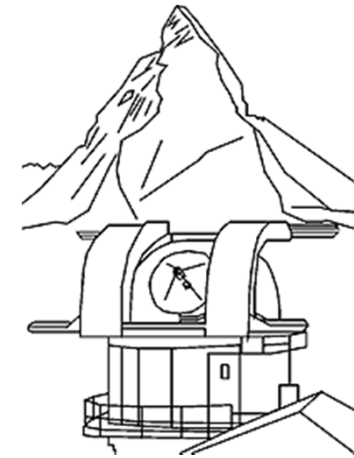
S. Brünken et al.
Nature

doi:10.1038/nature13924

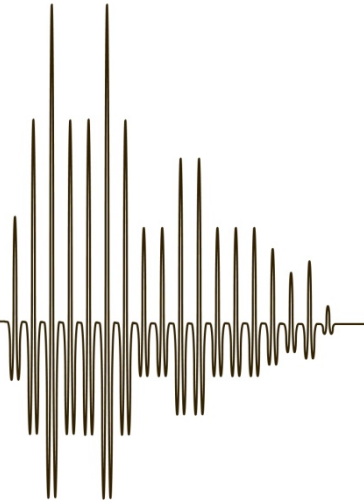


Cold Chemistry in Space and Laboratory

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Universität zu Köln



- H_2 Formation, OPR and Chemical Clocks
- $\text{H}_3^+ / \text{H}_2\text{D}^+$ Isotopic Fractionation, $\text{H}_3^+ / \text{H}_2\text{D}^+$, OPR
- $\text{H}_2\text{D}^+ + \text{H}_2$ THz Spectroscopy in Lab and Space

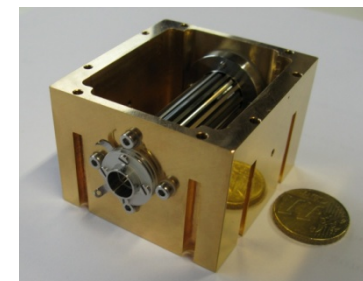
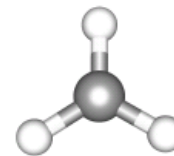
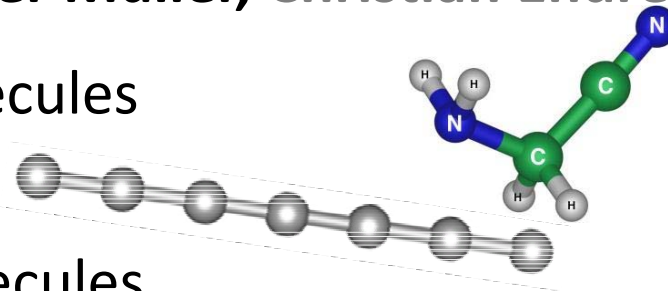


High-Resolution Spectroscopy of Interstellar Molecules

Cologne Astrophysics Group
Universität zu Köln

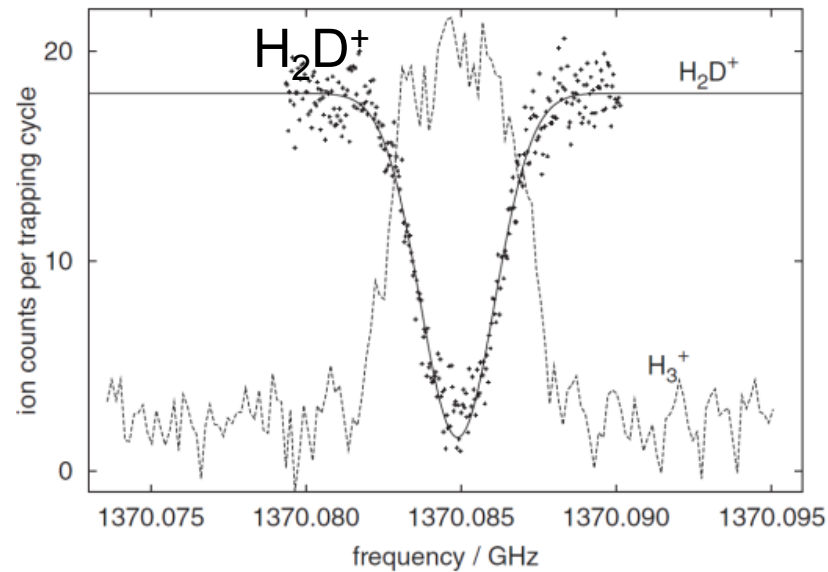


- Complex Molecules in Laboratory and Space
Frank Lewen, Holger Müller, Christian Endres
- Carbon Chain Molecules
Thomas Giesen
- Silicon Carbon Molecules
Sven Thorwirth
- He-Clusters **Leonid Surin**
- Trap Experiments
Sandra Brünken, Oskar Asvany, Pavol Jusko



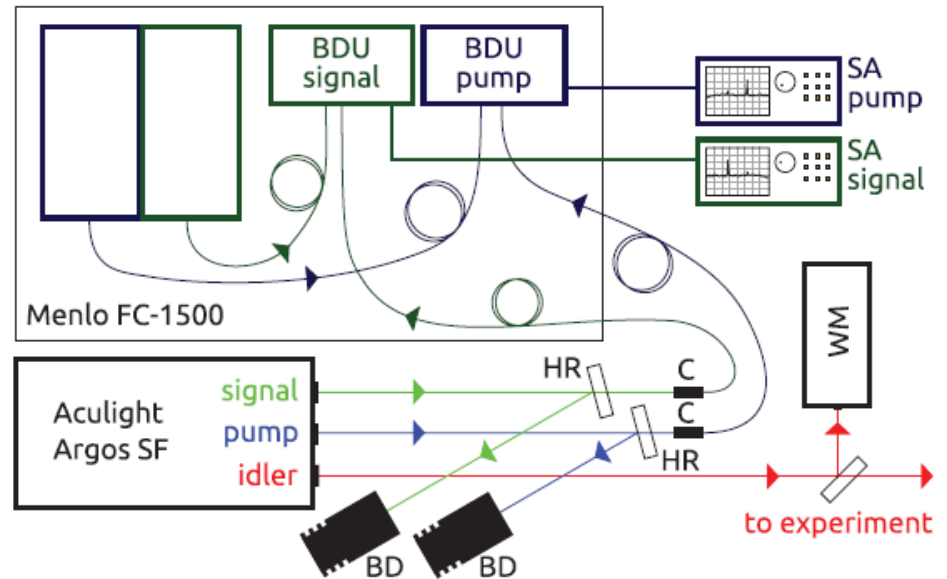
THz Action Spectroscopy

LIR



Asvany et al. 2008, Phys Rev Lett

Frequency Comb

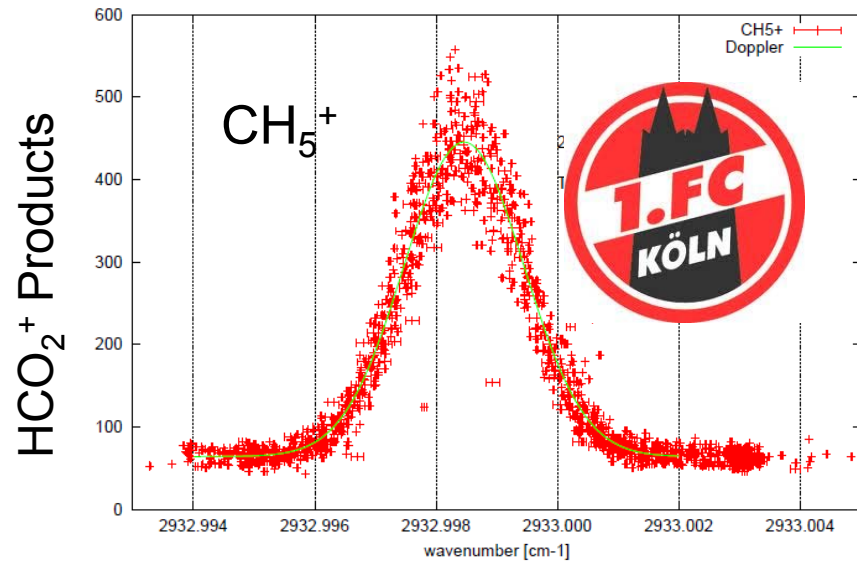


Gärtner et al. 2013, J. Phys. Chem. A

Asvany et al. 2012, Rev Sci Instr

Infrared Action Spectroscopy

LIR

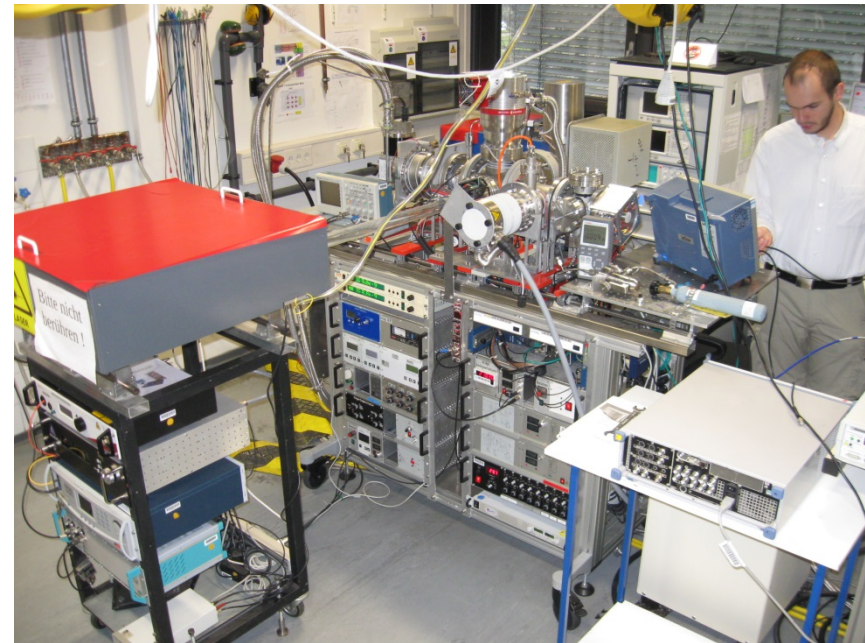


2932.998459(7) cm^{-1}

$T = 20.9 \pm 0.4 \text{ K}$

Asvany et al. 2012, Rev Sci Instr

COLogne TRAP

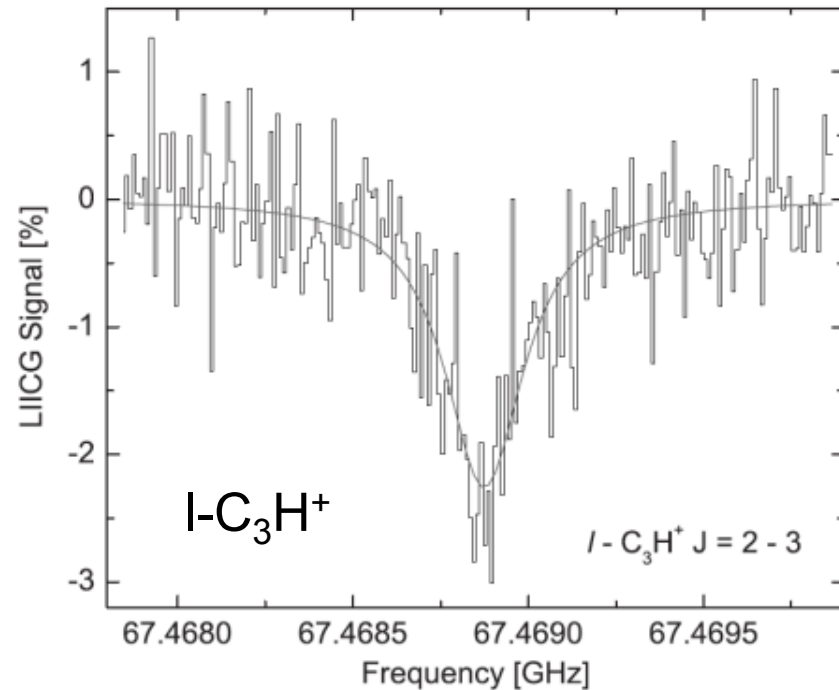


Accuracy: 0.2 MHz

Asvany et al. 2013, Appl Phys B

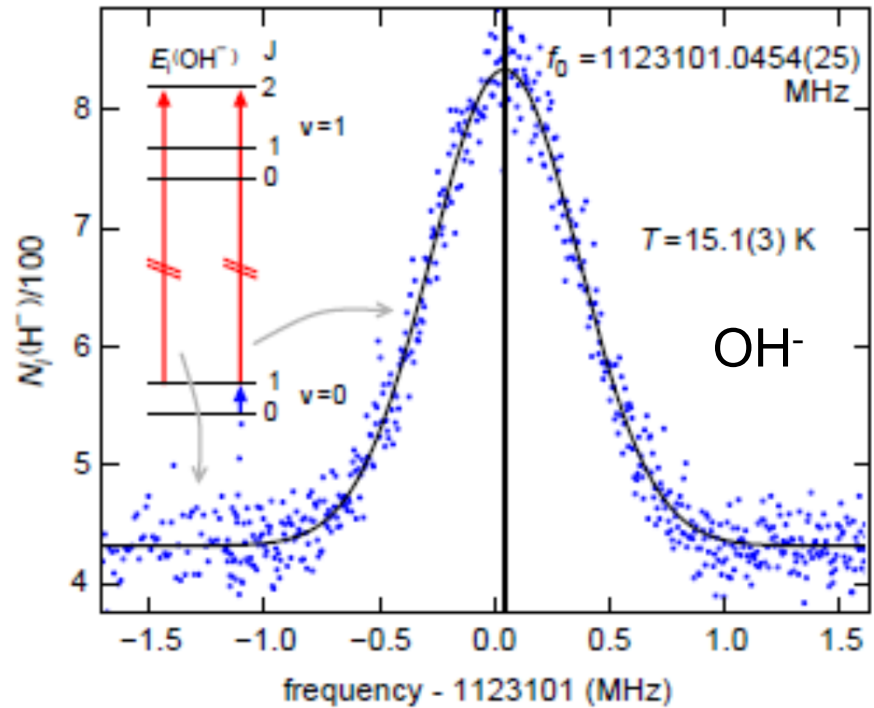
THz Action Spectroscopy

LIICG



Asvany et al. 2013, Appl Phys B
Brünken et al. 2014, ApJL

2-photon LIR



Jusko et al. 2014, Phys Rev Lett