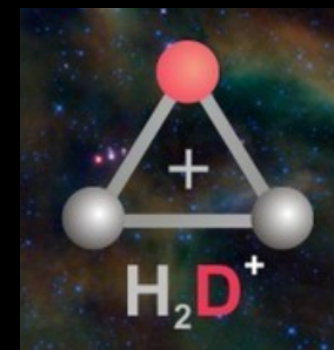
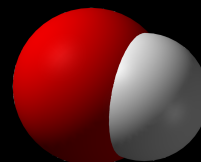
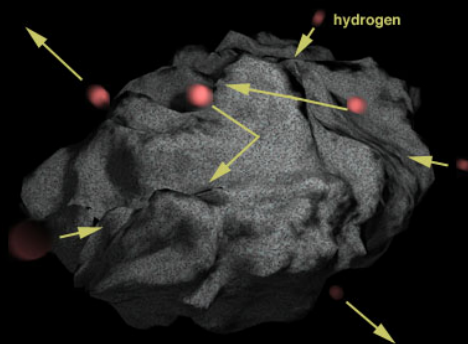
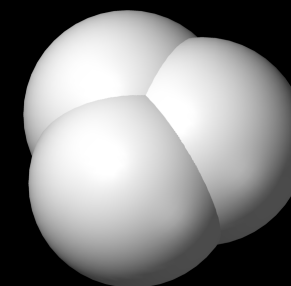
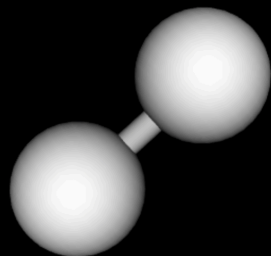


Astrochemistry with SOFIA

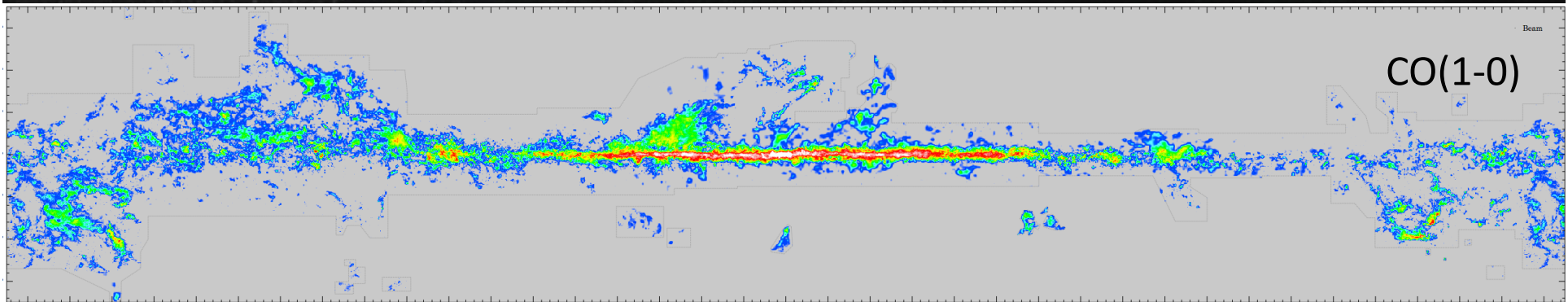


Paola Caselli
Center for Astrochemical Studies
Max-Planck-Institute for Extraterrestrial Physics



Molecular clouds in the Milky Way

~ 100,000 light years



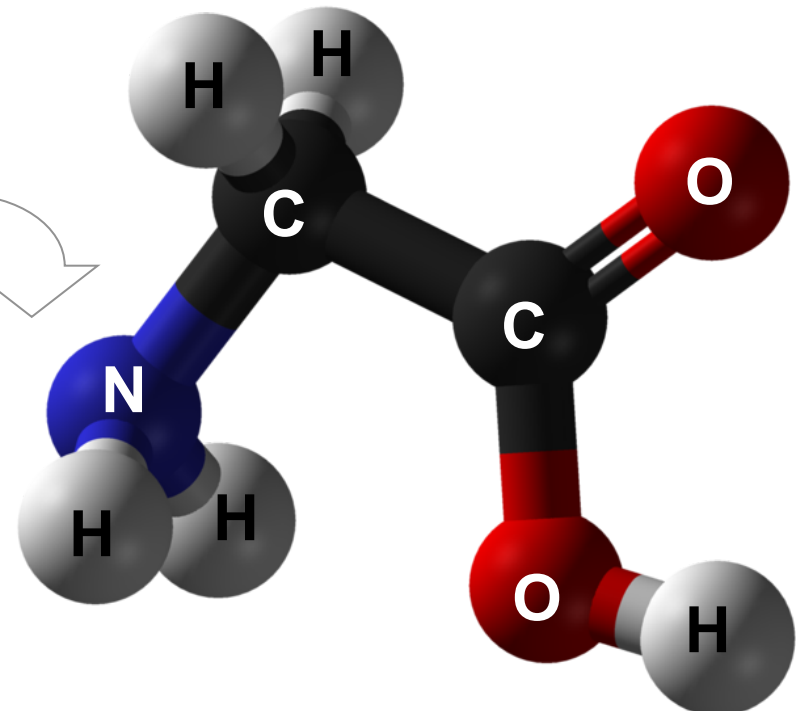
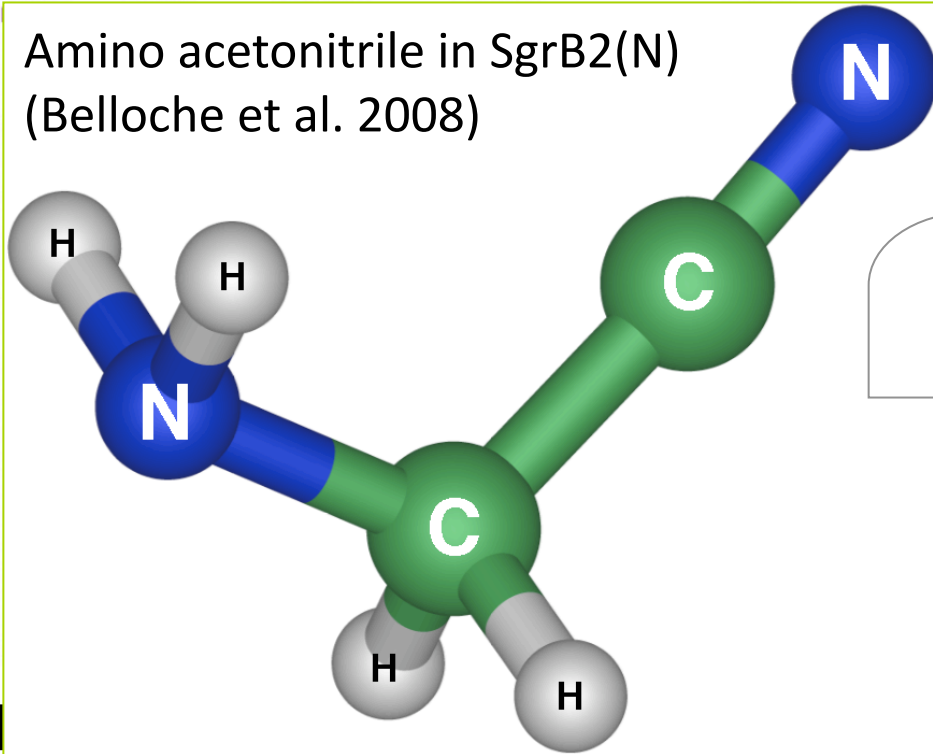
Dame, Hartmann & Thaddeus 2001

Interstellar molecules and the seeds of life

Known Interstellar Molecules (Total ~200)

Number of Atoms								
2	3	4	5	6	7	8	9	
H ₂	H ₂ O	NH ₃	SiH ₄	CH ₃ OH	CH ₃ COH	CH ₃ CO ₂ H	CH ₃ CH ₂ OH	
OH	H ₂ S	H ₃ O ⁺	CH ₄	NH ₂ CHO	CH ₃ NH ₂	HCO ₂ CH ₃	(CH ₃) ₂ O	
SO	SO ₂	H ₂ CO	CHOOH	CH ₃ CN	CH ₃ CCH	CH ₃ C ₂ CN	CH ₃ CH ₂ CN	
SO ⁺	HN ₂ ⁺	H ₂ CS	HCCCN	CH ₃ NC	CH ₂ CHCN	C ₇ H	H(CC) ₃ CN	
SiO	HNO	HNCO	CH ₂ NH	CH ₃ SH	HC ₄ CN	H ₂ C ₆	H(CC) ₂ CH ₃	
SiS	SiH ₂ ?	HNCS	NH ₂ CN	C ₅ H	C ₆ H	C ₇	C ₈ H	

Amino acetonitrile in SgrB2(N)
(Belloche et al. 2008)



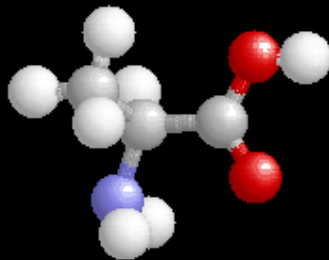
Glycine - the simplest amino acid

Pre-biotic molecules in meteorites

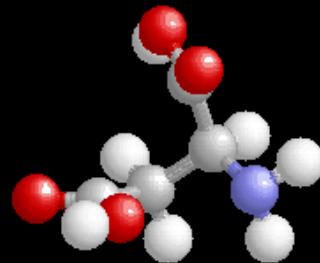


ALLENDE, CV3, MEXICO
Photo & Collection
Harald Seehlik

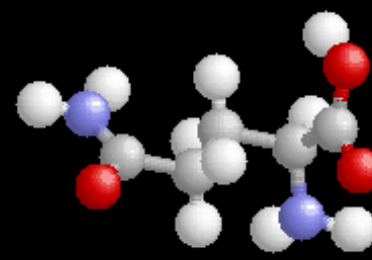
**~70 amino acids have been identified in
meteorites;
21 of these are used in life.**



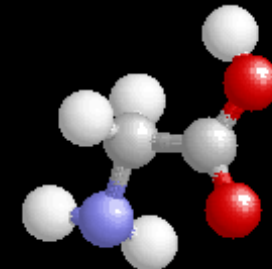
L-Alanine



L-Aspartic Acid

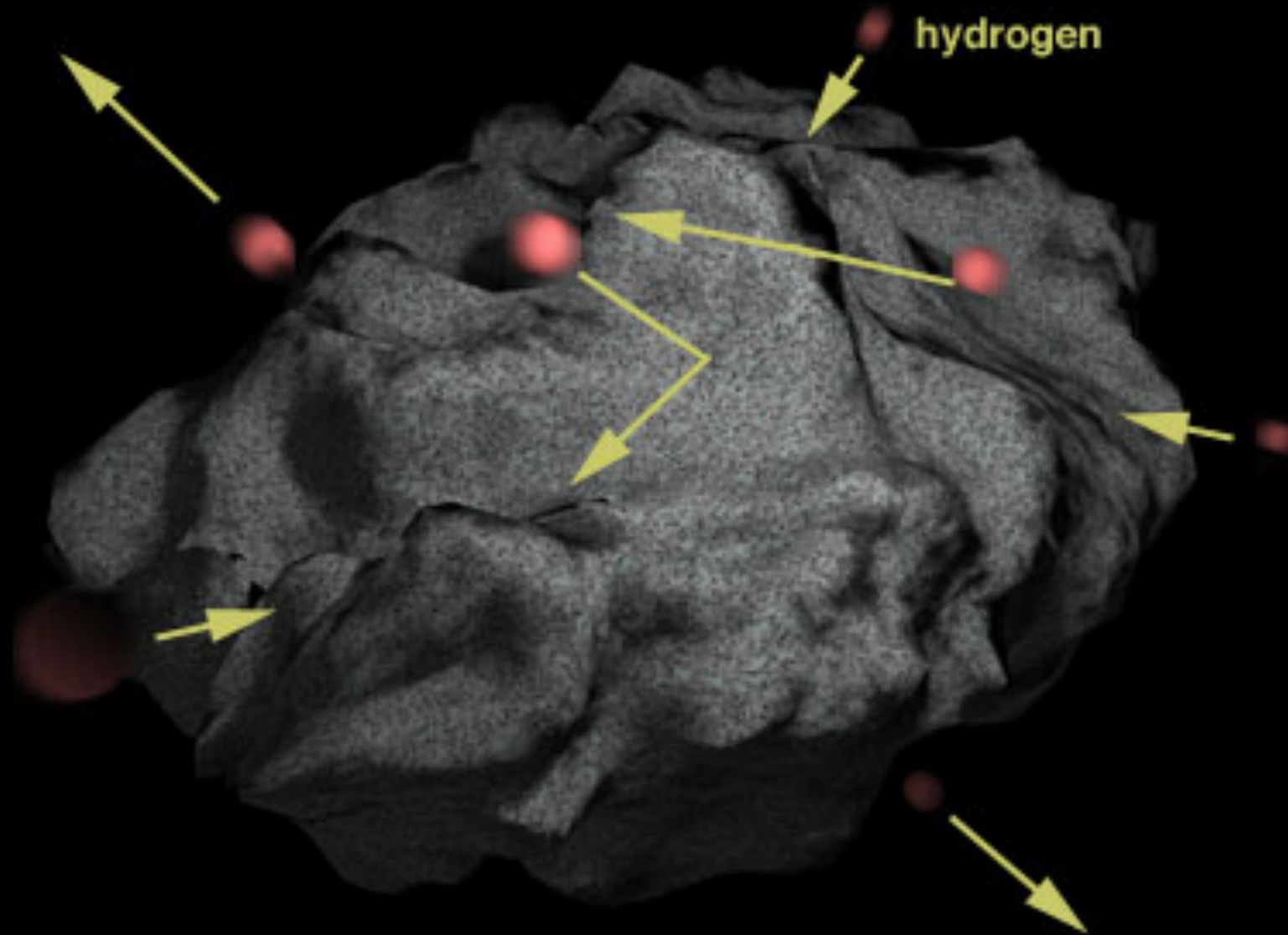


L-Glutamine

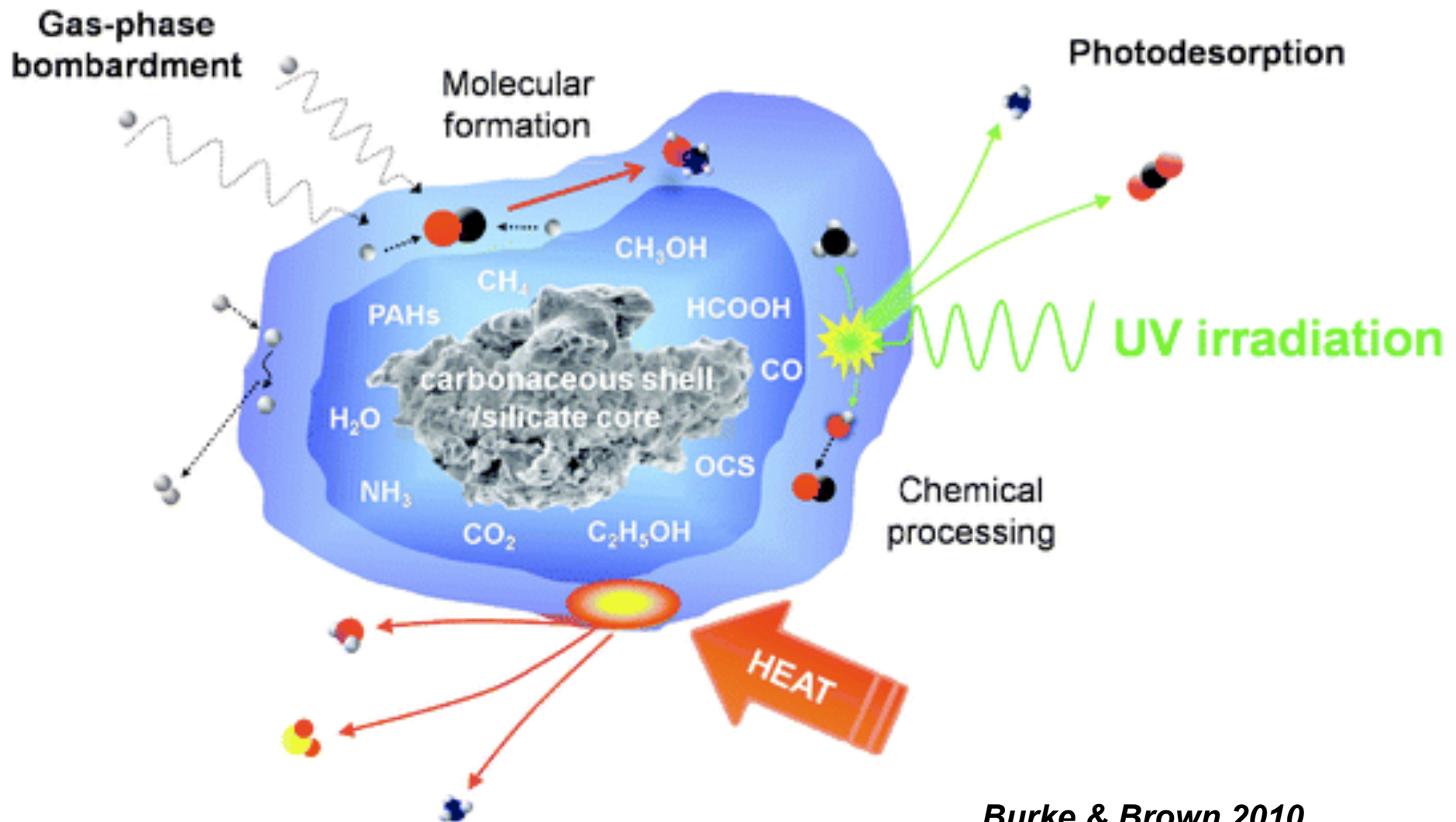


Glycine

Interstellar molecules and the seeds of life



Interstellar molecules and the seeds of life

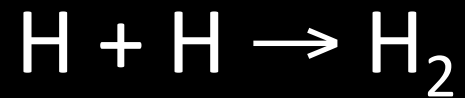


Outline

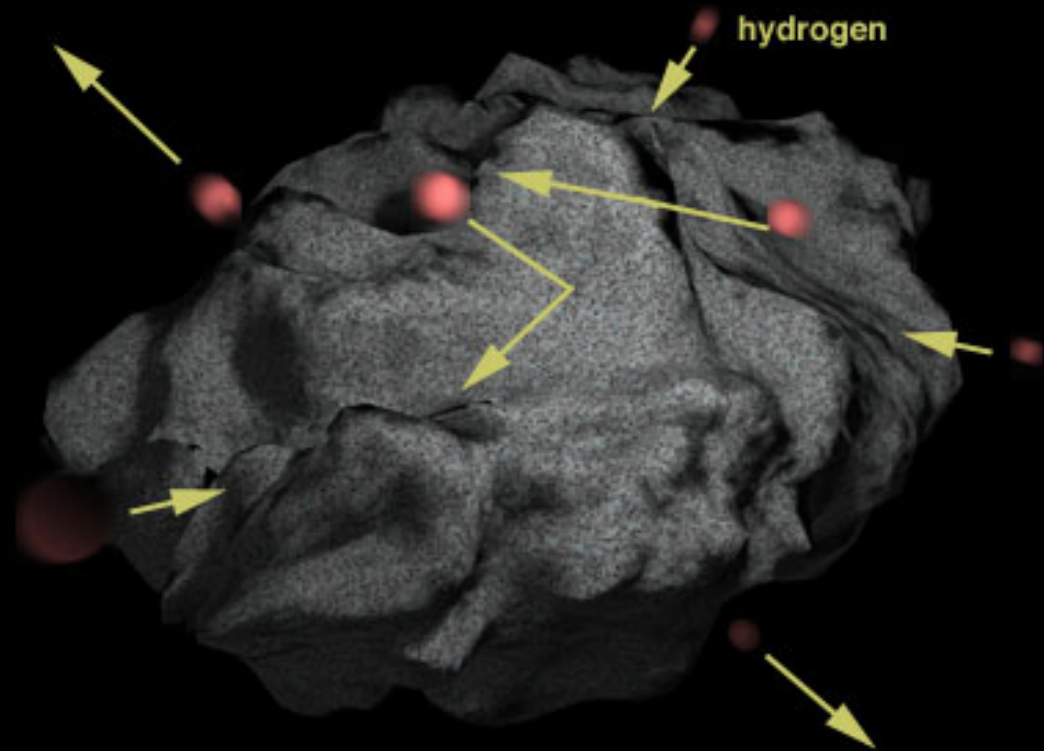
- Formation of H_2
- Formation of H_3^+
- Deuterium fractionation
- The ortho-to-para H_2 ratio
- SOFIA discovery of para- H_2D^+
- SOFIA and oxygen chemistry

The formation of H₂

The reaction that starts the chemistry in the interstellar medium is the one between two hydrogen atoms to form molecular hydrogen:



This reaction happens on the surface of dust grains.



The H₂ formation rate (cm⁻³ s⁻¹) is given by (e.g. Gould & Salpeter 1963; Hollenbach & Salpeter 1970; Jura 1974; Pirronello et al. 1999; Cazaux & Tielens 2002; Bergin et al. 2004; Cuppen & Herbst 2005; Cazaux et al. 2008):

$$R_{H_2} = \frac{1}{2} n_H v_H A n_g S_H \gamma$$
$$\cong 10^{-17} - 10^{-16} \text{ cm}^{-3} \text{ s}^{-1}$$

n_H ≡ gas number density

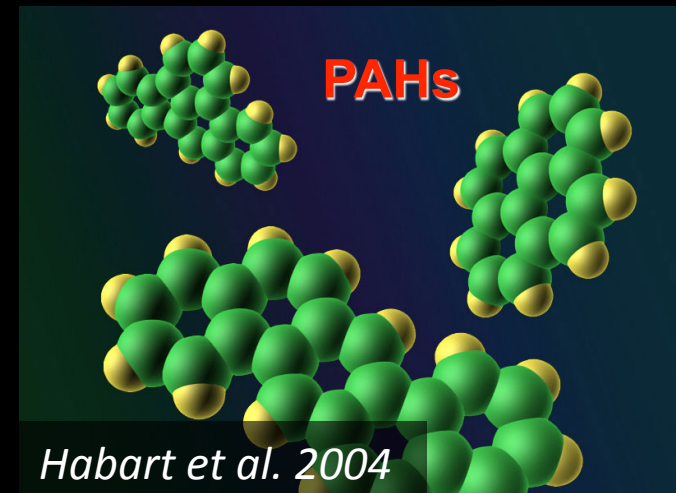
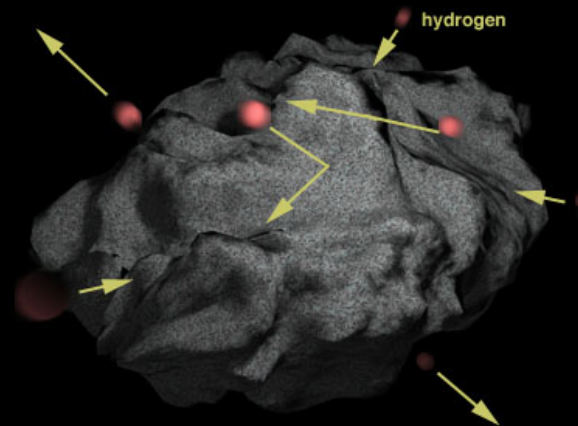
v_H ≡ H atoms speed in gas-phase

A ≡ grain cross sectional area

n_g ≡ dust grain number density

S_H ≡ sticking probability

γ ≡ surface reaction probability

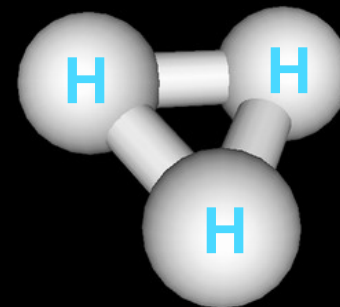
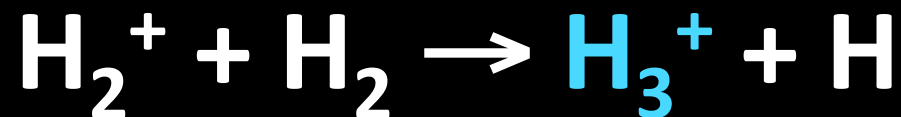


The formation of H_3^+

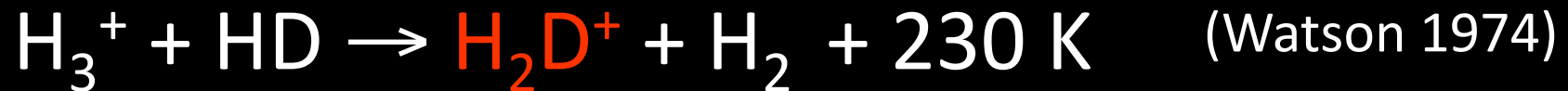
After the formation of molecular hydrogen, **cosmic rays** ionize H_2 initiating fast routes towards the formation of complex molecules in dark clouds:



Once H_2^+ is formed (in small percentages), it very quickly reacts with the abundant H_2 molecules to form H_3^+ , the most important molecular ion in interstellar chemistry:



Deuterium Fractionation at $T < 20$ K



$\text{H}_2\text{D}^+ / \text{H}_3^+$ increases if the abundance of gas phase neutral species decreases (Dalgarno & Lepp 1984):

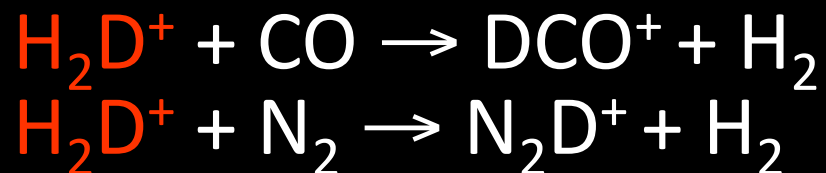
LOWER
 H_2D^+ and H_3^+
destruction
rates

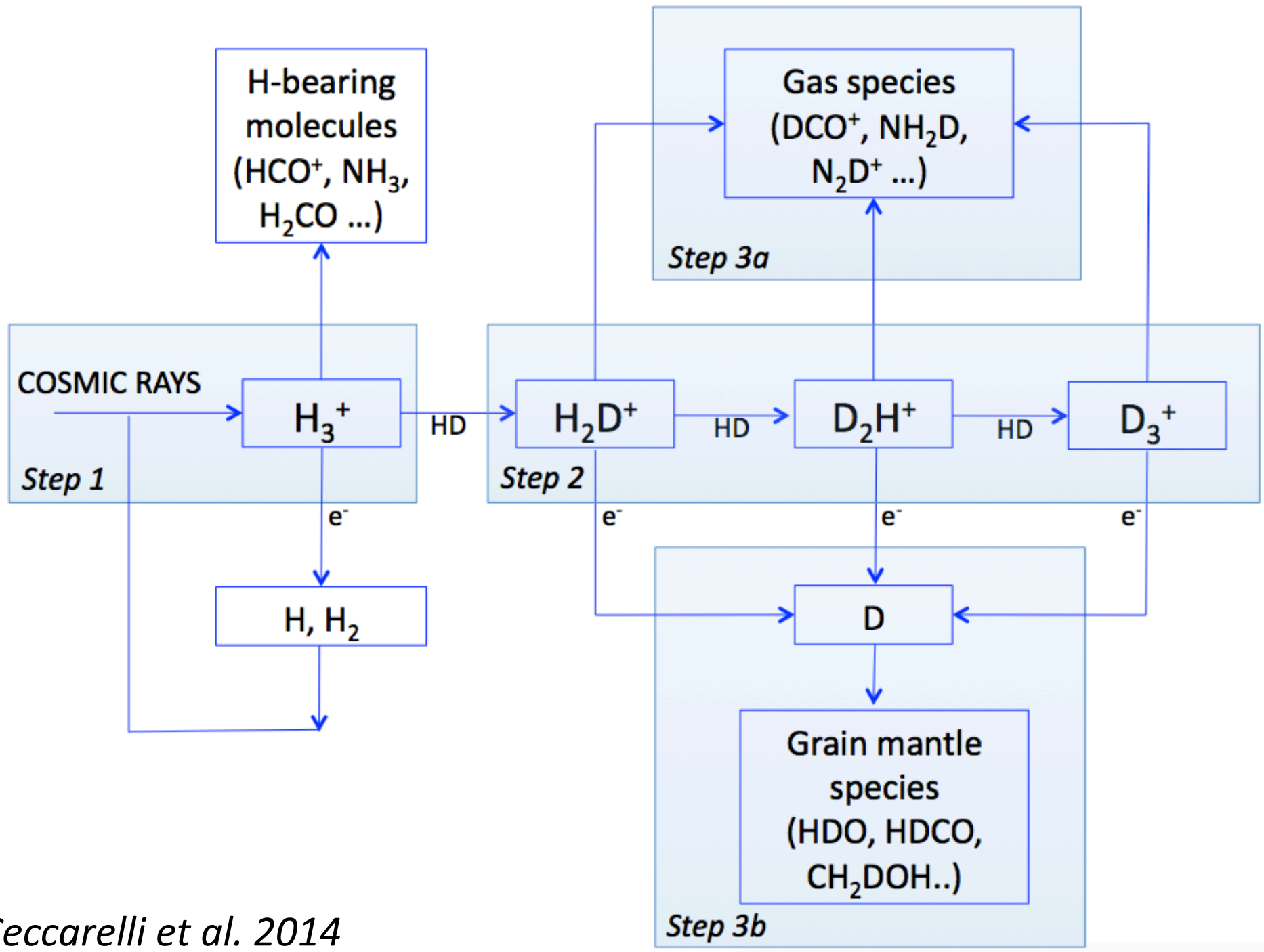
+

HIGHER
 H_2D^+
formation
rate

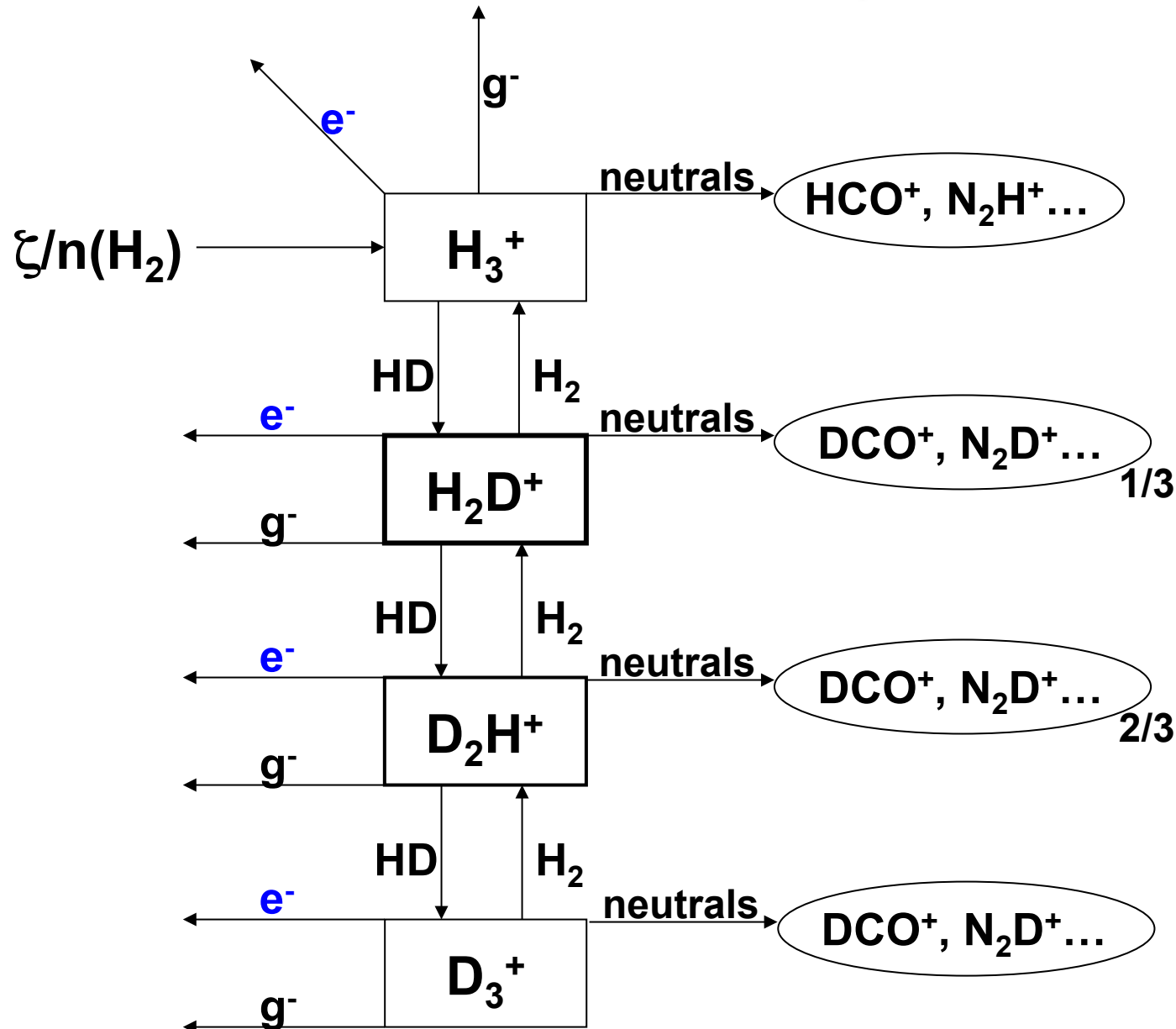
=

LARGER
 $\text{H}_2\text{D}^+ / \text{H}_3^+$ abundance ratio
and deuterium fractionation



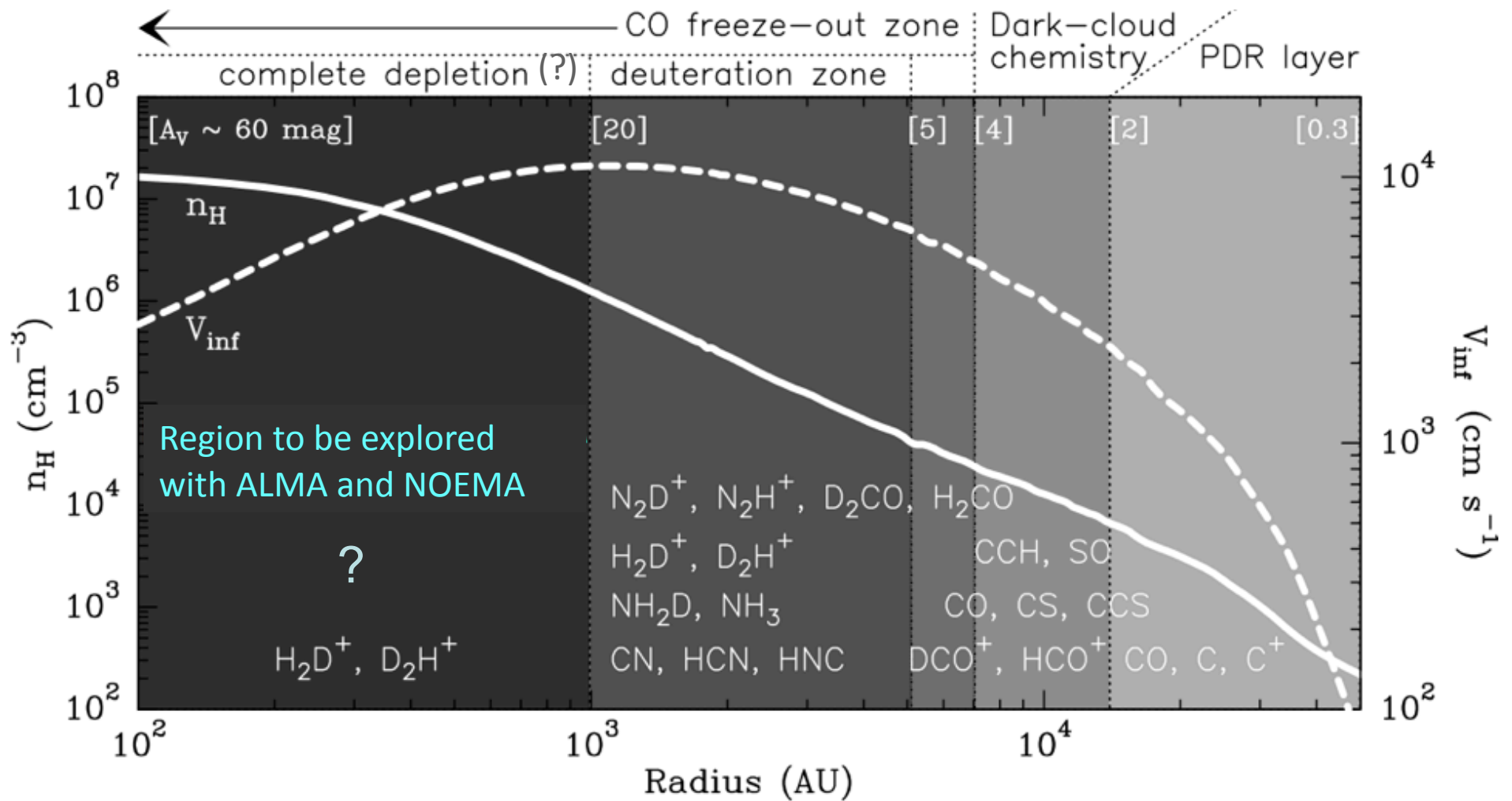


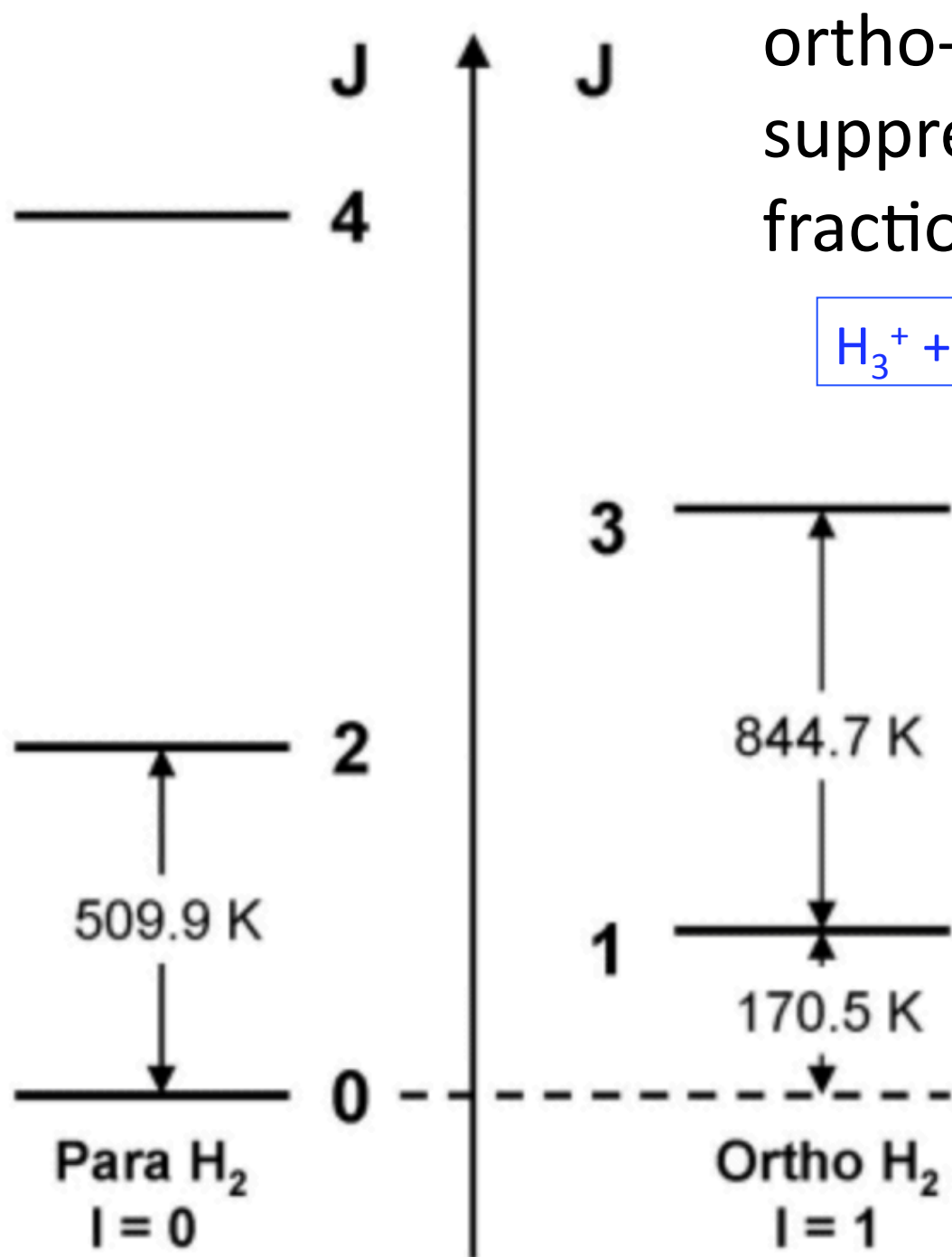
Deuterated molecular ions put limits on $x(e)$ and ζ



Guelin et al. 1977
Wootten et al. 1979
Guelin et al. 1982
Bergin et al. 1998
Caselli et al. 1998
Dalgarno 2006

Deuterated molecules are the best tracers of pre-stellar cores

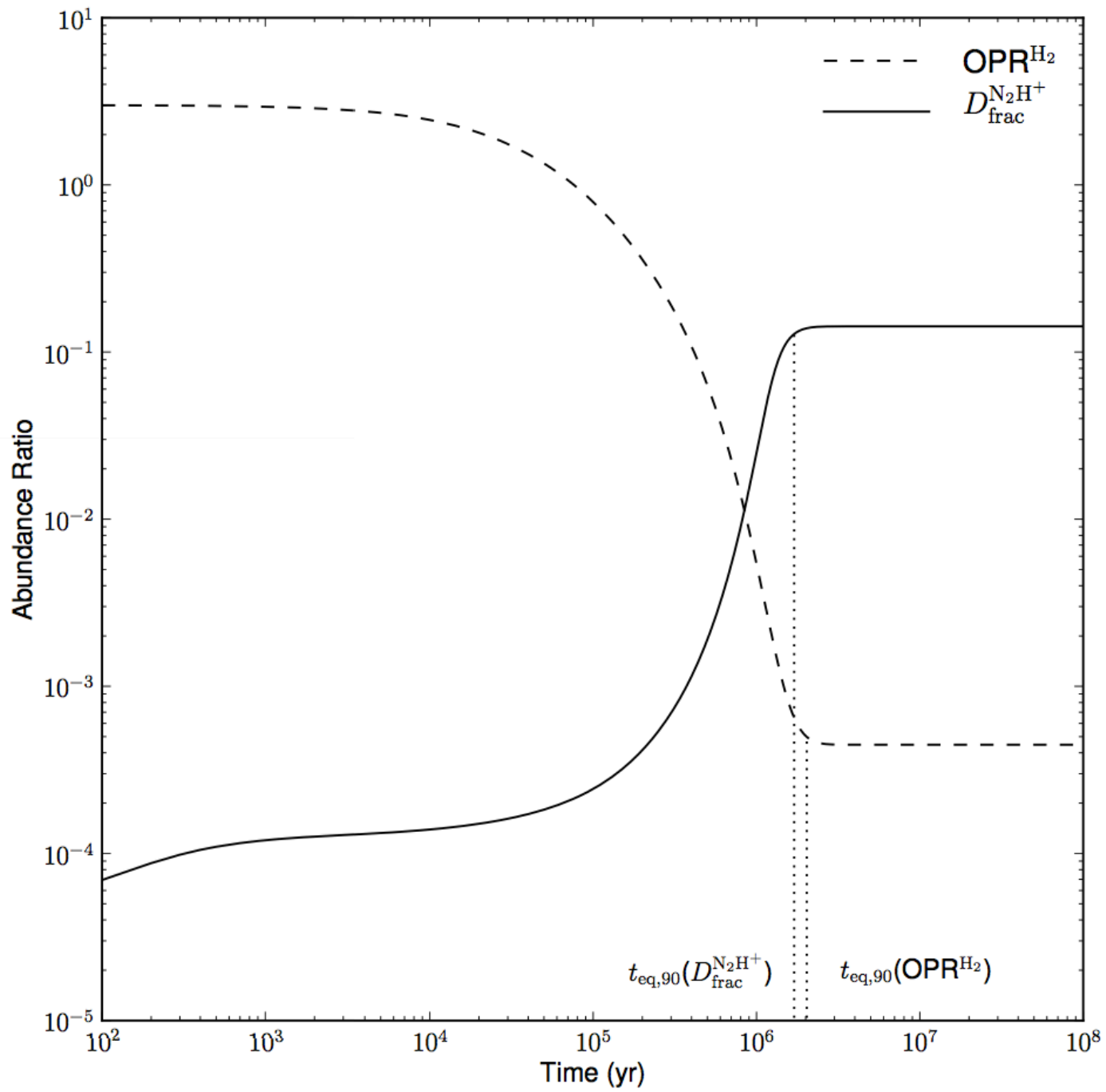




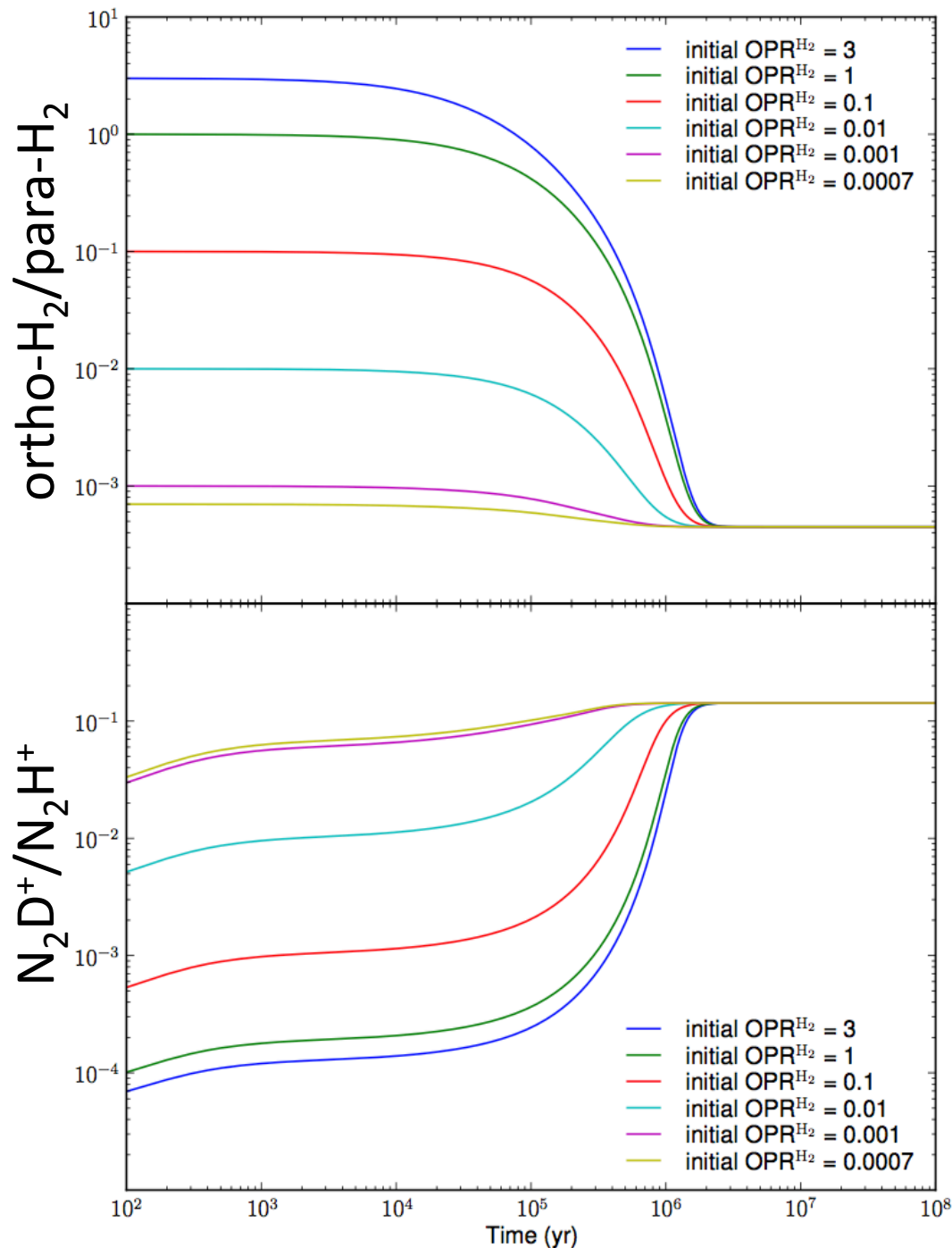
ortho-H₂ can slow down /
suppress the deuterium
fractionation



Pagani et al. 1992
Gerlich et al. 2002
Hugo et al. 2009
Kong et al. 2015



Kong et al. 2015

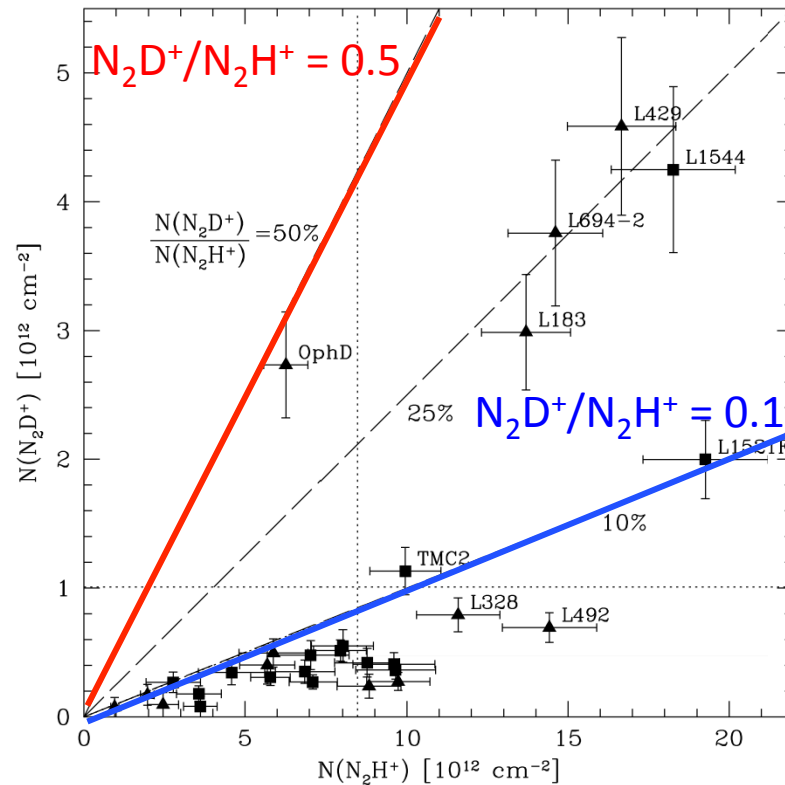


The conversion from ortho to para H_2 is a slow process and it is required to explain the observed large deuterium fractions.

$D_{\text{frac}} > 0.1$, requires collapse to be proceeding at rates about 10 times slower than that of free-fall collapse.

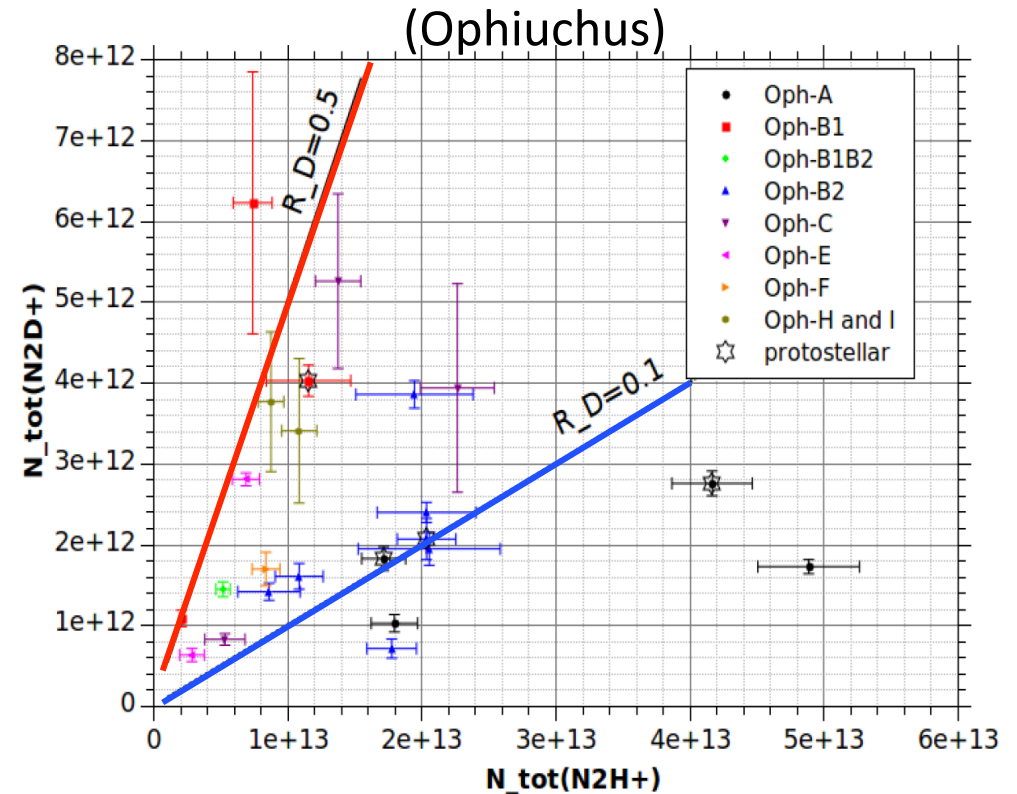
Large (variations of) deuterium fractions

Cores in different molecular clouds



Crapsi et al. 2005

Cores in the same molecular cloud



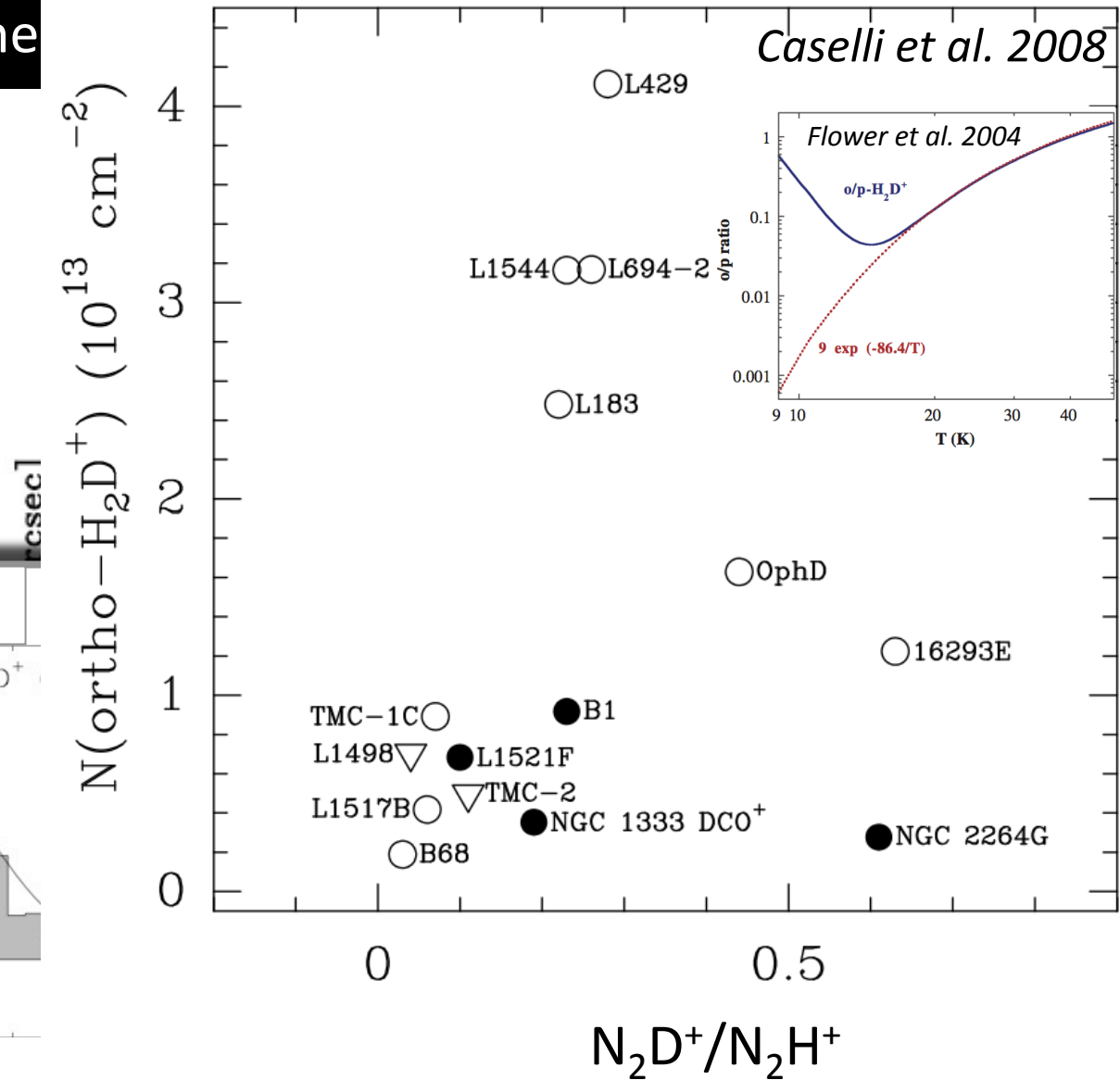
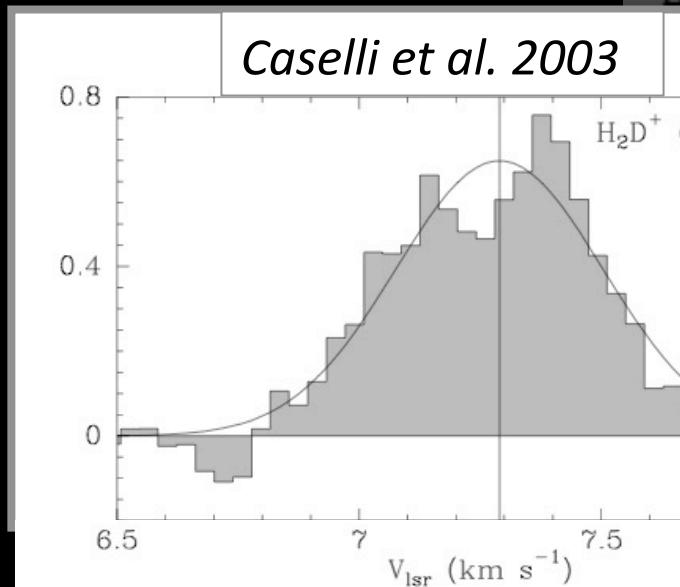
Punanova et al., in prep.

See also *Friesen et al. 2010; Friesen, Kirk & Shirley 2013; Schnee et al. 2013*

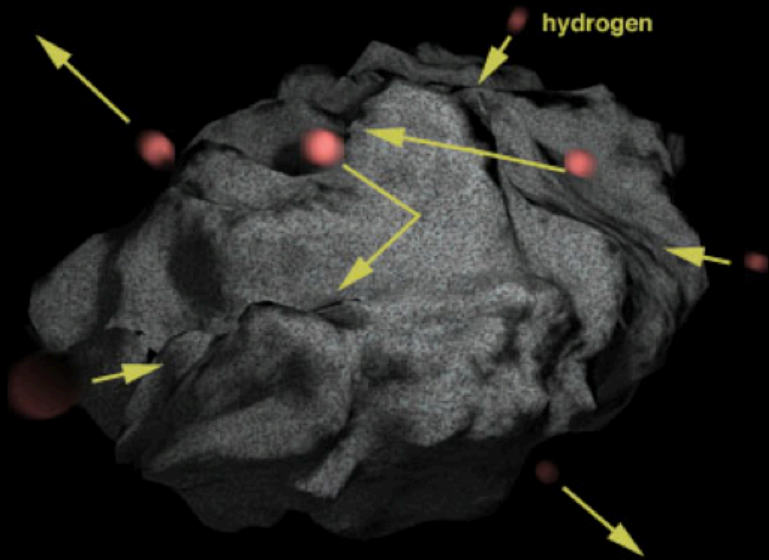
ortho-H₂D⁺

The 372 GHz o-H₂D⁺ line

Only models including all multiply deuterated forms of H₃⁺ can reproduce these data (Roberts et al. 2003; Walmsley et al. 2004; Aikawa et al. 2005)



How much ortho-H₂ is in molecular clouds?



Upon formation: $oH_2/pH_2 = 3$

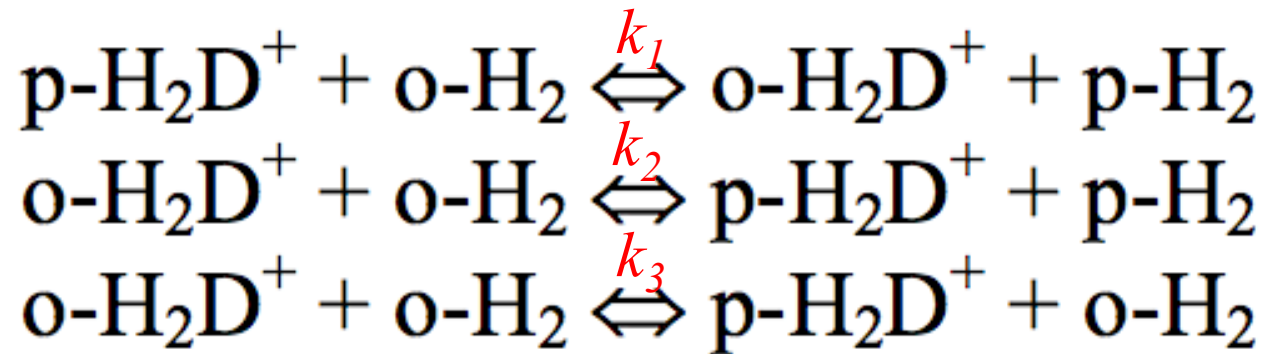
In diffuse clouds: $oH_2/pH_2 \sim 0.3-0.8$
(Crabtree et al. 2011)

In the pre-stellar core L183:
 $oH_2/pH_2 \sim 0.1$ (Pagani et al. 2009)

In the starless core B68: $oH_2/pH_2 \sim 0.015$
(Maret & Bergin 2007)

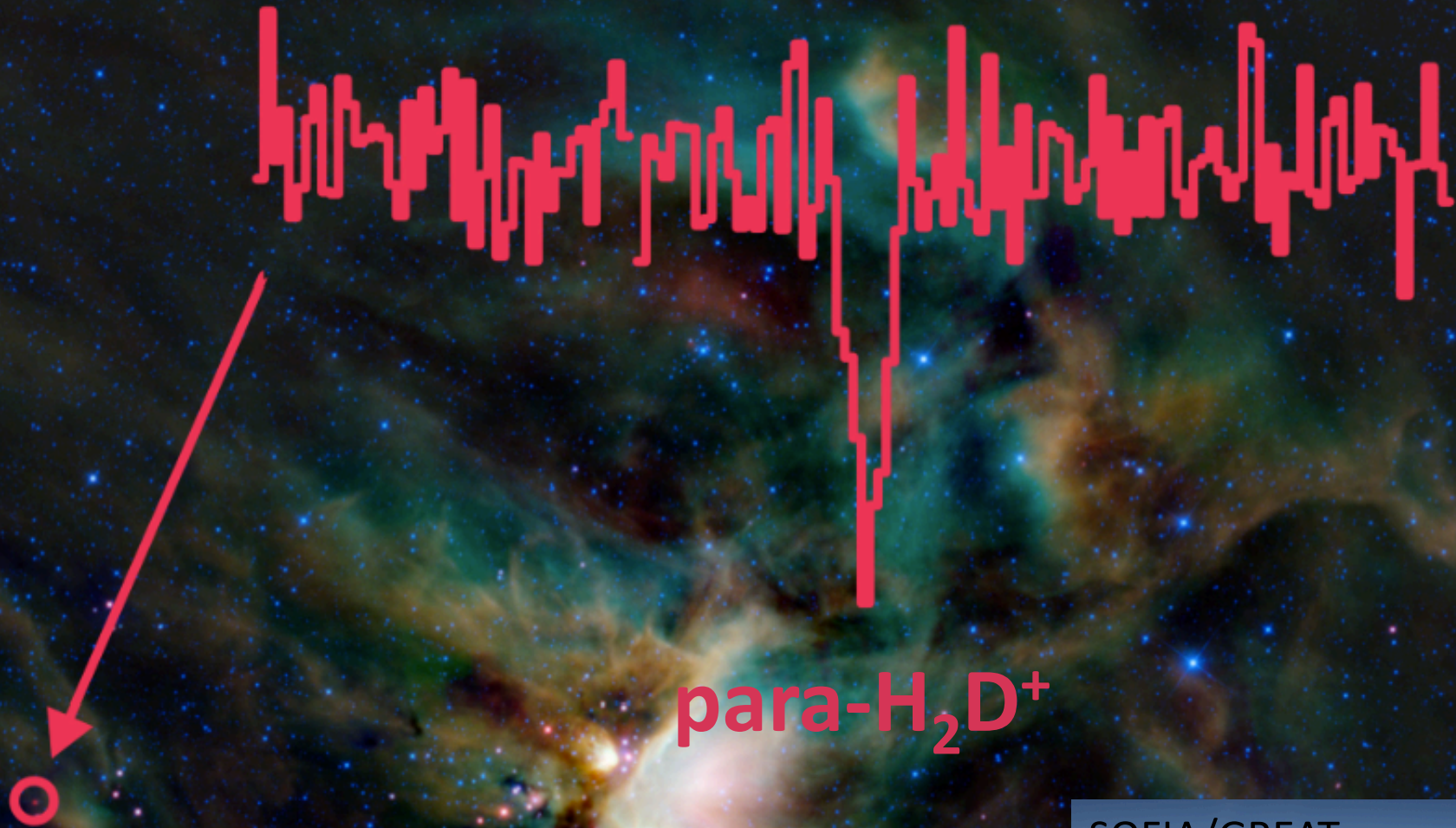
In the pre-stellar core L1544: $oH_2/pH_2 \sim 0.003$
(Kong et al. 2015)

Analytical relation between the H₂ and H₂D⁺ ortho-to-para ratios



$$\frac{[\text{o-H}_2\text{D}^+]}{[\text{p-H}_2\text{D}^+]} = \frac{(k_1^+ + k_3^-) \times [\text{o-H}_2] / [\text{p-H}_2] + k_2^-}{(k_2^+ + k_3^+) \times [\text{o-H}_2] / [\text{p-H}_2] + k_1^-}$$

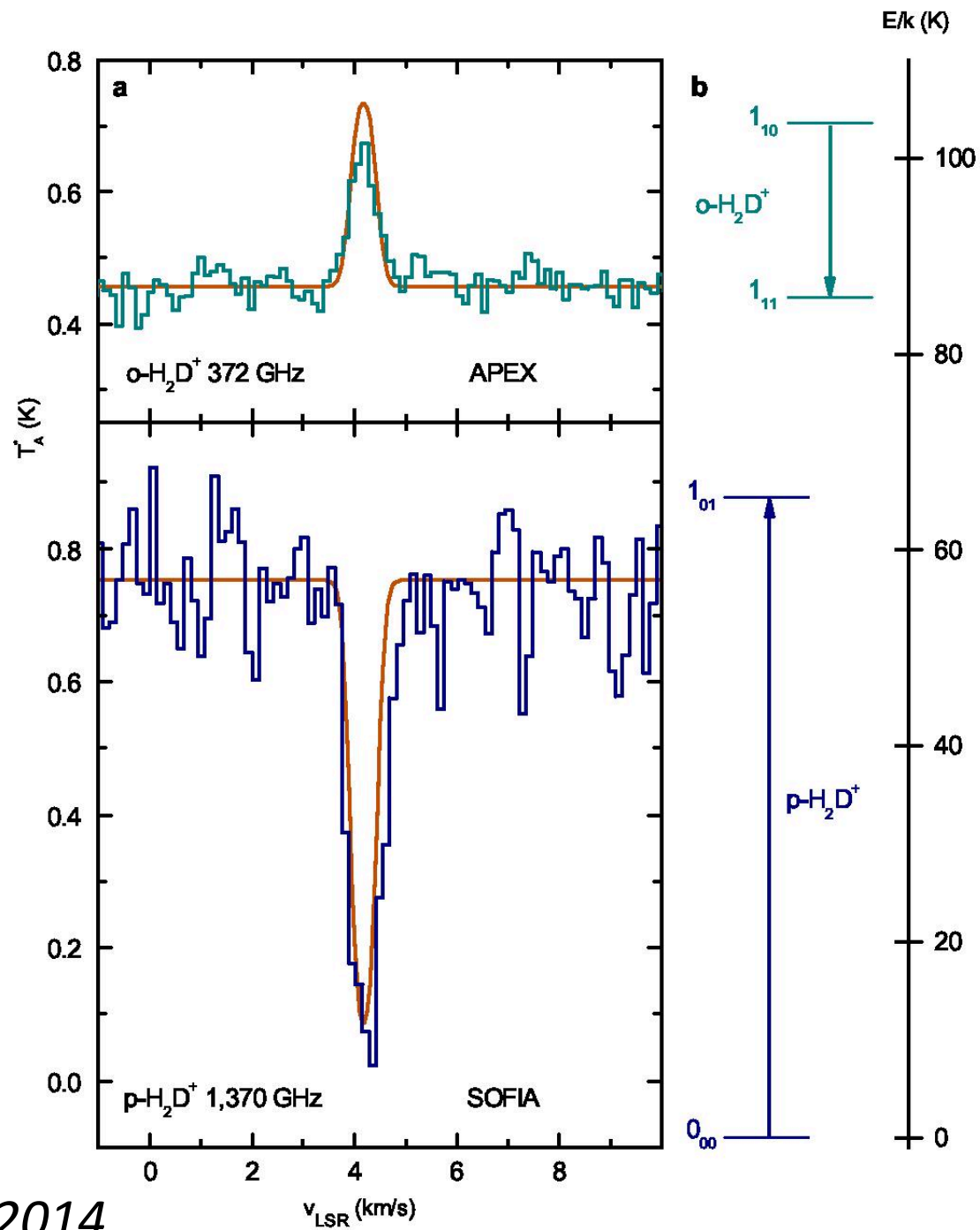
FIRST DETECTION OF para-H₂D⁺ TOWARD IRAS16293-2422



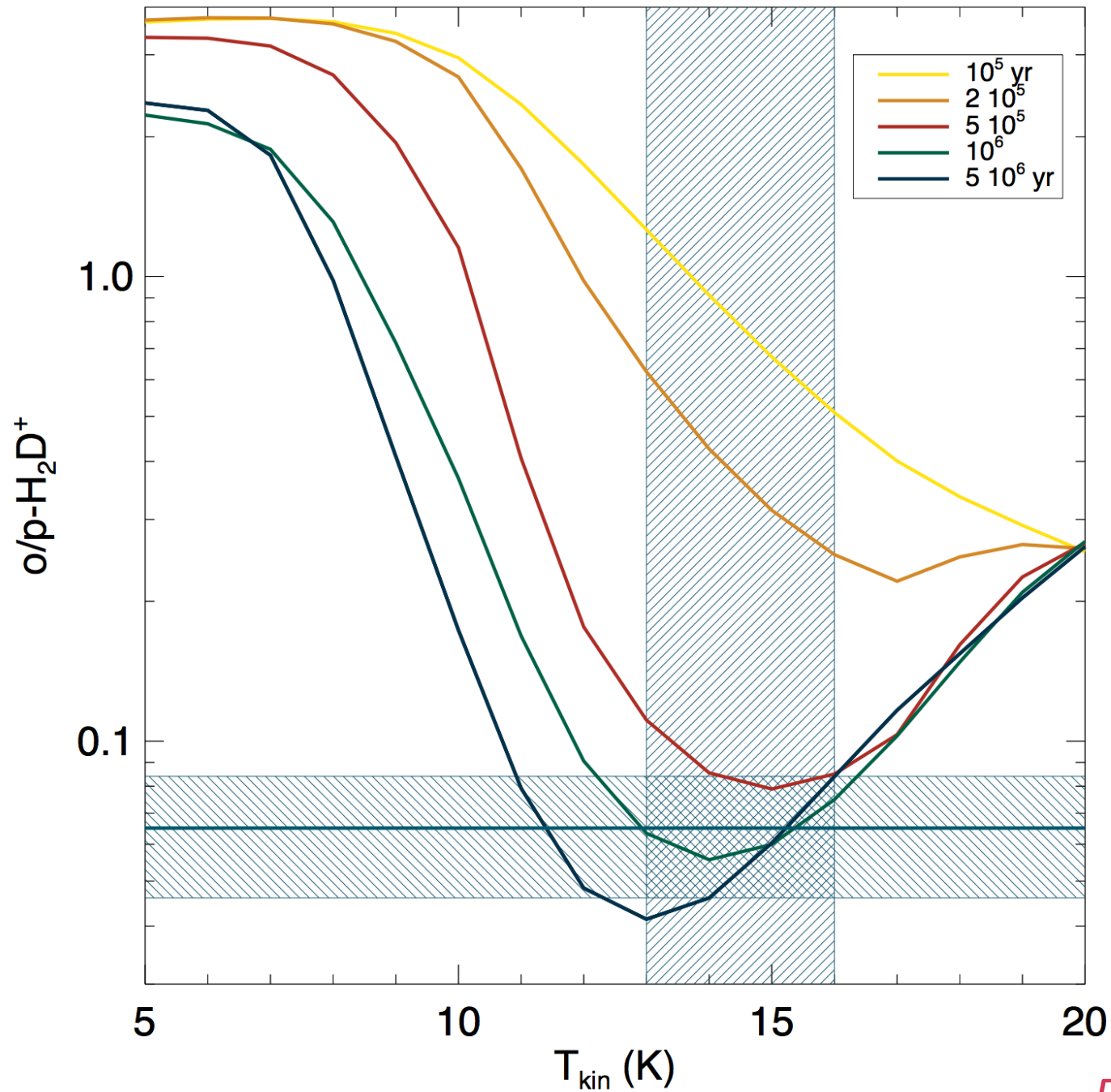
para-H₂D⁺



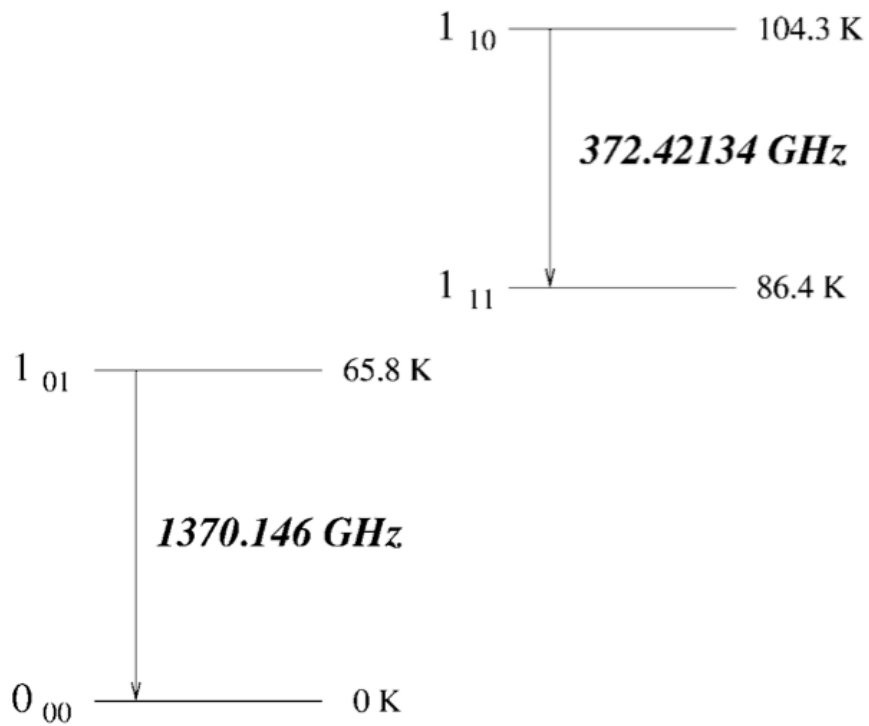
Brünken, Sipilä, Chambers, Harju, Caselli, Asvany, Honingh, Kamiński, Menten, Stutzki, Schlemmer 2014, Nature



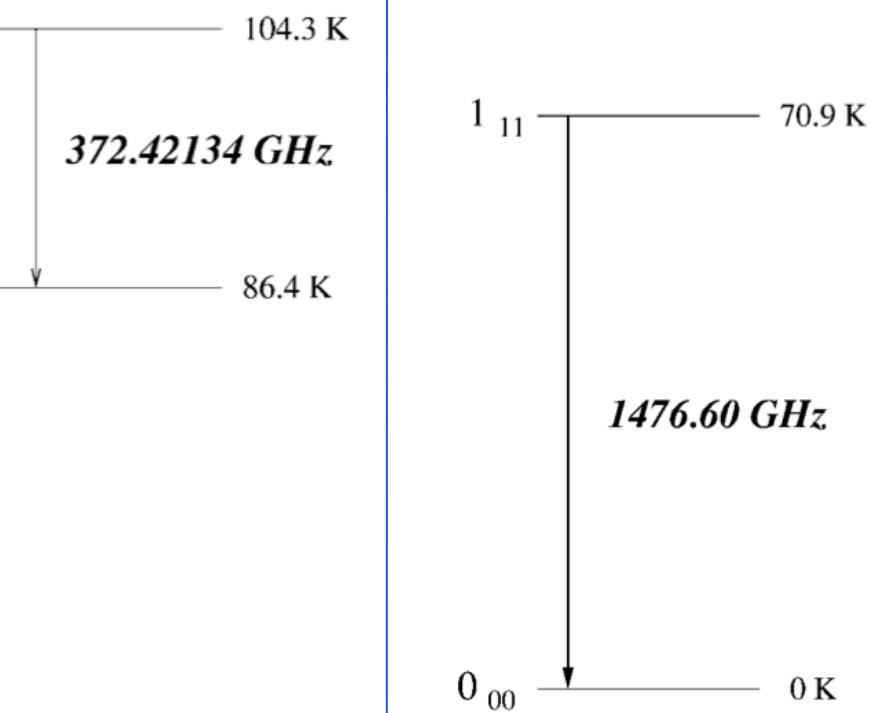
Brünken et al. 2014



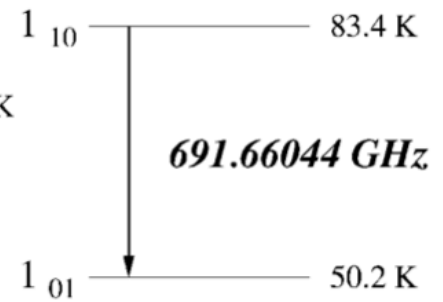
The molecular gas in the cool envelope has been subject to chemical processing for at least one million years.



PARA H_2D^+



ORTHO D_2H^+



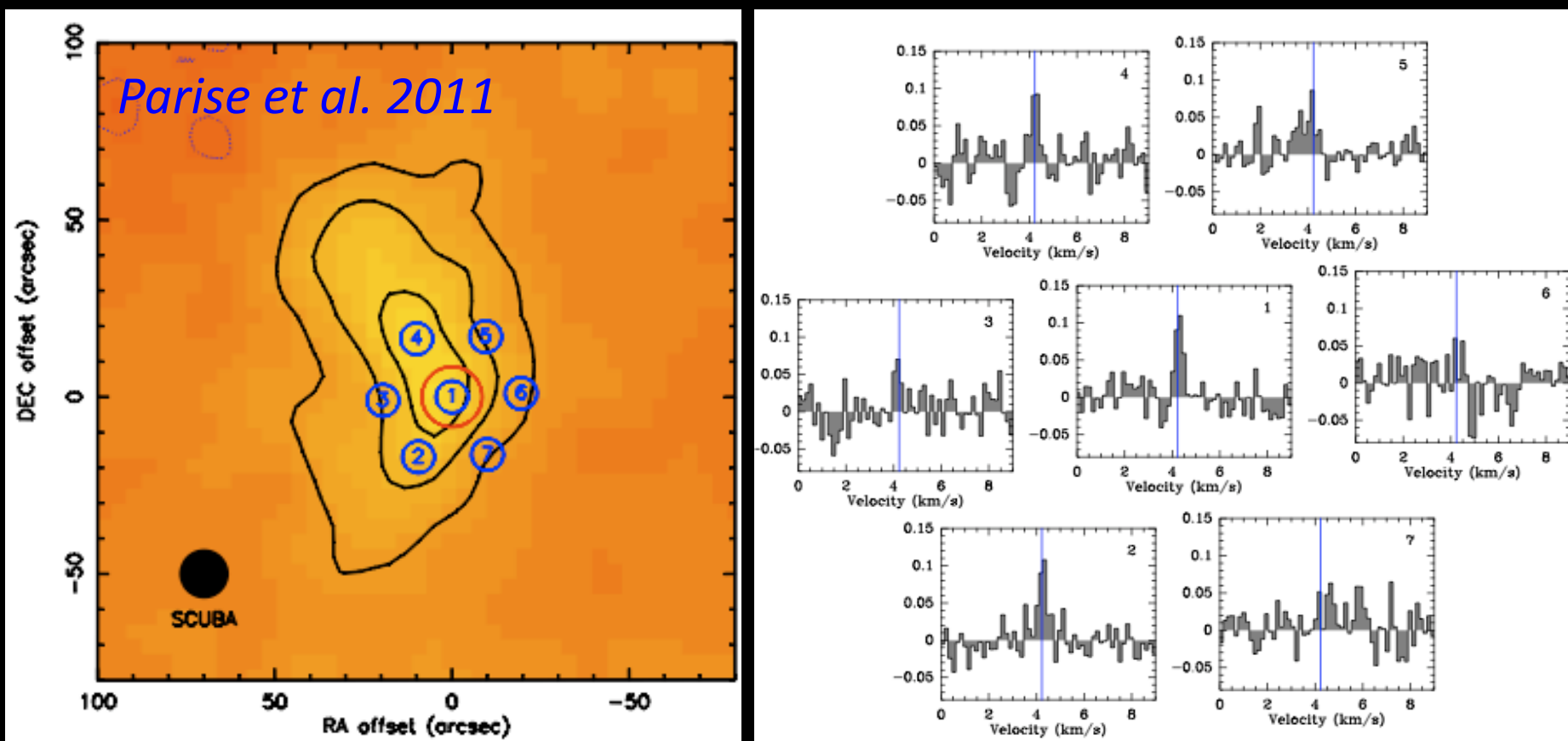
PARA D_2H^+



from Vastel et al. 2004

para-D₂H⁺

Extended para-D₂H⁺ emission (~40'' ~5000 AU) toward L1688/H-MM1 (692 GHz; APEX-CHAMP+)



See Vastel et al. (2004) for first detection of para-D₂H⁺

Ro-vibrational lines

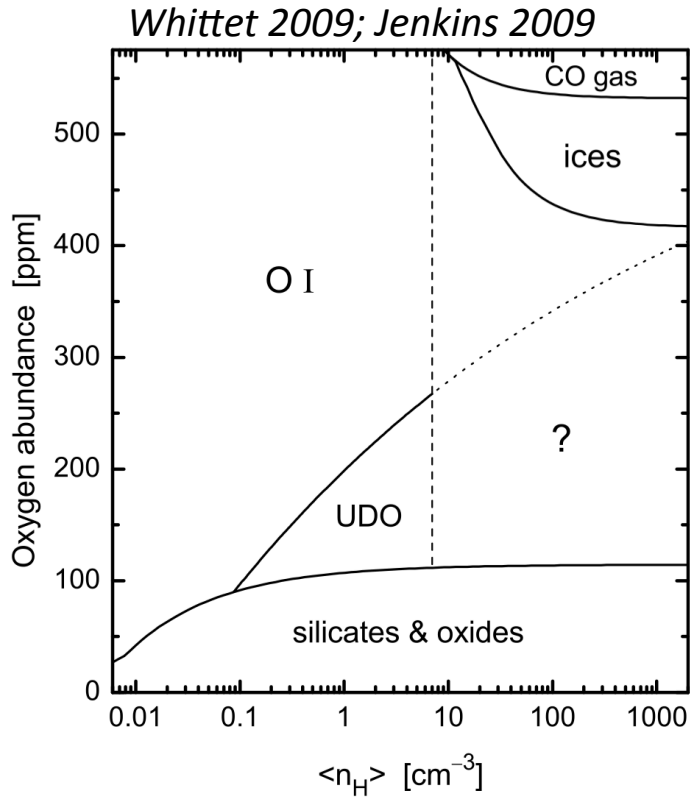
Table 3. Requisite column densities for unit optical depth in the absorption line centre for the deuterated forms of H_3^+ . Spectroscopic data from Ramanlal & Tennyson (2004).

Species	Upper level	Lower level	E_l (cm^{-1})	Wavelength (μm)	$N_{\tau=1}$ (cm^{-2})
H_2D^+ (para)	(1, 1, 1) ^a	(0, 0, 0)	0.0	4.162	1.9×10^{14}
H_2D^+ (ortho)	(0, 0, 0)	(1, 1, 1)	60.0	4.395	6.1×10^{14}
H_2D^+ (ortho)	(2, 0, 2)	(1, 1, 1)	60.0	4.136	7.1×10^{14}
H_2D^+ (ortho)	(2, 2, 0)	(1, 1, 1)	60.0	3.985	4.9×10^{14}
D_2H^+ (ortho)	(1, 1, 1)	(0, 0, 0)	0.0	4.965	4.7×10^{14}
D_2H^+ (ortho)	(1, 0, 1)	(0, 0, 0)	0.0	4.72	5.5×10^{14}
D_2H^+ (para)	(1, 1, 0)	(1, 0, 1)	34.9	5.02	6.9×10^{14}
D_2H^+ (para)	(2, 0, 2)	(1, 0, 1)	34.9	4.63	7.6×10^{14}
D_3^+ (ortho)	(0, 1, 1, 0, 1) ^b	(0, 0, 0, 0, 0)	0.0	5.296	2.2×10^{13}
D_3^+ (meta)	(0, 1, 0, 1, 1)	(0, 0, 1, 1, 0)	32.3	5.548	1.6×10^{14}
D_3^+ (meta)	(0, 1, 2, 1, -1)	(0, 0, 1, 1, 0)	32.3	5.198	9.9×10^{13}
D_3^+ (meta)	(0, 1, 2, 1, 1)	(0, 0, 1, 1, 0)	32.3	5.166	9.1×10^{13}
D_3^+ (para)	(0, 1, 1, 0, -1)	(0, 0, 1, 0, 0)	43.6	5.433	4.5×10^{14}

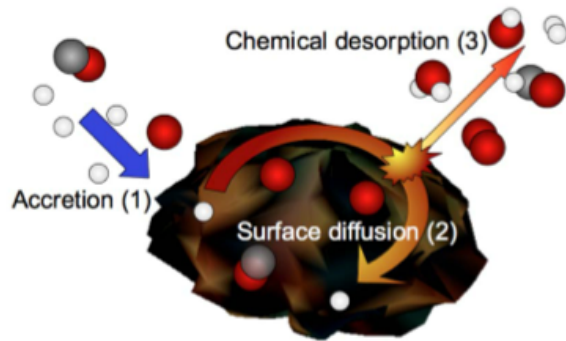
^a (J, K_a, K_c).

^b (ν_1, ν_2, J, G, U).

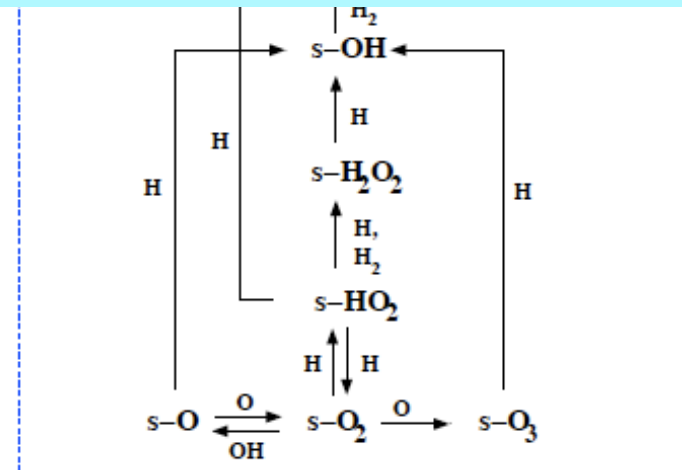
Oxygen Chemistry



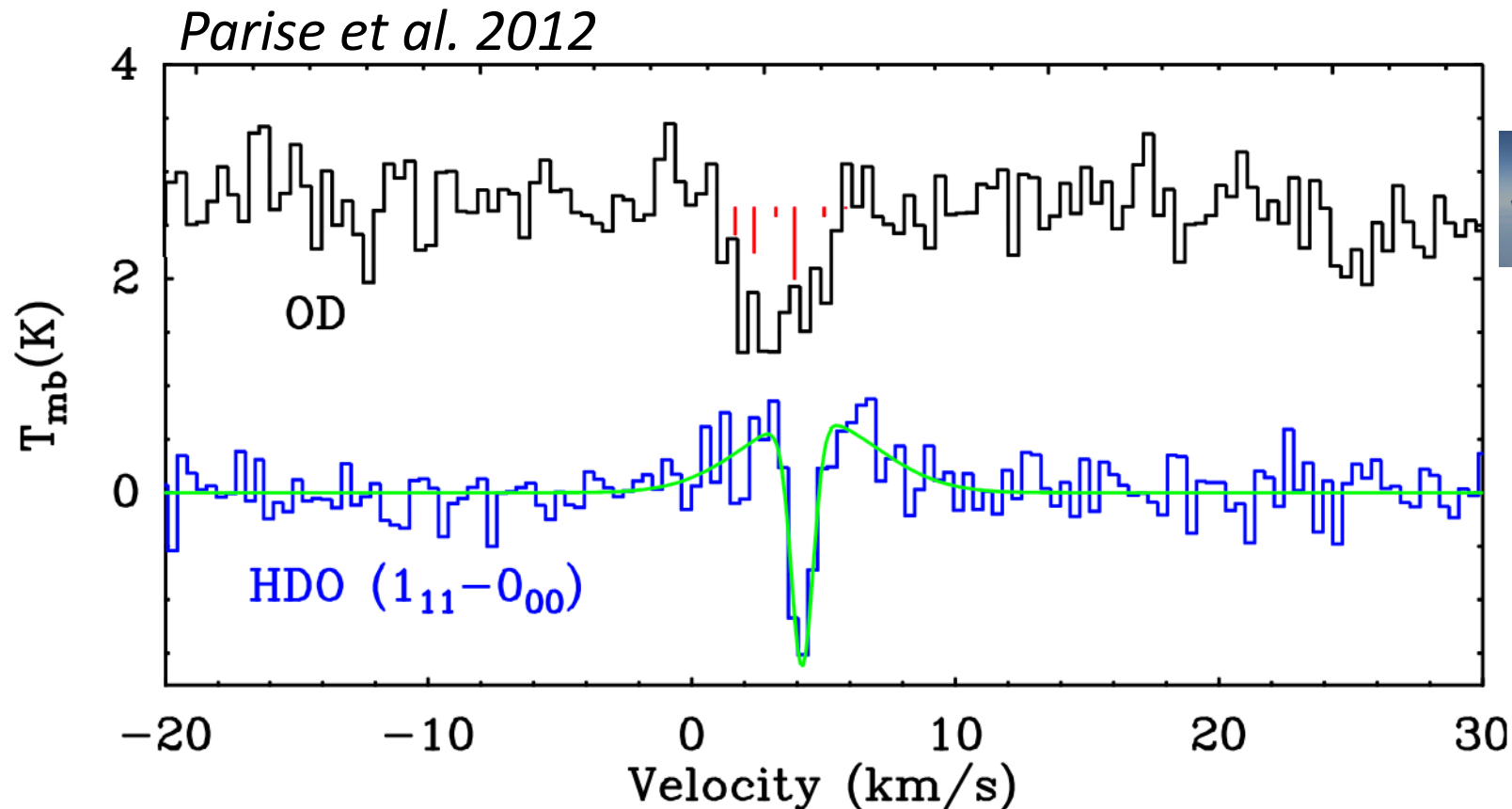
*It will be important in the future to obtain spectra of the highest quality for stars in lines of sight that sample diffuse-ISM dust over a range of visual extinctions, especially in the waveband containing the $5.85\mu\text{m}$ carbonyl feature as it should provide the most sensitive quantitative test for oxygenated organics. Both the **Stratospheric Observatory for Infrared Astronomy** and the **James Webb Space Telescope** will have instruments well matched to this task. – Whittet 2009*



Tielens & Hagen 1982;
Cuppen & Herbst 2007;
Miyauchi et al. 2008; Ioppolo et al. 2008, 2010; Cazaux et al. 2010, 2011; Taquet et al. 2013; Dulieu et al. 2013

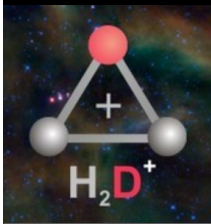


First detection of OD outside the Solar System



OD/HDO between 17 and 90 (high compared to model predictions → gas phase reprocessing through dissociative recombination of H_2DO^+ ?)

Parallel observations of OD and OH could provide valuable constraints on the formation and fractionation of water.



Summary



With **SOFIA** we can explore the foundations of astrochemistry:

- the deuteration of H_3^+ via HD – the starting point of D fractionation
- the ortho-to-para H_2 , via observations of para- H_2D^+ (and ortho- H_2D^+ with APEX) – important for D fractionation and cloud ages
- OH, OD (, carbonyl?) – oxygen chemistry and water fractionation – synergy with Herschel
- SH and sulfur chemistry (→ Neufeld)